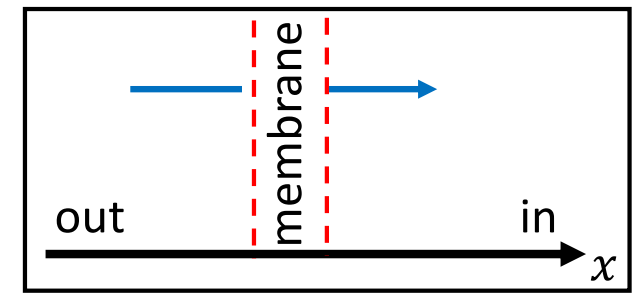


11.1-2 Ion pumping -> sodium anomaly in all animal cells



Fluxes, $J_i, J_{p,i}$ of ion species i , **out -> in** positive

Forces, $\Delta\mu_i = kT\Delta \ln c_i + q_i\Delta V + v_i\Delta P$, **in-out** positive

Linear transport: $J_i = J_{p,i} - g_i\Delta\mu_i$
 $= J_{p,i} - g_i\Delta \ln c_i + g_iq_i\Delta V + g_iv_i\Delta P$

steady state $= 0$

$J_{p,i} = \alpha_i J_p$ ion pump flux, $\alpha_{Na^+} = 3$, $\alpha_{K^+} = -2$, $J_p > 0$

J_i flux through membrane

g_i conductance through membrane

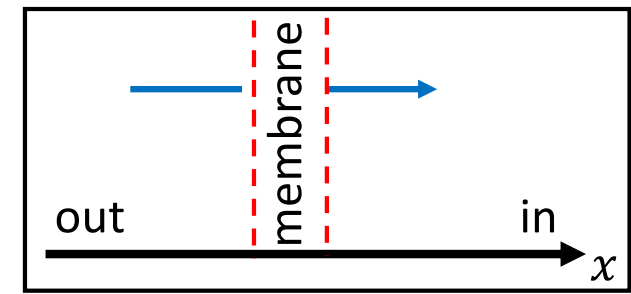
q_i charge of ion i

ΔV (electrical) membrane potential

$v_i = c_i^{-1}$ (partial) volume of ion i (ideal mixture)

$\Delta P = 0$ assume no pressure difference

11.1-2 Ion pumping -> sodium anomaly in all animal cells



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charge neutrality inside:
$$c_{Na^+,in} + c_{K^+,in} - c_{Cl^-,in} - \frac{\rho_m}{e} = 0$$

flux charge neutrality:
$$j_{Na^+} + j_{K^+} + j_{Cl^-} = 0 \tag{1}$$

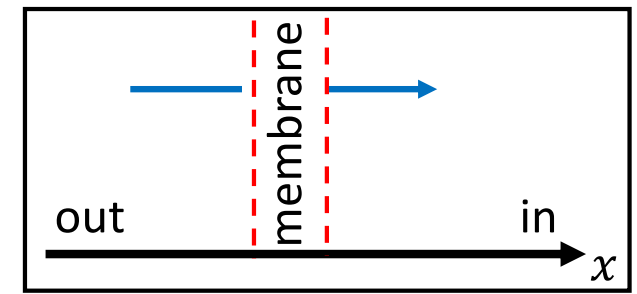
steady state:
$$\frac{3j_p}{g_{Na^+}} = kT \ln \frac{c_{Na^+,o}}{c_{Na^+,i}} + e\Delta V \tag{2}$$

$$-\frac{2j_p}{g_{K^+}} = kT \ln \frac{c_{K^+,o}}{c_{K^+,i}} + e\Delta V \tag{3}$$

$$0 = kT \ln \frac{c_{Cl^-,o}}{c_{Cl^-,i}} - e\Delta V \tag{4}$$

4 equations, 4 unknowns: $c_{Na^+,i}, c_{K^+,i}, c_{Cl^-,i}, \Delta V$

11.1-2 Ion pumping -> sodium anomaly in all animal cells



charge neutrality inside: $c_{Na+,in} + c_{K+,in} - c_{Cl-,in} - \frac{\rho_m}{e} = 0$

flux charge neutrality: $j_{Na+} + j_{K+} + j_{Cl-} = 0$ (1)

steady state: $\frac{3j_p}{g_{Na+}} = kT \ln \frac{c_{Na+,o}}{c_{Na+,in}} + e\Delta V$ (2)

$$-\frac{2j_p}{g_{K+}} = kT \ln \frac{c_{K+,o}}{c_{K+,in}} + e\Delta V$$
 (3)

$$0 = kT \ln \frac{c_{Cl-,o}}{c_{Cl-,in}} - e\Delta V$$
 (4)

4 equations, 4 unknowns: $c_{Na+,in}, c_{K+,in}, c_{Cl-,in}, \Delta V$

Only permeating ions not pumped $\Delta V \approx V_i^N = -\frac{kT}{q_i} \ln \frac{c_{i,o}}{c_{i,i}}$

Steady state = resting potential $\Delta V = V^0$

Book notation $\mathcal{G}_i = g_i q_i$

11.1-2 Ion pumping -> sodium anomaly in all animal cells

$$V_i^N = -\frac{kT}{q_i} \ln \frac{c_{i,o}}{c_{i,i}}, \quad \mathcal{G}_i = g_i e$$

flux charge neutrality: $j_{Na+} + j_{K+} + j_{Cl-} = 0$ (1)

steady state: $\frac{3j_p}{\mathcal{G}_{Na+}} = -V_{Na+}^N + \Delta V$ (2)

$$-\frac{2j_p}{\mathcal{G}_{Na+}} = -V_{K+}^N + \Delta V$$
 (3)

$$0 = V_{Cl-}^N - \Delta V$$
 (4)

(2)+(3) $2\mathcal{G}_{Na+}(\Delta V - V_{Na+}^N) = 3\mathcal{G}_{K+}(\Delta V - V_{K+}^N)$

$$\Delta V = \frac{2\mathcal{G}_{Na+}V_{Na+}^N + 3\mathcal{G}_{K+}V_{K+}^N}{2\mathcal{G}_{Na+} + 3\mathcal{G}_{K+}}$$

Pumping =>

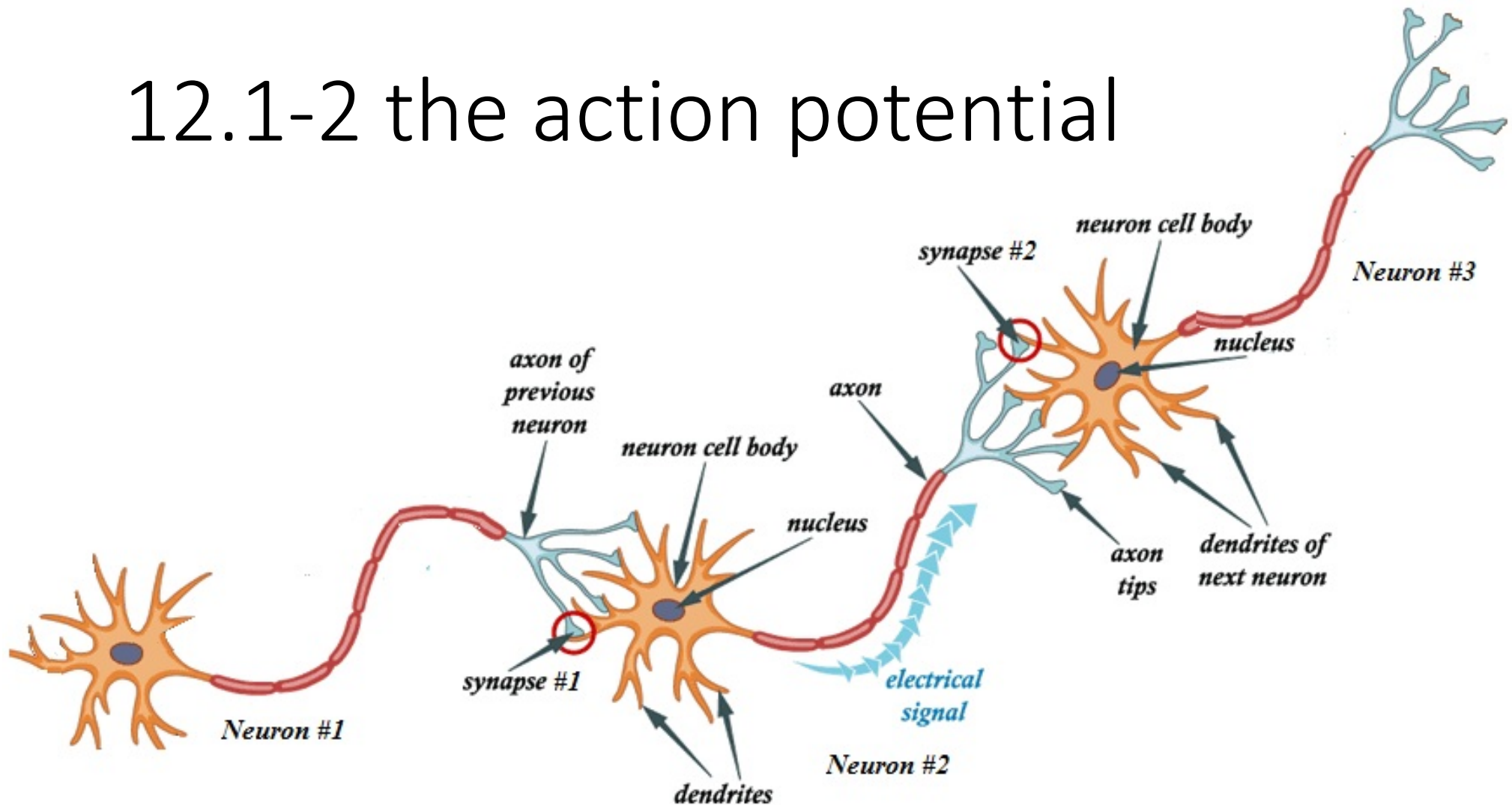
ion imbalance

| | $c_{i,o}$ [mM] | $c_{i,in}$ [mM] | V_i^N [mV] | g_i/g_{K+} |
|-----------------|----------------|-----------------|--------------|----------------------|
| K ⁺ | 20 | 400 | -75 | 1 |
| Na ⁺ | 440 | 50 | 54 | $\frac{1}{25}$ (* 8) |
| Cl ⁻ | 560 | 52 | -59 | $\frac{1}{2}$ |

=> $\Delta V = -72$ mV, but eq. (4): $\Delta V = -59$ mV

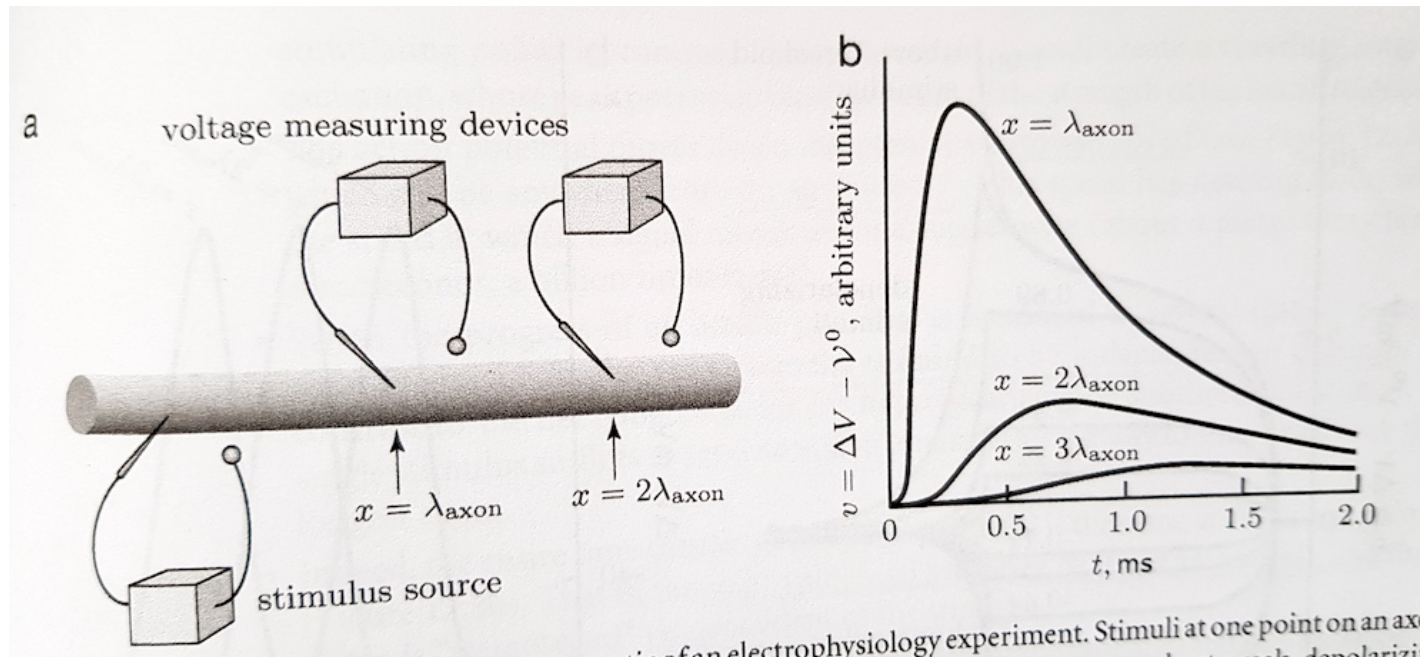
(1) can be used to correct (4), but (1) is not complete because other ions present ++

12.1-2 the action potential



- How can a leaky cable carry a sharp signal over long distances?
- Nonlinearity in cell membrane's conductance => excitable medium => regenerates signal

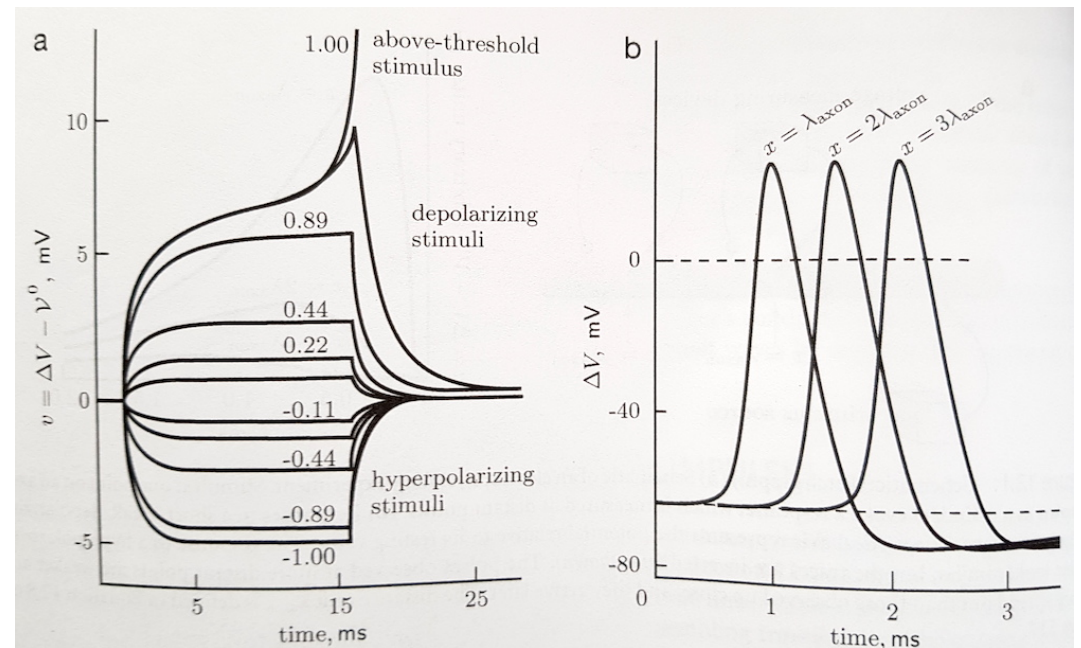
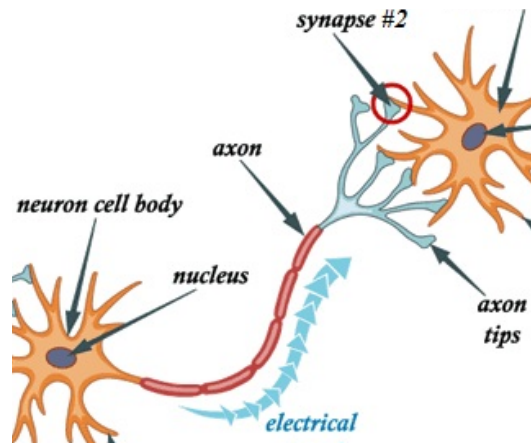
Nerve impulses???



- We are looking for propagating waves of $\Delta V - V^0$
- We only have a dissipative equation:

$$\dot{j}_i = j_{p,i} - g_i \Delta\mu_i$$

Observations of action potential in axon



- When stimulated beyond a threshold the axon changes polarization for a short while
- This pulse travels along the axon at constant speed (0.1-120 m/s)
- Peak potential and pulse shape are independent of distance
- afterhyperpolarization
- harder to stimulate during a refractory period
- The peak and the shape is independent of the exact triggering pulse

Squid giant axon

Numerical example:

| | $c_{i,o}$ [mM] | $c_{i,in}$ [mM] | V_i^N [mV] | g_i/g_{K^+} |
|-----------------|----------------|-----------------|--------------|----------------------|
| K ⁺ | 20 | 400 | -75 | 1 |
| Na ⁺ | 440 | 50 | 54 | $\frac{1}{25}$ (* 8) |
| Cl ⁻ | 560 | 52 | -59 | $\frac{1}{2}$ |