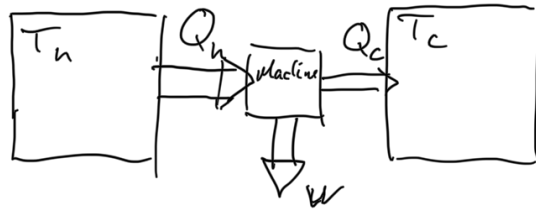


Engines & refrigerators



Reservoirs: $dT=0$

Machine is cyclic:
 $\oint dU, dS$

Efficiency $e = \frac{W}{Q_h} \leq 1$

1st law: $Q_h = Q_c + W$

$$e = \frac{Q_h - Q_c}{Q_h} = 1 - \frac{Q_c}{Q_h} \quad Q_c \rightarrow 0$$

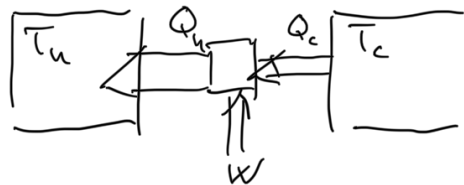
2nd law: Entropy with heat flow $dS = \frac{Q}{T}$

Machine $dS = 0$ $S_{out} \geq S_{in}$

$$\frac{Q_c}{T_c} \geq \frac{Q_h}{T_h} \Rightarrow \frac{Q_c}{Q_h} \geq \frac{T_c}{T_h}$$

$$e = 1 - \frac{Q_c}{Q_h} = 1 - \frac{T_c}{T_h}$$

Refrigerator



cyclic $dU=0$
 machine $dS=0$

1st law $W + Q_c = Q_h$

2nd law $\frac{Q_h}{T_h} \geq \frac{Q_c}{T_c} \Rightarrow \frac{Q_h}{Q_c} \geq \frac{T_h}{T_c}$

Fridge e : Coeff of performance $COP = \frac{Q_c}{W} = \frac{Q_c}{Q_h - Q_c} = \frac{1}{\frac{Q_h}{Q_c} - 1}$

maximize $\frac{Q_c}{W} \leq \frac{1}{\frac{T_h}{T_c} - 1} = \frac{T_c}{T_h - T_c}$

$COP \leq \frac{T_c}{T_h - T_c} \leq 1$

Heat pump maximize $\frac{Q_h}{W}$ $e < \frac{T_h}{T_h - T_c} > 1$

Carnot cycle $e \geq 1 - \frac{T_c}{T_h}$

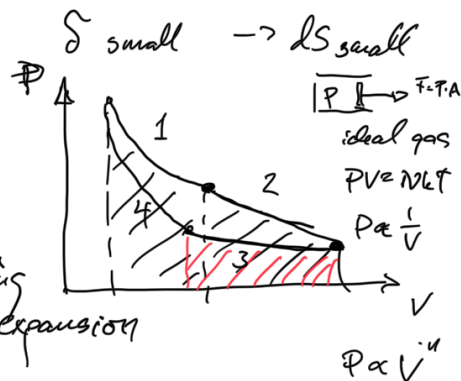
Work medium $Q \rightarrow \boxed{\dots} \rightarrow W$ $Q \rightarrow \boxed{\text{gas}} \rightarrow PV = NkT$

Least entropy production $Q = T dS$
 $dS = \frac{Q}{T}$

1st heat \rightarrow gas at const. temp.

$$T_h \quad T_g = T_h - \delta$$

isothermal + expansion
 $W = -P \cdot \Delta V$



2 Let gas work without producing heat $Q = 0$ adiabatic expansion

3 Compress isothermally $W = P \Delta V$ $T_g = T_c + \delta$

4 Do work on gas without heat $Q = 0$ adiab. compression

Carnot cycle $e = 1 - \frac{T_c}{T_h}$

