

# Lecture 9

Canonical ensemble

Average values, equipartition, speed  
distribution, free energy

# Recap+ plan

## Statistical mechanics

### Models

- Two-state:
- Einstein crystal
- Ideal gas
- MD

Interacting systems -----> Equilibrium  
irreversible  
 $\Omega \rightarrow \max \Omega$

### Counting states @constant ( $NVU$ )

$$\Omega(N, n) = \frac{N!}{(N-n)!n!}$$

$$\Omega(N, q) = \frac{(N-1+q)!}{(N-1)!q!}$$

$$\Omega(U, V) = f(N)V^N U^{3N/2}$$

measure & average

### Counting states @constant ( $NVT$ )

Partition function

$$Z = \sum_i e^{-\beta \varepsilon_i}$$

### Counting states @constant ( $\mu VT$ )

Gibbs sum

$$Z_G = \sum_i e^{-\beta(\varepsilon_i - \mu N_i)}$$

Maxwell-Boltzmann speed distribution

$$S = k \ln \Omega$$

$$F = -kT \ln Z$$

$$U - TS = -kT \ln Z_G$$

$$\Delta S_{tot} \geq 0$$

$$T_1 = T_2$$

$$\Delta F \leq 0$$

$$\Delta G \leq 0$$

### Implications:

- Thermodynamic identity  $dU = TdS - PdV$
- 1<sup>st</sup> law  $dU = Q + W$
- Ideal gas:  $PV = NkT$ , equipartition, heat capacity, entropy of mixing, heat, work, state variables, path independence

$$P_1 = P_2$$

$$\mu_1 = \mu_2$$

$$F \equiv U - TS$$

$$G \equiv F + PV$$

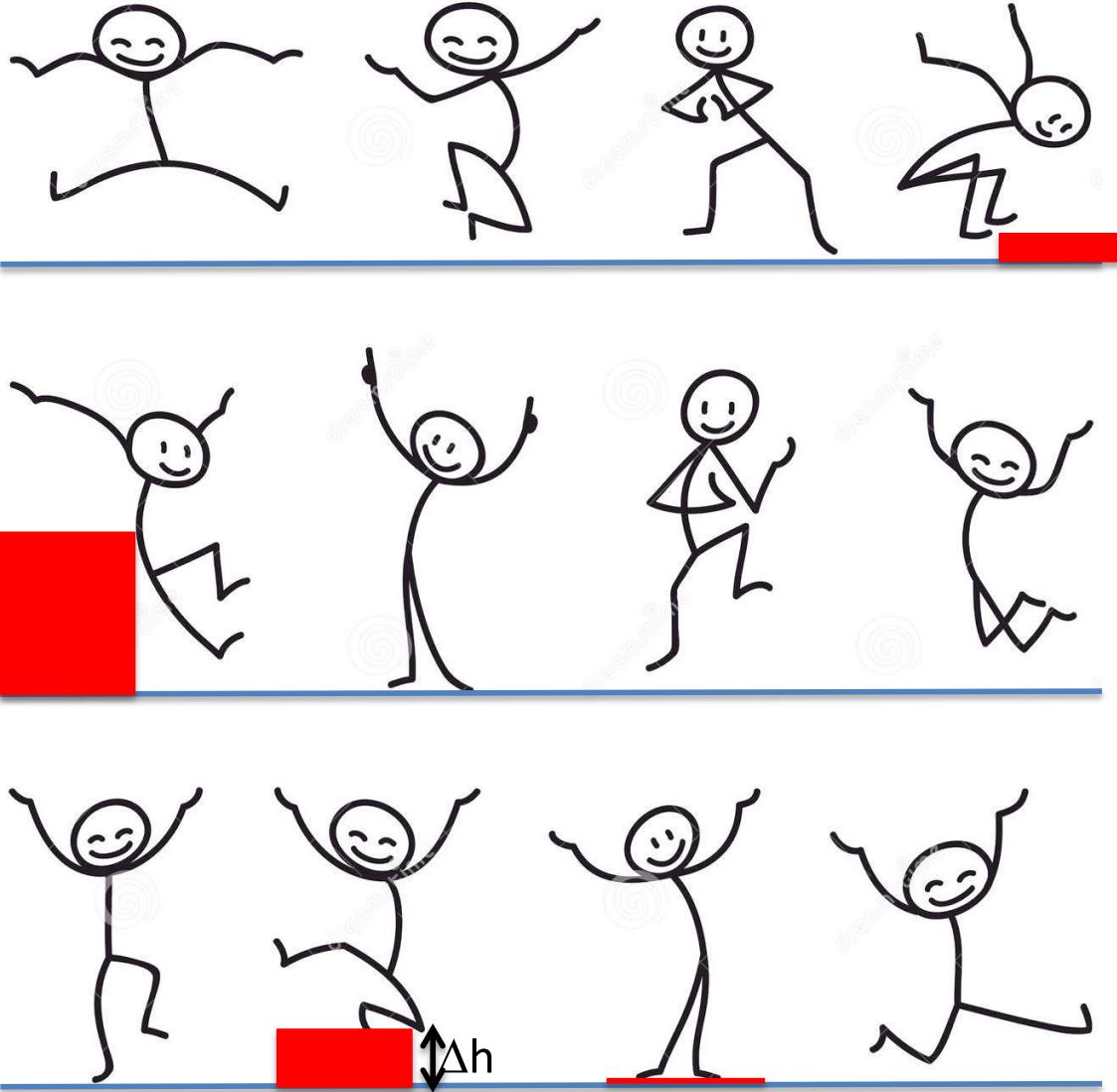
$$dF = -PdV - SdT$$

$$dG = VdP - SdT$$

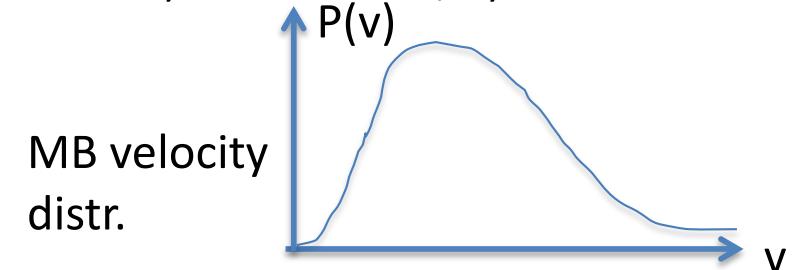
## Thermodynamics

Experiments: measure and model

# Probability of jumping people



To states: 1) On the floor, 2) on red box



MB velocity  
distr.

Mean kinetic energy  
 $m\langle v^2 \rangle / 2 = kT$

Mean potential energy  
 $mg\langle h \rangle = m\langle v^2 \rangle / 2$

Relative probability  
$$\begin{aligned} P(2)/P(1) &= \exp(-E_2/kT)/\exp(-E_1/kT) \\ &= \exp(-(E_2-E_1)/kT) \\ &= \exp(-\Delta E/kT) \\ &= \exp(-mg\Delta h/kT) \end{aligned}$$

# Microcanonical & canonical ensemble

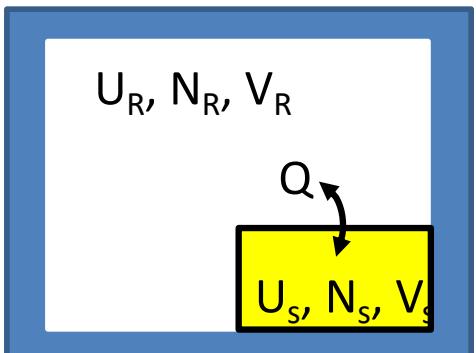


Microcanonical, (NVU) kept constant

Number of microstates in a macrostate: Multiplicity  $\Omega$

Probability of a macrostate:  $P = \Omega / \sum \Omega$

Entropy:  $S = k \ln \Omega$



Canonical, (NVT) kept constant

Exchanges  $Q$  with (NVU) reservoir to keep  $T$  constant

Boltzmann factor:  $e^{-\beta \varepsilon_i}, \quad \beta = \frac{1}{kT}$

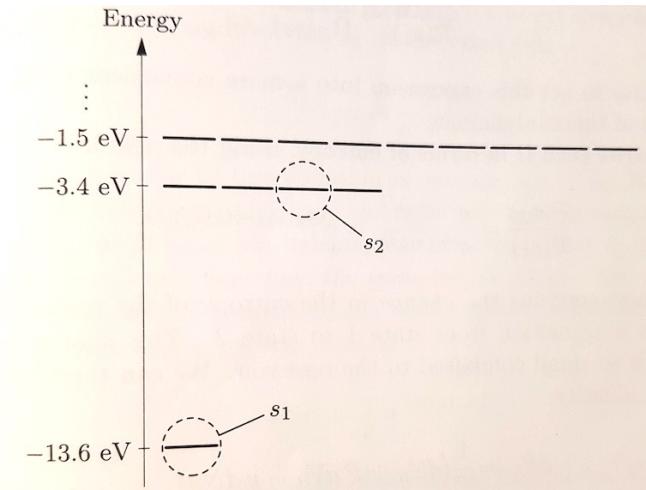
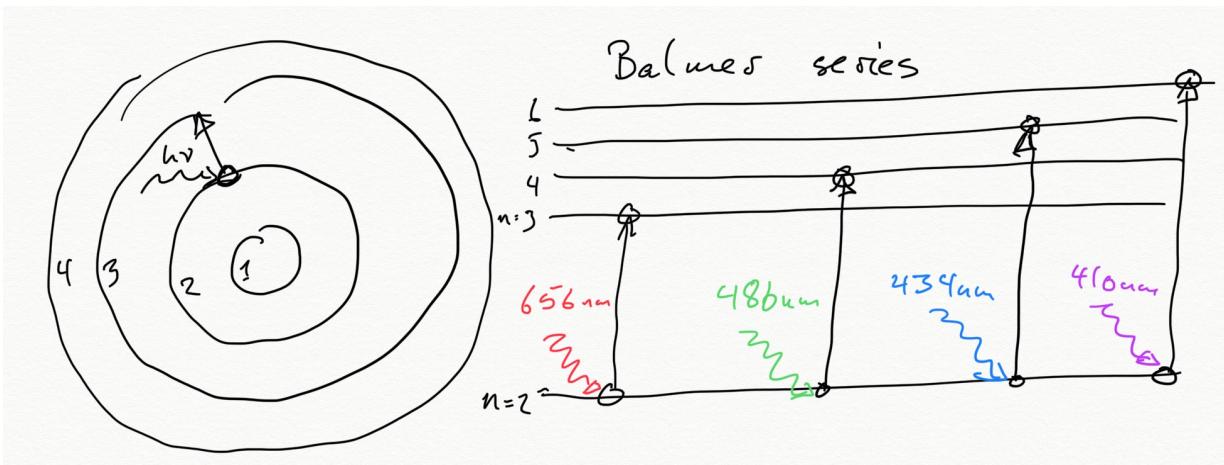
Partition function: sum over

all possible microstates:  $Z = \sum_i e^{-\beta \varepsilon_i}$

Probability of a microstate:  $P_i = \frac{e^{-\beta \varepsilon_i}}{Z}$

Relative probability:  $\frac{P_2}{P_1} = e^{-\beta(\varepsilon_2 - \varepsilon_1)}$

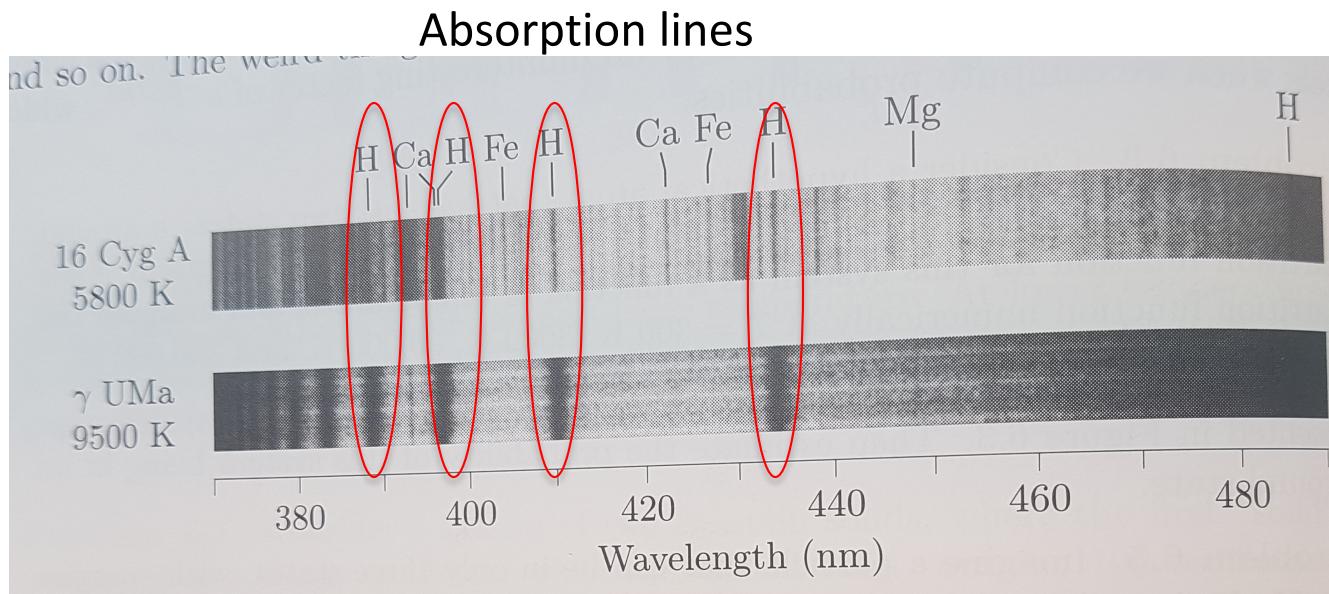
# Example: Hydrogen in the stars



Probability of excited vs. ground state:  $\frac{P(s_2)}{P(s_1)} = e^{-\beta(E_2-E_1)}$  on sun ( $T = 5800\text{K}$ )

$$\frac{(E_2-E_1)}{kT} = \frac{13.6-3.4}{8.6 \cdot 10^{-5} \cdot 5800} = 20.4 \Rightarrow \frac{P(s_2)}{P(s_1)} = 1.4 \cdot 10^{-9}$$

Excited atoms absorbs visible photons when  $n = 2 \rightarrow i$ ,  $i > 2$



5800 K,  $P(s_2)/P(s_1) = 10^{-9}$

9500 K,  $P(s_2)/P(s_1) = 10^{-6}$

# Average values, Energy

Probability of a microstate:  $P(s) = \frac{e^{-\beta E(s)}}{Z}$

Partition function  $Z = \sum_s e^{-\beta E(s)}$

The average value of a fluctuating quantity is the first moment of its distribution:

$$\bar{X} = \langle X \rangle = \sum_s X(s)P(s)$$

For the Boltzmann distribution

$$\bar{X} = \frac{1}{Z(T)} \sum_s X(s)e^{-\beta E(s)}$$

Average energy  $U = \bar{E} = \frac{1}{Z(T)} \sum_s E(s)e^{-\beta E(s)}$

$$\frac{\partial Z}{\partial \beta} = \sum_s \frac{\partial}{\partial \beta} e^{-\beta E(s)} = \sum_s -E(s)e^{-\beta E(s)}$$

=>  $U = \bar{E} = -\frac{1}{Z} \frac{\partial Z}{\partial \beta}$

# Thermodynamics:Helmholtz free energy

## Internal energy

- ▶  $dU = TdS - PdV + \mu dN$

## Helmholtz free energy

- ▶  $F \equiv U - TS$
- ▶  $dF = dU - TdS - SdT$
- ▶ Thermodynamic identity for F:  $dF = -PdV - SdT + \mu dN$
- ▶  $dF = \left(\frac{\partial F}{\partial V}\right)_{N,T} dV + \left(\frac{\partial F}{\partial T}\right)_{N,V} dT + \left(\frac{\partial F}{\partial N}\right)_{V,T} dN$
- ▶  $P = -\left(\frac{\partial F}{\partial V}\right)_{N,T}, S = -\left(\frac{\partial F}{\partial T}\right)_{N,V}, \mu = \left(\frac{\partial F}{\partial N}\right)_{V,T}$

Natural variables: N, V, T, like the Canonical ensemble!

# Average values: Helmholtz free energy

Microcanonical ensemble (NVU):

Fundamental variables:

$$\Omega, S$$

Micro to macro:

$$S = k \ln \Omega$$

Second law:

$$\Delta S \geq 0$$

Derived definitions:

$$T^{-1} \equiv \left( \frac{\partial S}{\partial U} \right)_{N,V}, P \equiv T \left( \frac{\partial S}{\partial V} \right)_{N,U}, \mu \equiv -T \left( \frac{\partial S}{\partial N} \right)_{U,V}$$

Canonical ensemble (NVT):

Fundamental variables:

$$Z, F = U - TS$$

Second law:

$$\Delta F \leq 0$$

Derived definitions:

$$P = - \left( \frac{\partial F}{\partial V} \right)_{N,T}, S = - \left( \frac{\partial F}{\partial T} \right)_{N,V}, \mu = \left( \frac{\partial F}{\partial N} \right)_{T,V}$$

Micro to macro:

$$F = U + T \left( \frac{\partial F}{\partial T} \right)_{N,V}$$

$$\color{red} F = -kT \ln Z$$

$$\frac{\partial \ln Z}{\partial T} = \frac{\partial \beta}{\partial T} \frac{\partial \ln Z}{\partial \beta} = \frac{-1}{kT^2} \frac{1}{Z} \frac{\partial Z}{\partial \beta} = \frac{U}{kT^2}$$

$$\Rightarrow \frac{\partial F}{\partial T} = - \ln Z \frac{\partial kT}{\partial T} - kT \frac{\partial \ln Z}{\partial T} = \frac{F}{T} - \frac{U}{T}$$

$F$  is minimum in equilibrium

$\theta_R$

B

$$\begin{aligned} \text{Total energy } U &= U_R + U_S \\ \text{" entropy" } S &= S_R + S_S \end{aligned}$$

$$S = S_R(U - U_S) + S_S(U_S)$$

$$\approx S_R(U) - U_S \left( \frac{\partial S_R}{\partial U} \right)_{K, N} + S_S(U_S)$$

$$\Rightarrow S \approx S_R(U) - \frac{U_S}{T} + S_S(U_S)$$

$$TS = TS_R(U) - U_S + TS_S(U_S)$$

$\underbrace{\quad}_{- F_S}$

Equilibrium  $U$  is constant  $\Rightarrow S_R(U)$  is constant  
 $S_{\max} \Leftrightarrow F$  is minimum.

# Paramagnetism

One fluctuating spin

- magnet. moment:  $\mu$

- spin:  $s = \pm \frac{1}{2}$

- energy:  $\epsilon_s = -2s\mu B$

Probability up:

$$P_{\uparrow} = \frac{1}{Z} e^{-\beta\epsilon_s} = \frac{1}{Z} e^{\beta\mu B}$$

Probability down:

$$P_{\downarrow} = \frac{1}{Z} e^{-\beta\epsilon_s} = \frac{1}{Z} e^{-\beta\mu B}$$

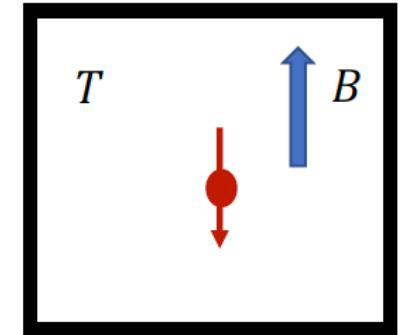
Partition function:

$$Z = e^{-\beta\mu B} + e^{\beta\mu B} = 2\cosh(\beta\mu B)$$

Average energy:  $\bar{\epsilon} = -\frac{1}{Z} \frac{\partial Z}{\partial \beta} = -\frac{2\mu B \sinh \beta\mu B}{2 \cosh(\beta\mu B)} = -\mu B \tanh(\beta\mu B)$

Average mag. mom.:  $\bar{\mu} = (+\mu)P_{\uparrow} + (-\mu)P_{\downarrow} = \frac{\mu}{Z} (e^{\beta\mu B} - e^{-\beta\mu B})$   
 $= \mu \tanh(\beta\mu B)$

N spins, magnetization:  $M = N\bar{\mu}$ , energy:  $U = -N\mu B \tanh(\beta\mu B)$



# Equipartition theorem

Systems with quadratic degrees of freedom

$$E(q) = cq^2$$

Partition function  $Z = \sum_q e^{-\beta cq^2} = \frac{1}{\Delta q} \sum_q \Delta q e^{-\beta cq^2}$

$$\Delta q \ll kT, \quad = \frac{1}{\Delta q} \int_{-\infty}^{\infty} dq e^{-\beta cq^2} = \frac{1}{\Delta q} \sqrt{\frac{\pi}{\beta c}} = C \beta^{-\frac{1}{2}}$$

Mean energy  $\bar{E} = U = -\frac{1}{Z} \frac{\partial Z}{\partial \beta} = -\frac{\beta^{\frac{1}{2}}}{C} \left( -\frac{1}{2} C \beta^{-\frac{3}{2}} \right) = \frac{1}{2} kT$

Kinetic energy  $\overline{E_k} = \frac{1}{2} m \overline{v^2} = \frac{1}{2} m (\overline{v_x^2} + \overline{v_y^2} + \overline{v_z^2}) = \frac{3}{2} kT$

RMS speed  $v_{rms} = \sqrt{\overline{v^2}} = \sqrt{\frac{3kT}{m}}$

# Partition function of many particles

## Getting the counting right

1 particle, N states     $Z_1 = \sum_s^N e^{-\beta \epsilon_s}$

Identical particles

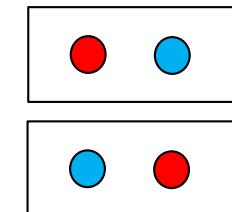
Noninteracting / independent particles

no extra states are created due to having more particles

**Distinguishable particles**

2 particles, N states: total energy  $\sum_i^2 \epsilon_{i,s} = \epsilon_{1,s} + \epsilon_{2,s}$

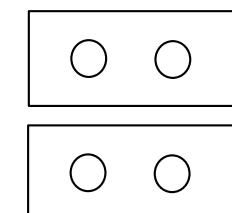
$$\begin{aligned} Z_2 &= \sum_s e^{-\beta(\epsilon_{1,s} + \epsilon_{2,s})} \\ &= \sum_s e^{-\beta \epsilon_{1,s}} e^{-\beta \epsilon_{2,s}}, \quad s = (s_1, s_2) \\ &= \sum_{s_1} \sum_{s_2} e^{-\beta \epsilon_{1,s_1}} e^{-\beta \epsilon_{2,s_2}} \\ &= Z_1 Z_1 = Z_1^2 \\ Z_N &= Z_1^N \end{aligned}$$



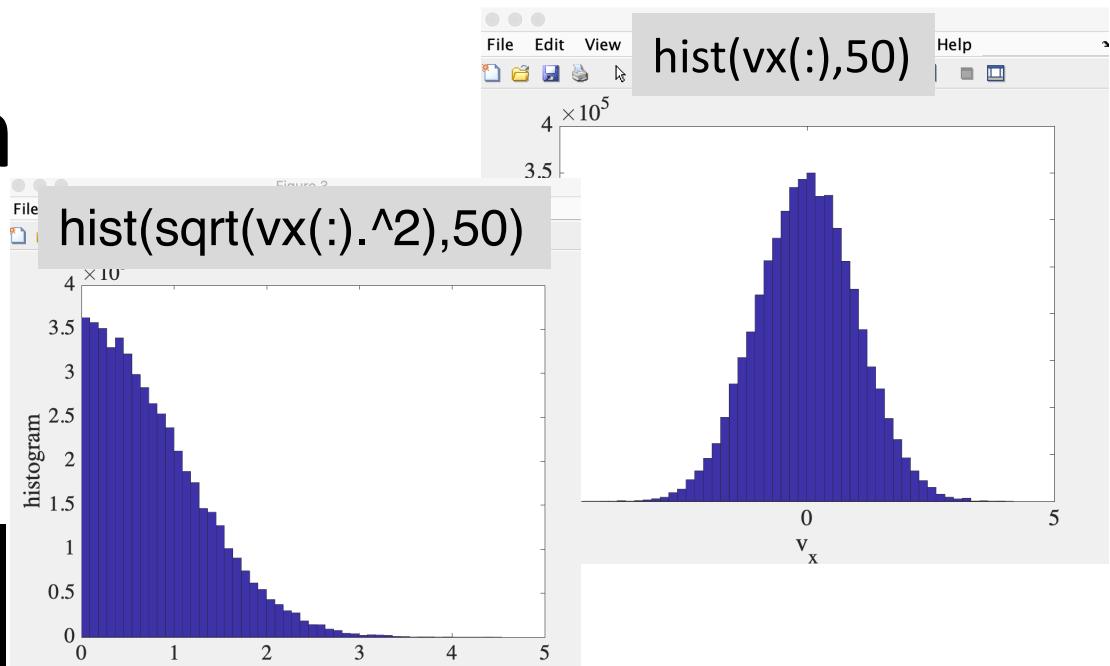
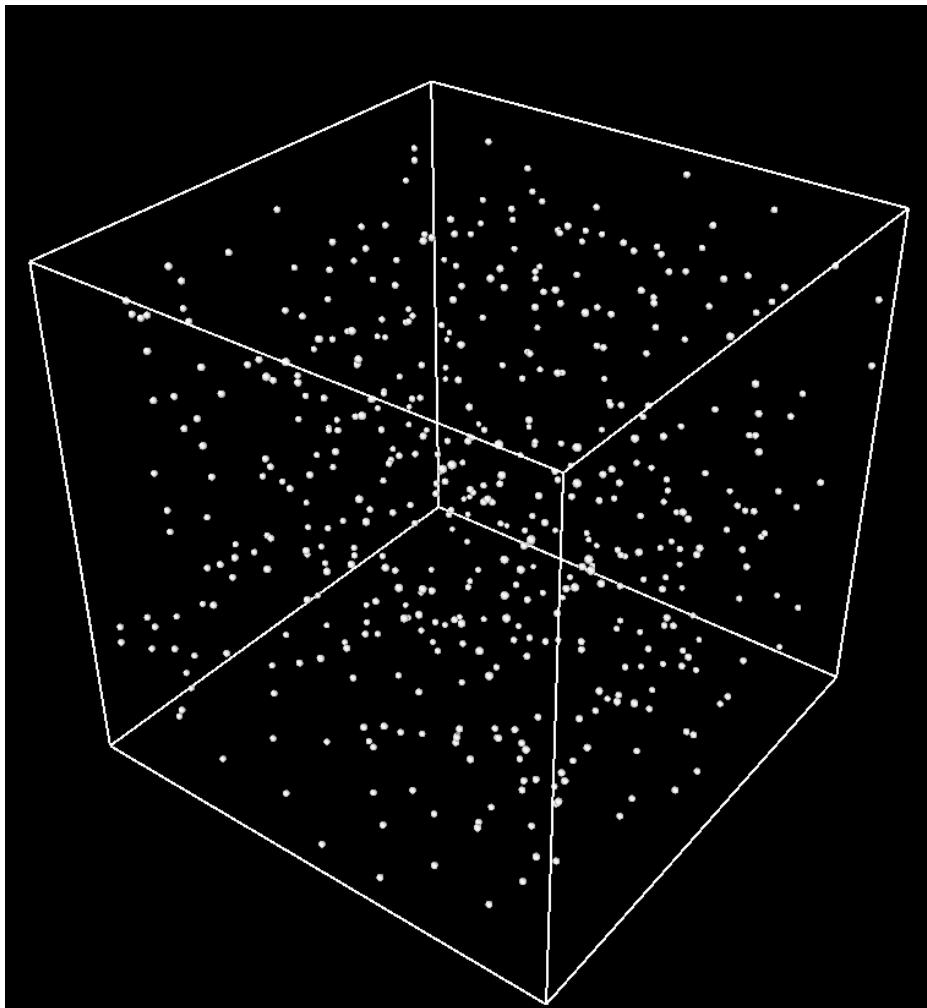
**Indistinguishable particles**

$$Z_2 = \frac{1}{2} Z_1^2$$

$$Z_N = \frac{1}{N!} Z_1^N$$

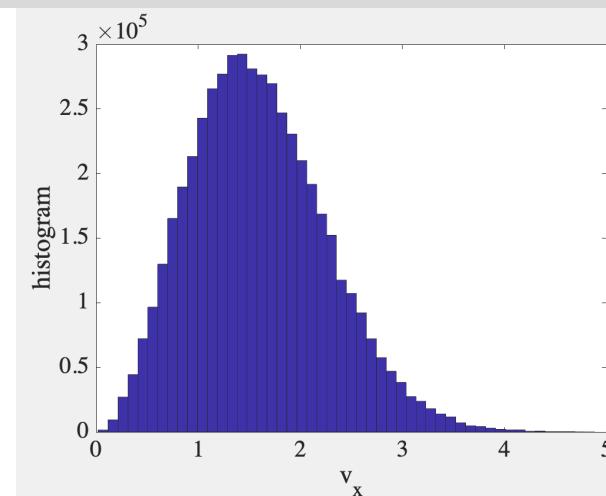


# Maxwell-Boltzmann speed distribution



- Average velocity  $\bar{v}_x = 0$
- Velocity is a vector, speed is a scalar
- Velocity is in 3D
- Probability of speed  $v$ ,  $P(v) = 0$
- Probability distribution:  $P(v: v + dv) = D(v)dv$

hist(sqrt(vx(:).^2+vy(:).^2+vz(:).^2),50)



```

# Define variables
# Length of sides of cubic box
variable L equal 5
# Temperature:
variable T equal 1
# Density:
variable rho equal 0.001
# Pressure:
variable P equal 0.01
# Set Pconst=1 for pressure constant, Pconst=0 for constant volume
variable Pconst equal 0

units lj
dimension 3
lattice fcc ${rho}
region simbox block 0 $L 0 $L 0 $L
create_box 1 simbox
create_atoms 1 box

mass 1 1.0
velocity all create $T 87287
pair_style lj/cut 2.5
pair_coeff 1 1 1.0 1.0 2.5

if "{$Pconst} == 0" then &
"fix nvt all nvt temp $T $T 1" &
else &
"fix npt all npt temp $T $T 1 iso $P $P 1"
neigh_modify every 1 delay 0 check yes

#Thermalization run
run 1000

if "{$Pconst} == 0" then &
"unfix nvt" &
"fix nve all nve" &
else &
"unfix npt" &
"fix nph all nph iso $P $P 1"

#Start heat addition and run to stabilize
variable eFlux equal 1
#fix heat all heat 1 ${eFlux} region simbox
run 1000

#Define new thermodynamic output and trajectory dump and start production run
thermo_style custom step temp epair etotal press
thermo 100
dump 1 all custom 100 dump.lammpstrj id x y z vx vy vz
run 1000000

```

### \$ lmp\_serial < heatcapLJ.in

```

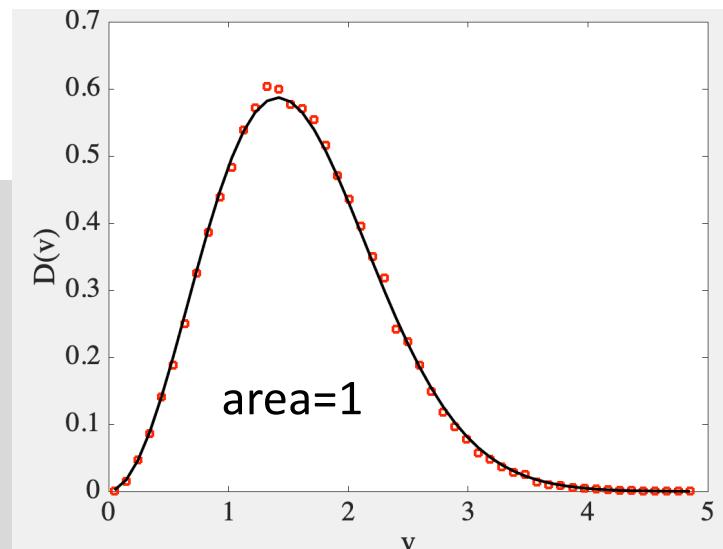
vsq=vx.^2+vy.^2+vz.^2;
speed=sqrt(vsq);
h = histogram(speed(:),50);
hv=h.Values;
s=sum(hv);
hbe=h.BinEdges;
dv=h.BinWidth;
hbc=hbe+dv/2;
v=hbc(1:end-1);
Dofv=hv/s/dv;
norm=sum(Dofv)*dv;
meanv=sum(v.*Dofv*dv)

```

$v_{rms} = \sqrt{\langle v_x^2 + v_y^2 + v_z^2 \rangle}$

$\text{plot}(v, Dofv, '.', v, 2*v.^2.*exp(-v.^2./2)./sqrt(2*pi), '-')$

$$kT = 1, m = 1 \Rightarrow$$



$$v$$

$$D(v)$$

$$\sum D(v)dv = 1$$

$$\bar{v} = \sum v D(v)dv = 1.5966$$

$$v_{rms} = \sqrt{\langle v_x^2 + v_y^2 + v_z^2 \rangle} = 1.7287$$

$$D(v) = \frac{2}{\sqrt{2\pi}} v^2 e^{-\frac{v^2}{2}}$$

Distribution function:  $[D(v)] = [v^{-1}]$

$$[P] = \left[ \int D(v)dv \right] = 1$$

# Maxwell-Boltzmann speed distribution

Probability distribution:

$$P(v: v + dv) = D(v)dv$$

Boltzmann probability:

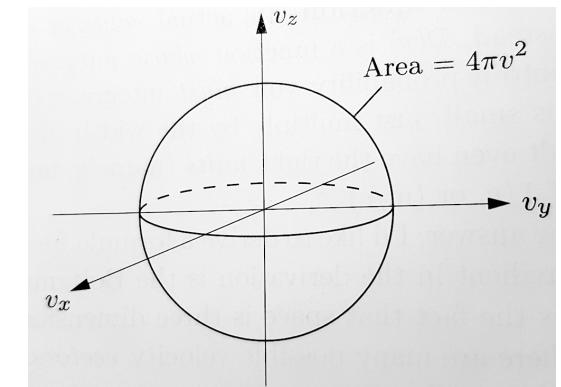
$$D(v)dv \propto e^{-\frac{E_k(v)}{kT}} dn = e^{-\frac{mv^2}{2kT}} dn$$

Number of vectors of speed  $v$ :

$$dn \propto 4\pi v^2 dv$$

Normalization

$$\int_0^\infty D(v)dv = 1$$



$$C \int_0^\infty e^{-\frac{mv^2}{2kT}} 4\pi v^2 dv = 1 \Rightarrow C = \left(\frac{m}{2\pi kT}\right)^{3/2}$$

M-B speed distribution

$$D(v) = \left(\frac{m}{2\pi kT}\right)^{3/2} 4\pi v^2 e^{-\frac{mv^2}{2kT}}$$

# Maxwell-Boltzmann speed distribution

Average speed:

$$\bar{v} = \int_0^{\infty} v D(v) dv = \sqrt{\frac{8kT}{\pi m}} (= 1.5958)$$

meanv=sum(v.\*Dofv\*dv)= 1.5966

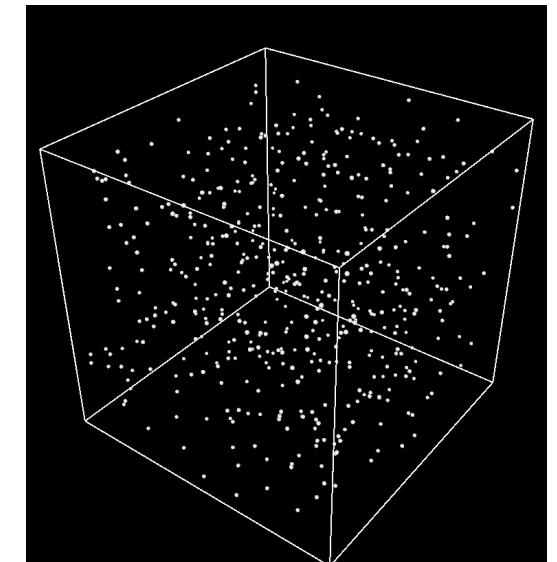
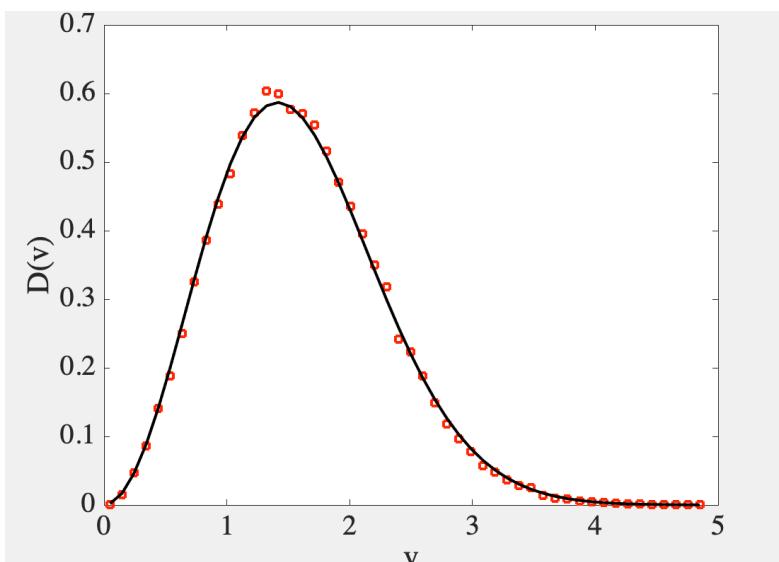
RMS speed:

$$v_{rms} = \sqrt{\int_0^{\infty} v^2 D(v) dv} = \sqrt{\frac{3kT}{m}} (= 1.7321)$$

vrms=sqrt(mean(vsq(:)))=1.7287

Most probable speed

$$\frac{\partial D(v)}{\partial v} = 0 \implies v_{mp} = \sqrt{\frac{2kT}{m}} (= 1.4142)$$



# Ensembles, counting and probabilities



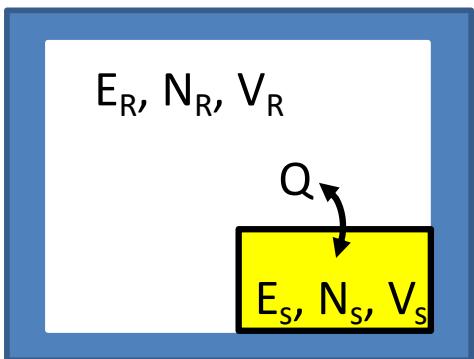
Microcanonical,  $(N, V, U)$  kept constant

Number of microstates,  $s$ , in a macrostate: Multiplicity  $\Omega$

Probability of macrostate:  $P = \Omega / \sum \Omega$

Probability of a microstate for a given macrostate:  $P(s) = 1/\Omega$

Entropy:  $S = k \ln \Omega$



Canonical,  $(N, V, T)$  kept constant

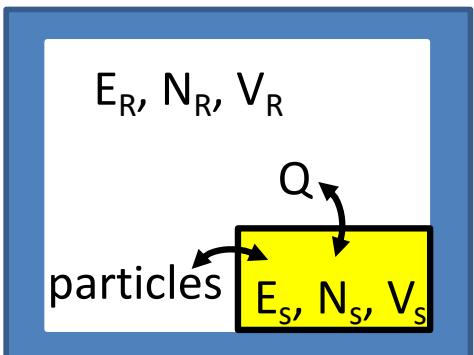
Exchanges  $Q$  with  $(N, V, U)$  reservoir to keep  $T$  constant

Boltzmann factor:  $e^{-\beta \varepsilon_i}$

Partition function: sum over all possible microstates:  $Z = \sum_i e^{-\beta \varepsilon_i}$

Probability of a microstate:  $P_i = e^{-\beta \varepsilon_i} / Z$

Free energy: Helmholtz,  $F = -kT \ln Z$



Grand canonical,  $(\mu, V, U)$  kept constant

Exchanges  $Q$  & particles with  $(N, V, U)$  reservoir to keep  $T$  &  $\mu$  constant

Gibbs factor:  $e^{-\beta(\varepsilon_i - \mu N_i)}$

Gibbs sum: sum over all possible microstates:  $Z_G = \sum_i e^{-\beta(\varepsilon_i - \mu N_i)}$

Probability of a microstate:  $P_i = e^{-\beta(\varepsilon_i - \mu N_i)} / Z_G$

Free energy: Grand potential,  $\Phi = -kT \ln Z_G = U - TS - \mu N$