

FYS2160 2023, Lecture 5

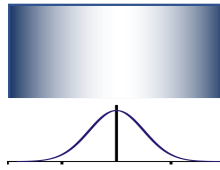
Dag Kristian Dysthe

Molecular dynamics

04.09.2023

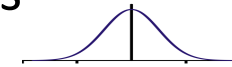
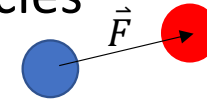
Thermodynamics

- Macroscopic
- Continuum matter
- Differentiable
- Necessary relations based on some axioms
 - Always true for all matter
 - Necessary tool for theory
 - Always present in applications (engineering, chemistry, geoscience...)
- All properties of matter (ΔH_m , ΔS_v , c_v , λ , D) must be measured



Statistical physics

- Microscopic
- Discrete particles
- Mechanics
- Statistical behaviour of simplified models
- Bottom up explanation of thermodynamics
- Properties of model matter (ΔH_m , ΔS_v , c_v , λ , D) can be calculated and measured in simulation



With **thermodynamics**, one can calculate almost everything crudely;

with **kinetic theory**, one can calculate fewer things, but more accurately;

and with **statistical mechanics**, one can calculate almost nothing exactly.

Eugene Wigner



More science quotes at Today in Science History todayinsci.com

- But statistical mechanics defines how to calculate thermodynamic properties from simulation of model systems.
- This makes stat. mech. even more important in the era of computers!
- This is why computation should be part of your thermodyn. & stat. mech. curriculum!

Molecular dynamics simulations

- Method for analyzing the physical movements of atoms and molecules (particles).
 - small systems 10^2 - 10^9 particles.
 - short duration: ns - μ s
 - dynamic motion, not QM effects

- Atoms interacting by pairwise
(neglect QM, manybody...)

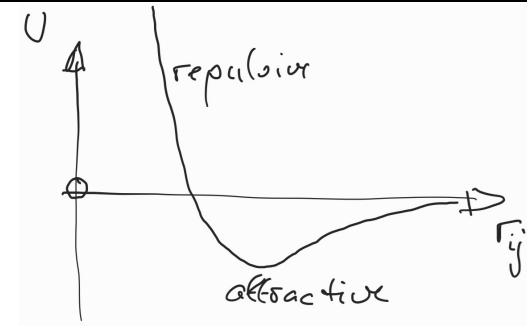
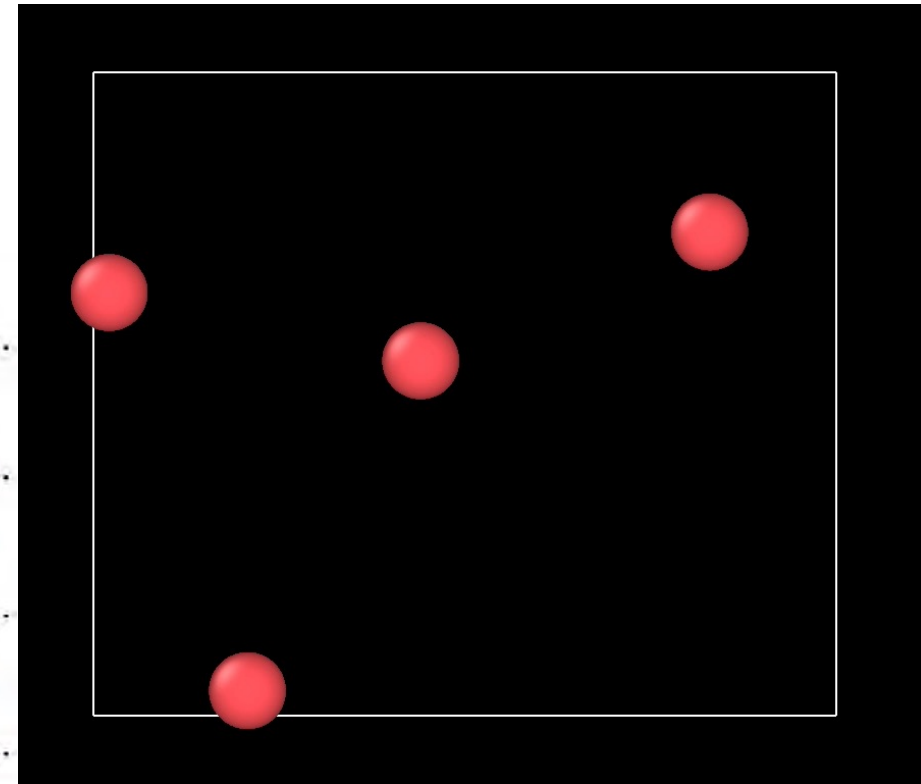
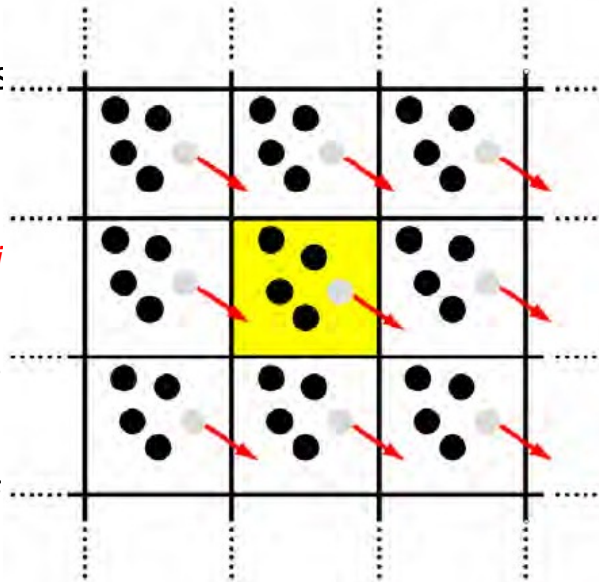
- Force on particle i : $-\mathbf{F}_i = \sum_j$

- Give atoms some initial posi

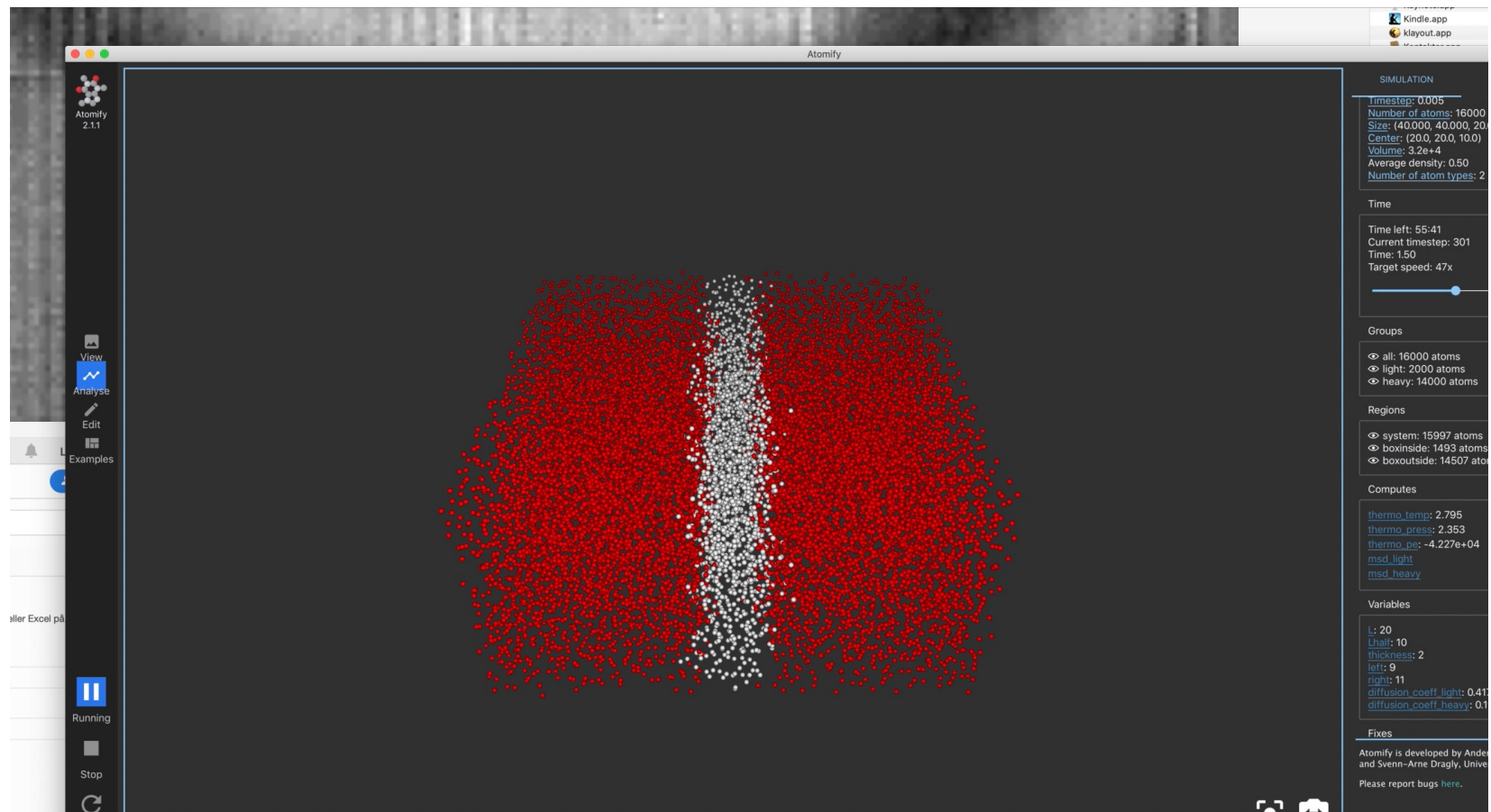
- Solve Newtons equations of $m \frac{d^2\mathbf{r}}{dt^2}$ discretized

- $\mathbf{r}_i(t + dt) = \mathbf{r}_i(t) + \mathbf{v}_i(t) dt$
- $\mathbf{v}_i(t + dt) = \mathbf{v}_i(t) + \mathbf{F}_i(t + dt)dt/m_i$

- Record **trajectories**, $\mathbf{r}_i(t)$, $\mathbf{v}_i(t)$, $\mathbf{F}_i(t)$, $U_i(t)$,... to compute statistical averages

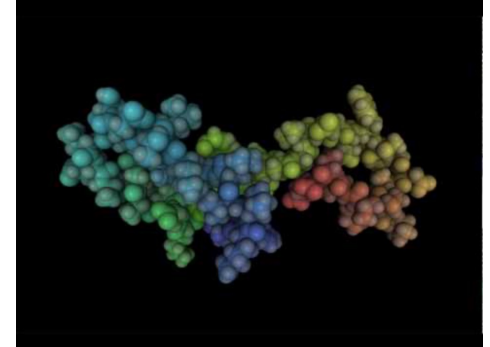
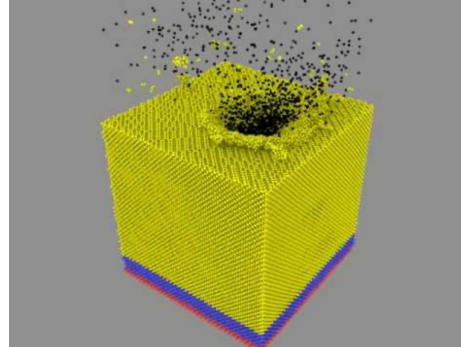
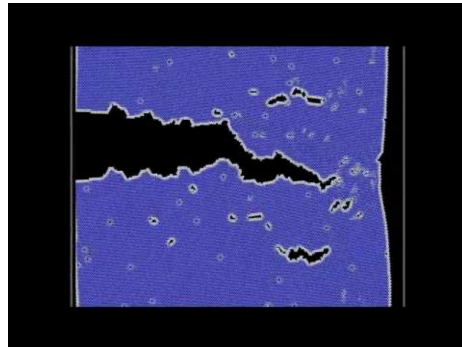
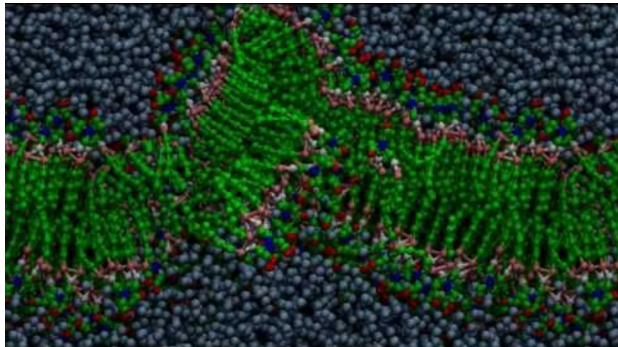


Simulation of a model system



Molecular dynamics simulations

- Method for analyzing the physical movements of atoms and molecules (particles).
 - small systems 10^2 - 10^9 particles.
 - short duration: ns - μ s
 - dynamic motion, not QM effects
- Atoms interacting by pairwise interatomic potentials: $U(\mathbf{r}_{ij})$ (neglect QM, manybody...)
- Force on particle i : $\mathbf{F}_i = \sum_j \frac{d}{dr_{ij}} U(\mathbf{r}_{ij})$
- Give atoms some initial positions \mathbf{r}_i and velocity \mathbf{v}_i
- Solve Newton's equations of motion: $\mathbf{F} = m\mathbf{a} = \frac{d\mathbf{p}}{dt} = m \frac{d^2\mathbf{r}}{dt^2}$ discretized
 - $\mathbf{r}_i(t + dt) = \mathbf{r}_i(t) + \mathbf{v}_i(t) dt$
 - $\mathbf{v}_i(t + dt) = \mathbf{v}_i(t) + \mathbf{F}_i(t) dt / m_i$
- Record **trajectories**, $\mathbf{r}_i(t)$, $\mathbf{v}_i(t)$, $\mathbf{F}_i(t)$, $U_i(t)$,... to compute statistical averages
- Dynamic behaviour of small systems directly visible and measurable

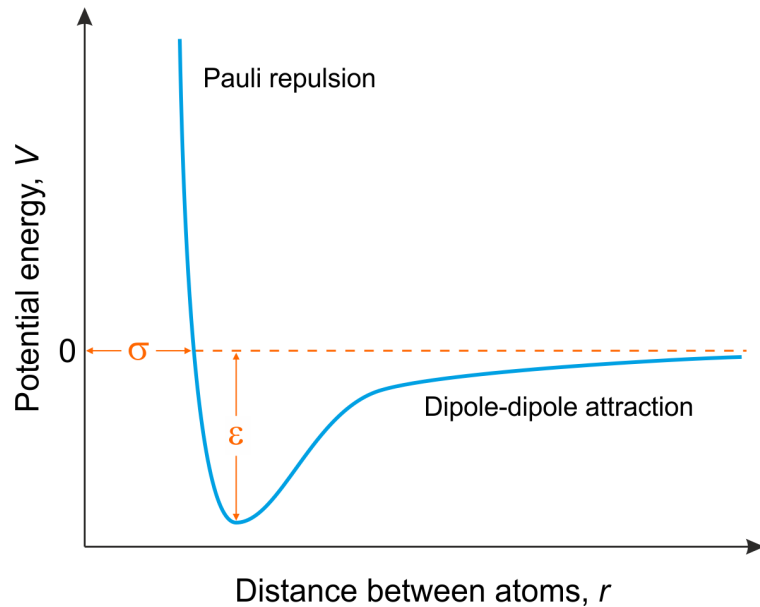


- Some thermodynamical properties (T , P , E , c_V) can be calculated from trajectory averages
- Some thermodynamical properties (S , G , F , c_V ,...) can be calculated from MD «experiments» (perturbation -> response)

Interatomic pair potentials

- Lennard-Jones

$$U(r) = 4\epsilon \left(\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right)$$



Example parameters:

- Argon:
 - $\sigma = 0.34 \text{ nm}$
 - $\epsilon/k_B = 120 \text{ K}$
 - $m = 40 \text{ u} = 6.6\text{e-}26 \text{ kg}$
- Methane:
 - $\sigma = 0.38 \text{ nm}$
 - $\epsilon/k_B = 148 \text{ K}$
 - $m = 16 \text{ u} = 6.6\text{e-}26 \text{ kg}$

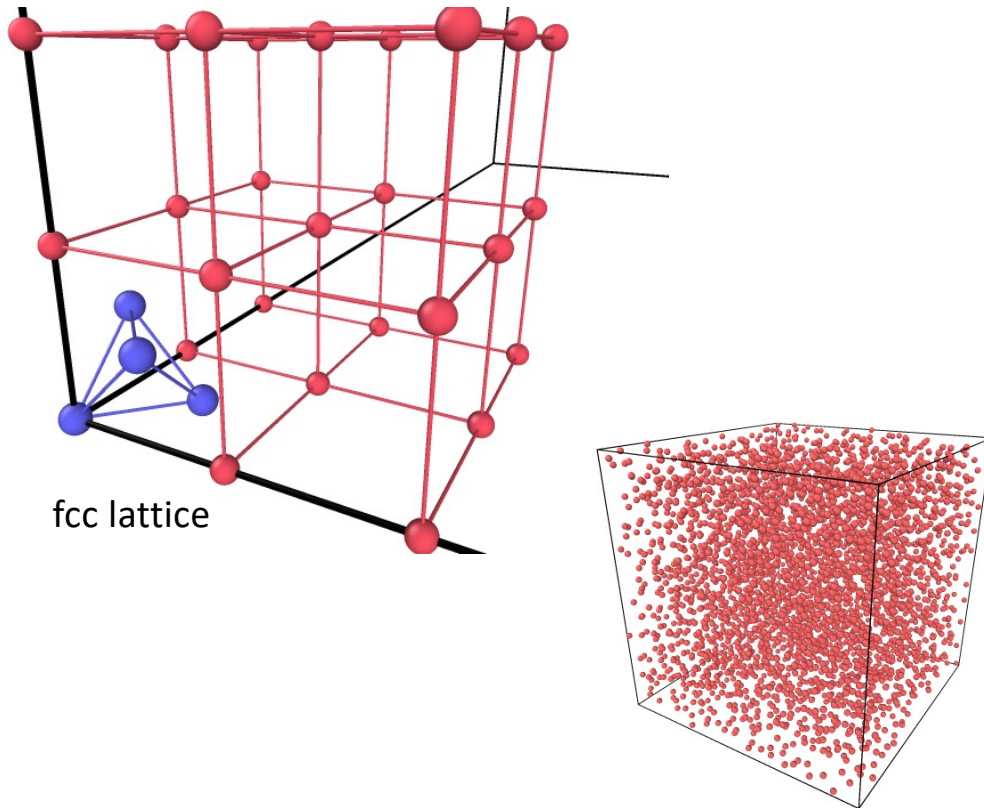
$$1 \text{ u} = 1.66\text{e-}24 \text{ g} = 1 \text{ g}/N_A$$

Reduced (dimensionless) units

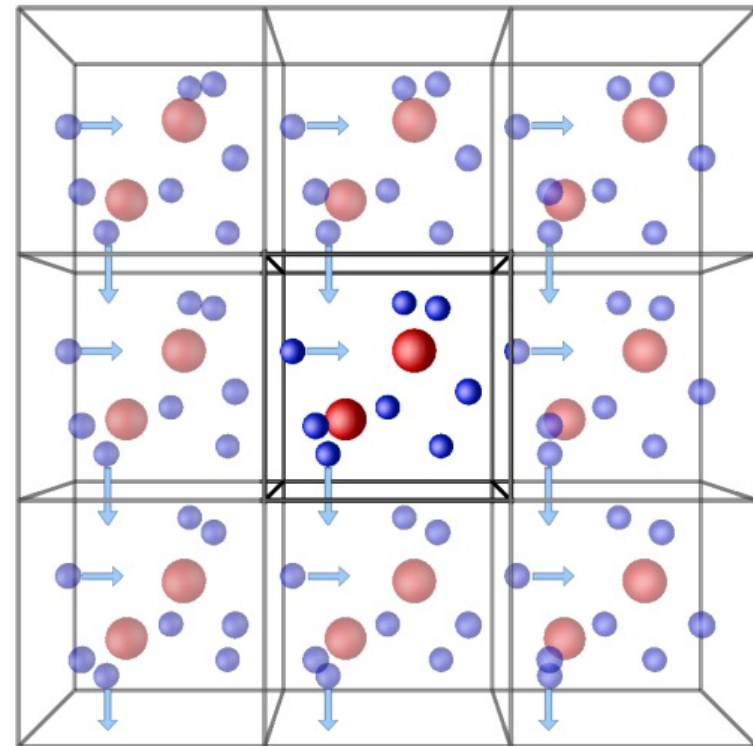
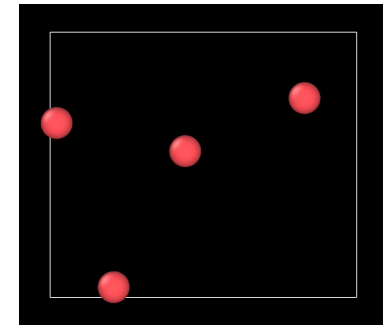
Property	Symbol	Reduced form
Length	r^*	$\frac{r}{\sigma}$
Time	t^*	$\frac{t}{\tau} = t \sqrt{\frac{\epsilon}{m\sigma^2}}$
Temperature	T^*	$\frac{k_B T}{\epsilon}$
Force	f^*	$\frac{f\sigma}{\epsilon}$
Energy	ϕ^*	$\frac{\phi}{\epsilon}$
Pressure	P^*	$\frac{P\sigma^3}{\epsilon}$
Number density	N^*	$N\sigma^3$
Density	ρ^*	$\rho\sigma^3$
Surface tension	γ^*	$\frac{\gamma\sigma^2}{\epsilon}$

Setting up

- Initializing: Lattice, melting

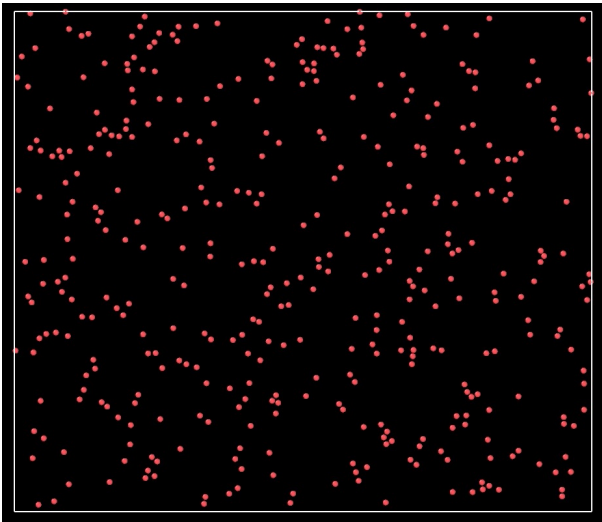


- Periodic boundary conditions

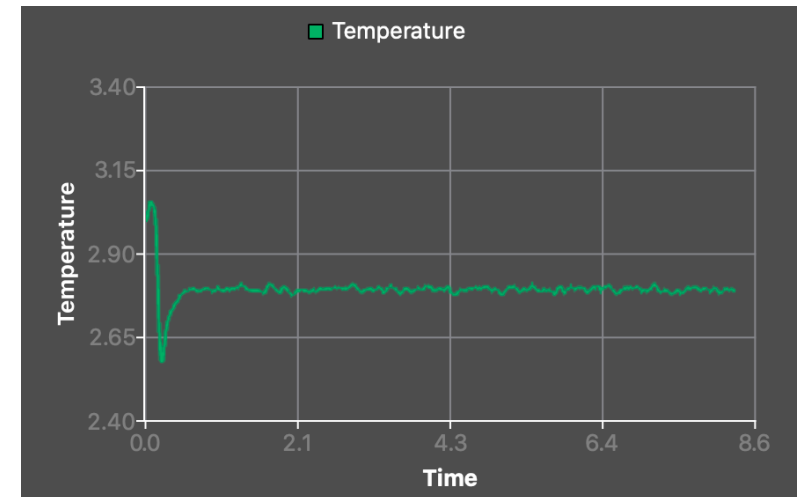


Analyzing trajectories

- Visualizing
 - Ovito
 - VMD
 - ...



- Calculating averages
 - Python
 - Matlab
 - ...



Or integrated: Atomify

LAMMPS Molecular Dynamics Simulator

Large-scale Atomic/Molecular Massively Parallel Simulator

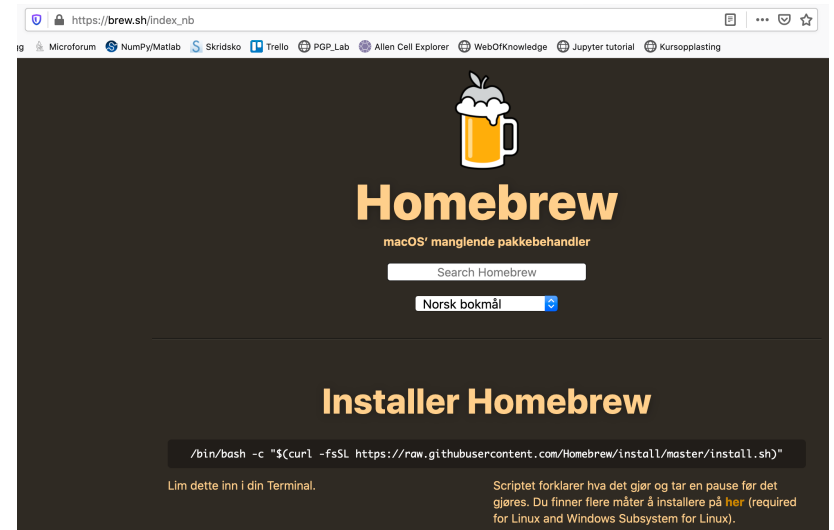
<https://lammps.sandia.gov>

Documentation:

<https://docs.lammps.org/>

Installing Lammps

- <https://docs.lammps.org/Install.html>
- Lines starting with dollar sign \$ means «in a terminal window write the command after the dollar sign»
- I did it on a Mac:
 - Homebrew installation (7 min)
 - `$ /bin/bash -c "$(curl -fsSL https://raw.githubusercontent.com/Homebrew/install/master/install.sh)"`
 - `$ brew install lammps` (3 min)
 - `$ brew test lammps -v` (1 min)
 - `$ brew install openkim-models` (1 min)
- Either build or download executable binaries
- Lammps binary is only ~15 MB



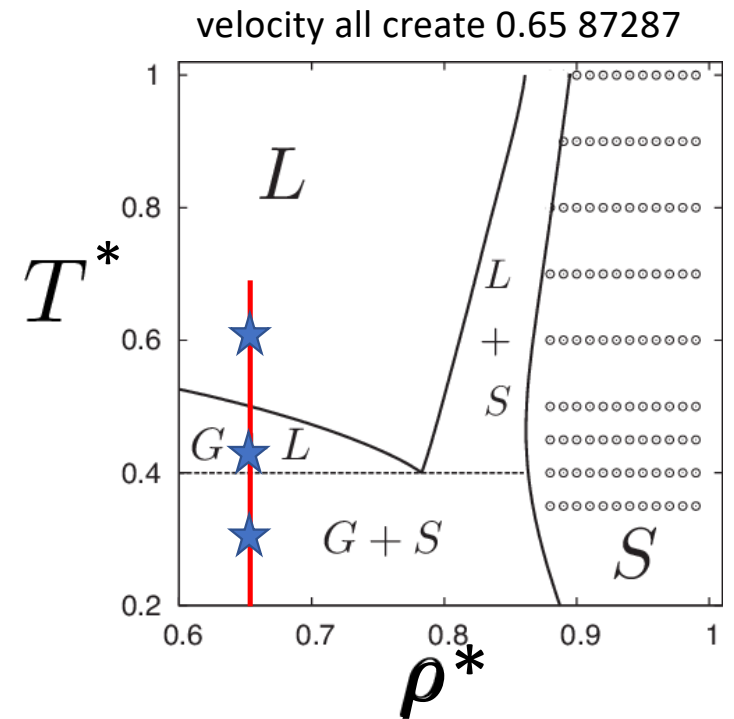
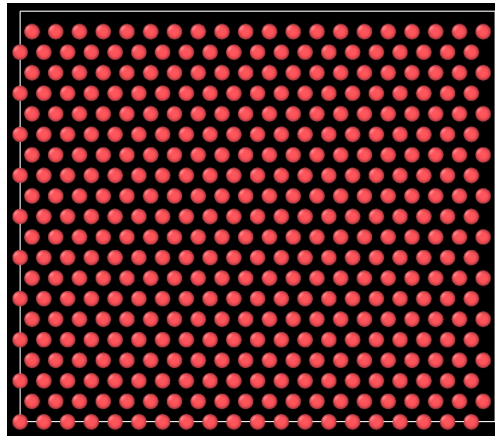
Installing

- Python:
 - Log files: (thermodynamic data): <https://github.com/henriasv/lammps-logfile>
 - Trajectory files (**python not working yet**)
- Matlab:
 - Trajectory files: readdump_all.m readdump_one.m
 - Log files (thermodynamic data):
 - readlog.m this does not work for me...
 - strip everything but the numbers and use `out=load('log.strip');`
- Visualization:
 - download ovito (<https://www.ovito.org/>)

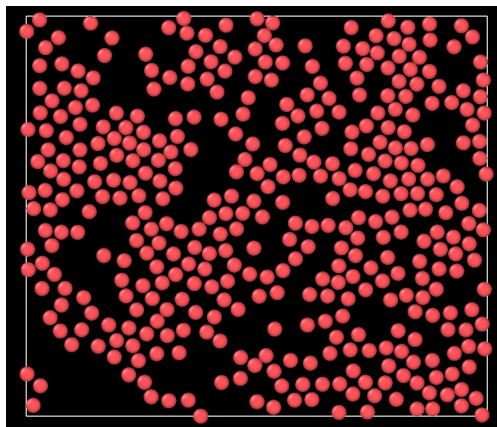
First simulation, run and visualize

- Make a directory for the simulation
- copy **in.myfirstmd** from today's lecture page/
- Run the simulation in a terminal window (Windows: use cmd, not Lammps-shell)
 - `$ lmp_serial < in.myfirstmd` (Mac)
 - `$ lmp -in in.myfirstmd` (Windows)
- Visualization:
 - download ovito (<https://www.ovito.org/>)
 - open ovito
 - File>Load File> **dump.lammpstrj**
- Change parameters and rerun
 - open in.myfirstmd in an editor (not Word!)
 - change temperatures in line «**fix 1 all nvt temp 0.3 0.3 0.3**» 0.3 -> 0.4 and 0.6
 - rerun simulation (`$ lmp_serial < in.myfirstmd`)
 - visualize new run (File>Load File> dump.lammpstrj)

What is the difference between runs?

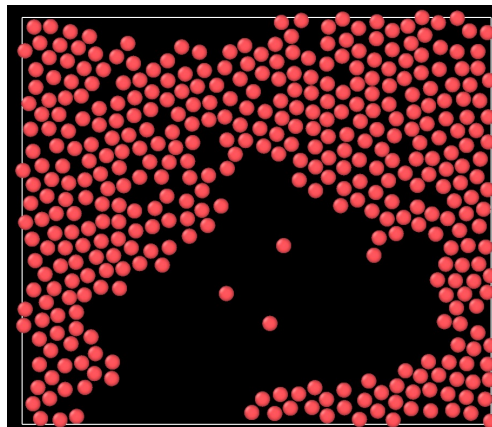


fix 1 all nvt temp 0.6 0.6 0.6



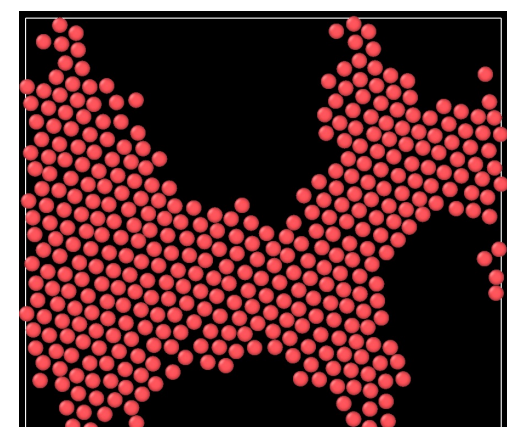
supercritical gas

fix 1 all nvt temp 0.4 0.4 0.4

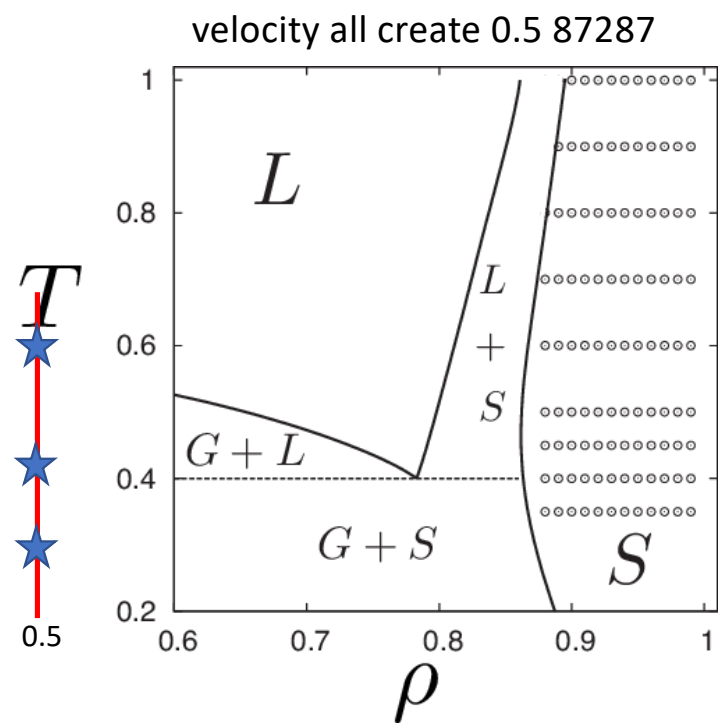
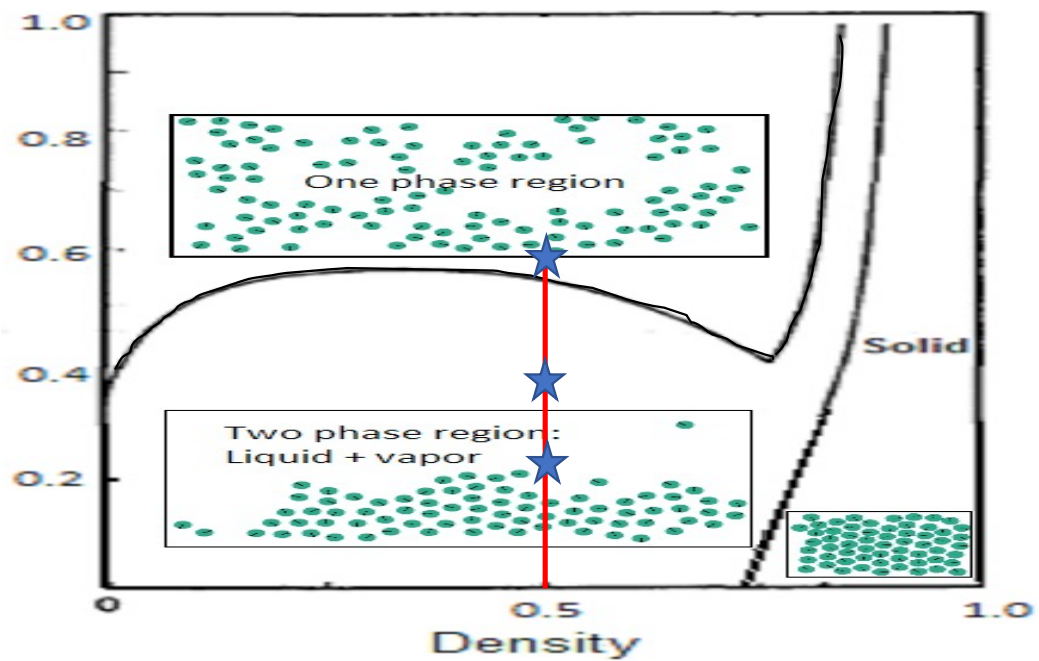


liquid

fix 1 all nvt temp 0.3 0.3 0.3



solid



2D Lennard-Jones

Lammps input file

in.myfirstmd

```
# 2d Lennard-Jones gas
```

```
units lj
```

```
dimension 2
```

```
boundary p p p
```

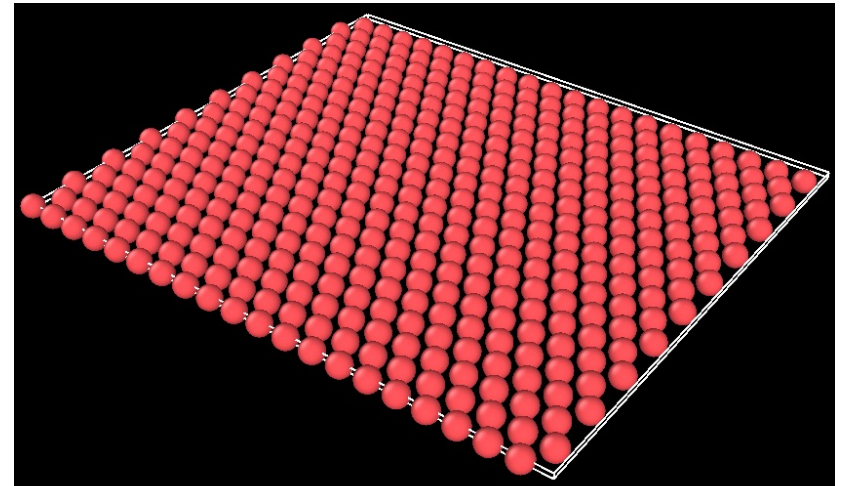
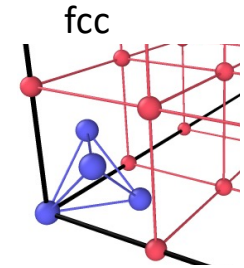
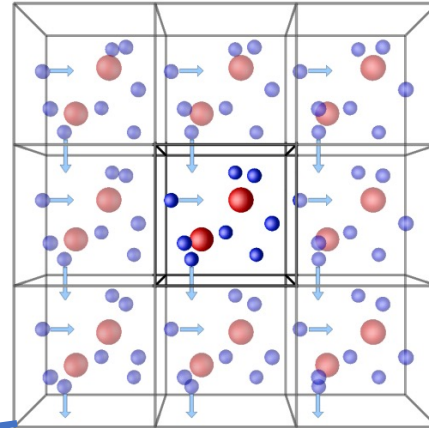
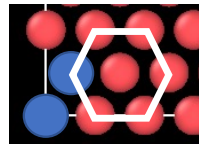
```
atom_style atomic
```

```
lattice hex 0.75
```

```
region simbox block 0 20 0 10 -0.1  
0.1
```

```
create_box 1 simbox
```

```
create_atoms 1 box
```



- **create_atoms** command
 - Syntax: `create_atoms type style args keyword values`
 - Example: `create_atoms 1 region 2`
 - This command creates atoms (or molecules) on a lattice, or a single atom (or molecule), or a random collection of atoms (or molecules), as an alternative to reading in their coordinates explicitly via a [read_data](#) or [read_restart](#) command. A simulation box must already exist, which is typically created via the [create_box](#) command. Before using this command, a lattice must also be defined using the [lattice](#) command

...

`lattice hex 0.75`

`region simbox block 0 20 0 10 -0.1 0.1`

`create_box 1 simbox`

`create_atoms 1 box`

...

- **create_box** command
 - Syntax: `create_box N region-ID keyword value`
 - N = # of atom types to use in this simulation
 - Example: `create_box 2 mybox`
 - This command creates a simulation box based on the specified region. Thus a [region](#) command must first be used to define a geometric domain.
- **region** command
 - Syntax: `region ID style args keyword arg`
 - Example: `region 2 block 0 10 0 10 -0.5 0.05`
 - *block* args = xlo xhi ylo yhi zlo zhi
 - This command defines a geometric region of space.
- **lattice** command
 - Syntax: `lattice style scale keyword values`
 - Example: `lattice hex 0.25`
 - style = *none* or *sc* or *bcc* or *fcc* or *hcp* or *diamond* or *sq* or *sq2* or *hex* or *custom*
 - scale = reduced density rho* (for LJ units)

Lammps input

in.myfirstmd

...

create_atoms 1 box

mass 1 1.0

velocity all create T 2.5 8

pair_style lj/cut 2.5

pair_coeff 1 1 1.0 1.0

fix 1 all nve

dump 1 all custom 10 dump.lammpstrj id type x y z vx vy vz

thermo_style custom time pe ke temp press

thermo 100

run 5000

Dump a snapshot of atom 1 quantities to file dump.lammpstrj every N timesteps in one of several styles (here style=custom)

```
ITEM: TIMESTEP
0
ITEM: NUMBER OF ATOMS
400
ITEM: BOX BOUNDS pp pp pp
0.0000000000000000e+00 2.4816129576055989e+01
0.0000000000000000e+00 2.1491398636470834e+01
-1.2408064788027995e-01 1.2408064788027995e-01
ITEM: ATOMS id type x y z vx vy vz
1 1 0 0 0 1.02727 0.196113 0
2 1 0.620403 1.07457 0 -1.65512 0.826505 0
3 1 1.24081 0 0 -0.616608 -2.40564 0
4 1 1.86121 1.07457 0 0.235659 1.33101 0
5 1 2.48161 0 0 -0.129845 1.05664 0
6 1 3.10202 1.07457 0 2.4403 0.620481 0
7 1 3.72242 0 0 -1.61211 -0.245549 0
8 1 4.34282 1.07457 0 -0.910977 -2.73645 0
9 1 4.96323 0 0 -0.693632 -2.58296 0
10 1 5.58363 1.07457 0 0.705439 -0.91755 0
11 1 6.20403 0 0 -1.76336 2.61689 0
```

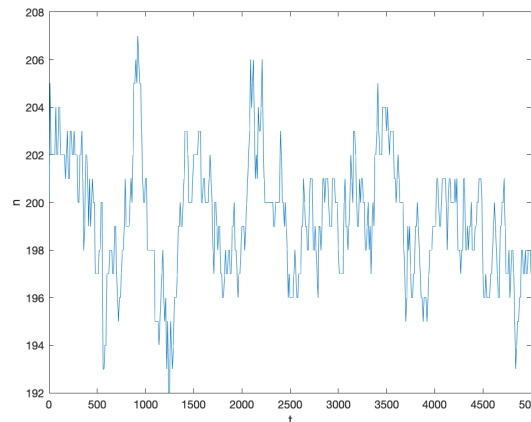
$$12 - \left(\frac{\sigma}{r} \right)^6 \quad r < r_c$$

ϵ	σ	r_c
1.0	1.0	2.5
2.0	2.0	2.5
1.4	1.5	2.5

thermo N = output thermodynamics every N timesteps

First run

- `$ lmp_serial < in.myfirstmd`
- download `readdump_all.m` and `readdump_one.m`
- run data analysis program in Matlab
- Vizualisation:
 - download ovito
 - open ovito
 - load `dump.lammpstrj`



```
data = readdump_all('dump.lammpstrj');
t = data.timestep;
nt = length(t);
nleft = zeros(nt,1);
box = data.x_bound;
halfsize = 0.5*box(:,2);
for it = 1:nt
    xit = data.atom_data(:,3,it);
    jj = find(xit<halfsize(it));
    nleft(it) = length(jj);
end
plot(t,nleft), xlabel('t'), ylabel('n')
```

<https://github.com/henriasv/lammps-logfile>

Python

```
import lammps_logfile
```

```
log = lammps_logfile.File("./log.lammps")
```

```
t = log.get("Time")
```

```
T = log.get("Temp")
```

```
Ek = log.get("KinEng")
```

```
Ep = log.get("PotEng")
```

```
P = log.get("Press")
```

```
import matplotlib.pyplot as plt
```

```
plt.plot(t, Ek+Ep)
```

```
plt.show()
```

```
plt.plot(t, T)
```

```
plt.show()
```

Matlab

Logfile:

```
out=load('log.strip');
```

```
t=out(:,1);
```

```
Ep=out(:,2);
```

```
Ek=out(:,3);
```

```
T=out(:,4);
```

```
P=out(:,5);
```

```
figure(2), plot(t,T)
```

```
xlabel('t'), ylabel('T')
```

```
figure(3), plot(t,Ep+Ek)
```

```
xlabel('t'), ylabel('E_{tot}')
```

Trajectory file:

```
data = readdump_all('dump.lammpstrj');
```

```
t = data.timestep;
```

```
vx=squeeze(data.atom_data(:,5,:));
```

```
vy=squeeze(data.atom_data(:,6,:));
```

```
vxsq=mean(vx.^2);
```

```
vysq=mean(vy.^2);
```

```
T=(vxsq+vysq)/2;
```

```
figure(1), plot(t,vxsq,'-r',t,vysq,'-b',t,T,'-k')
```

```
xlabel('t'), ylabel('T')
```


Computing macroscopic properties

- **compute** command

- Syntax: compute ID group-ID style args
- Define a computation that will be performed on a group of atoms. Quantities calculated by a compute are instantaneous values

- **compute temp** command

- Syntax: compute ID group-ID temp
- Example: compute 1 all temp
- Define a computation that calculates the temperature of a group of atoms.
- The temperature is calculated by the formula $KE = 3/2 N k T$, where KE = total kinetic energy of the group of atoms (sum of $1/2 m v^2$)
- This compute subtracts out degrees-of-freedom due to fixes that constrain molecular motion

Macroscopic properties

- Pressure = Force / Area

- $[P] = [F]/[A] = N/m^2$

- Newtonian mechanics

- $\vec{F} = m\vec{a} = \frac{d\vec{p}}{dt}$

- Used this to calculate pressure of ideal gas:

$$P_x = \frac{1}{A} \sum_i \frac{\Delta p_{x,i}}{\Delta t} = \frac{1}{A} N \frac{m\bar{v}_x^2}{2L} = \frac{Nk_B T}{V} = \rho k_B T$$

- When forces at distance:

- $P = \rho k_B T + \frac{1}{3V} \sum_{i < j} \vec{f}(\vec{r}_{ij}) \cdot \vec{r}_{ij}$

- second term: virial

- **compute pressure** command

- Syntax: compute ID group-ID pressure temp-ID keyword

- Example: compute 1 all pressure thermo_temp

- Define a computation that calculates the pressure of the entire system of atoms.

MD heat capacity experiment

- fix heat command
 - Syntax: `fix ID group-ID heat N eflux`
 - Add non-translational kinetic energy (heat) to a group of atoms in a manner that conserves their aggregate momentum.

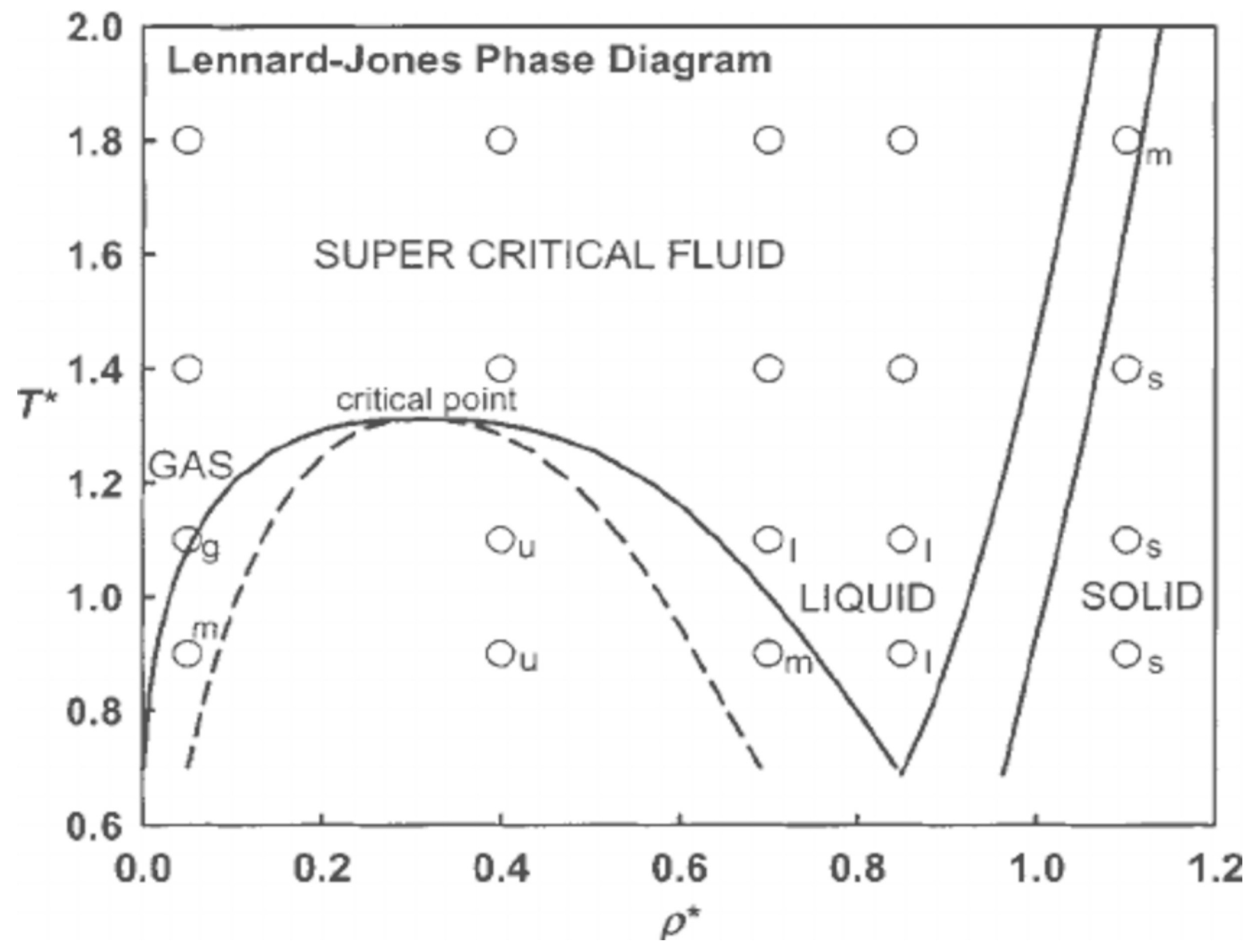
`variable eFlux equal 0.01`

`fix heat all heat 1 ${eFlux} region box`

Summary

- Molecular dynamics: experiments on model systems
- Statistical mechanics is used to find how to measure thermodynamic properties
- LAMMPS: Efficient implementation can run on telephone (Atomify)
- Visualization: Ovito, VMD, Atomify

3D



Integrating equations of motion

- Velocity Verlet

$$\mathbf{v}_i(t + \Delta t/2) = \mathbf{v}_i(t) + \mathbf{F}_i(t)/m_i \Delta t/2$$

$$\mathbf{r}_i(t + \Delta t) = \mathbf{r}_i(t) + \mathbf{v}_i(t + \Delta t/2) \Delta t$$

$$\mathbf{F}_i(t + \Delta t) = -\nabla V(\mathbf{r}_i(t + \Delta t))$$

$$\mathbf{v}_i(t + \Delta t) = \mathbf{v}_i(t + \Delta t/2) + \mathbf{F}_i(t + \Delta t)/m_i \Delta t/2 , ,$$