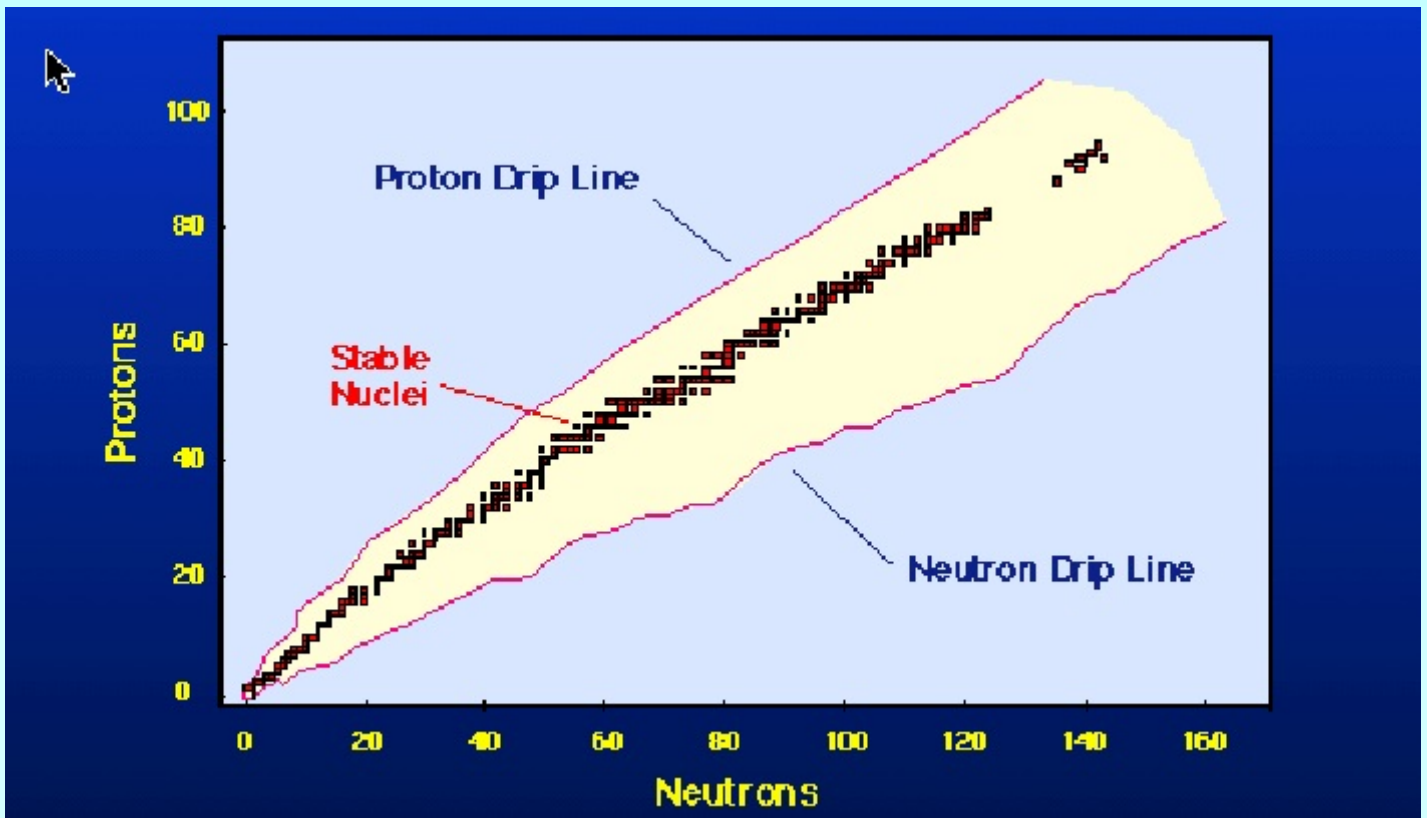




The valley of stability

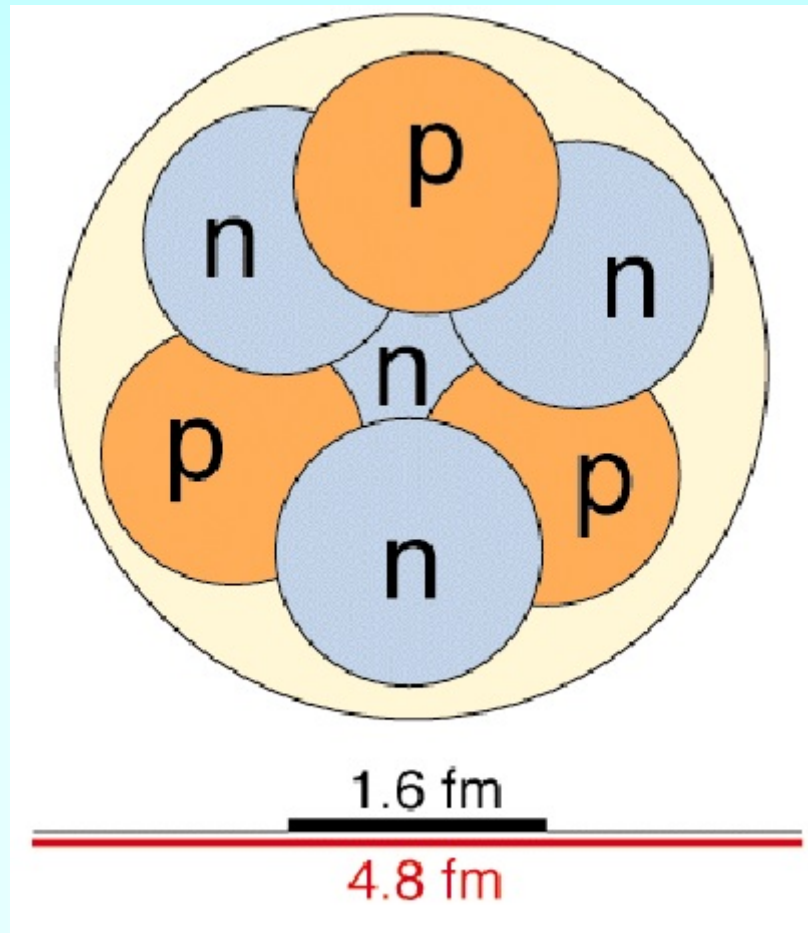


The stable nuclides are found in the indicated region.

It is important to notice that the N/Z ratio increases weakly towards higher mass numbers



How is the nucleus composed ?



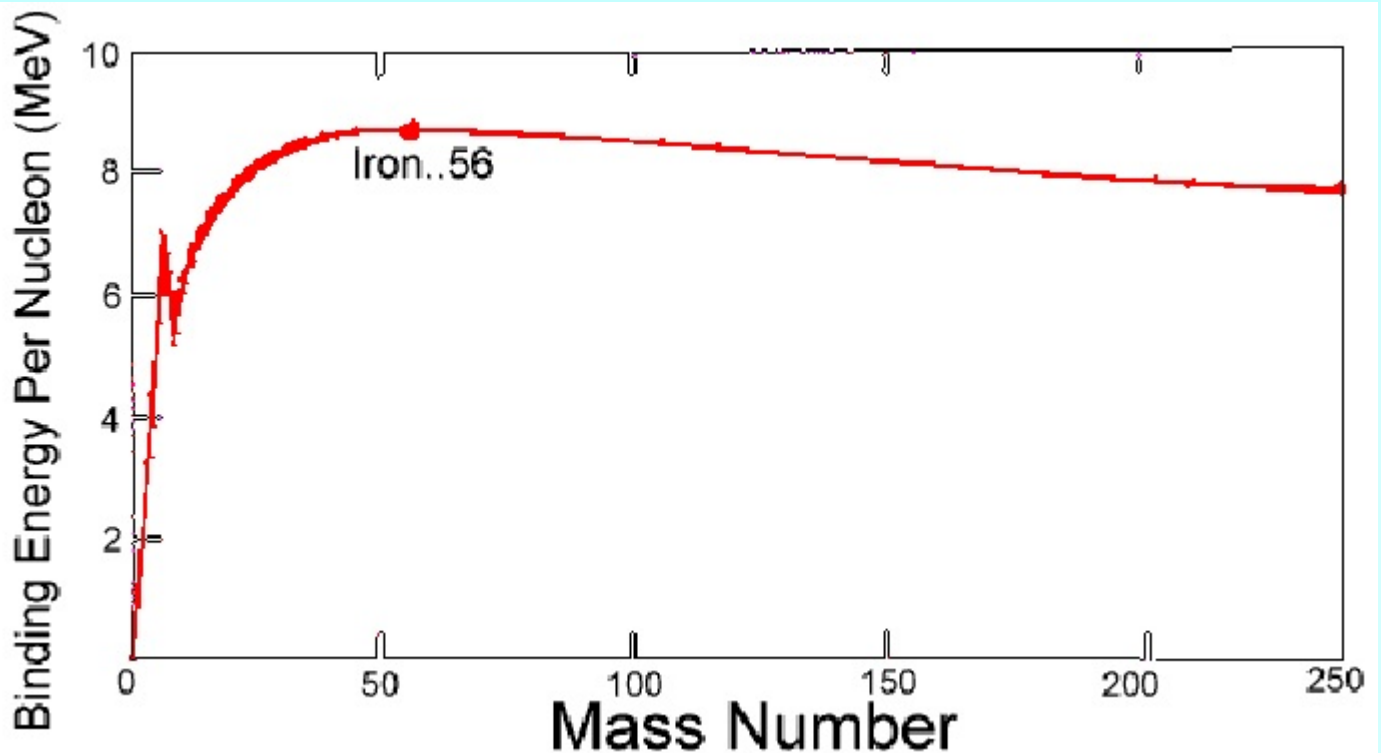
Size of a nucleon: approx. 1.6 fm

Size of the nucleus $\approx r_0 A^{1/3}$

1 fm (femtometer, fermi) = 10^{-15} m



Binding energy



The binding energy pr. nucleon as a function of the mass number

Increases rapidly at low masses,
with a maximum at ^{56}Fe

Sinks slowly towards high masses



From the chart of nuclides

- We have roughly
 - ▶ Stable nuclides: approx. 275 (black)
 - ▶ Nuclides occurring in nature: approx. 300
 - ▶ Total number of nuclides: approx. 2200
- The radioactive ones are:
 - ▶ β^- - neutronrich nuclei
 - ▶ β^+ - neutrondeficient nuclei
 - ▶ EC -neutrondeficient nuclei near stability
 - ▶ α - heavy nuclei
 - ▶ SF - very heavy nuclides
- Internal γ - everywhere
- Exotic forms:
 - ▶ Proton-emission
 - ▶ ^{14}C -emission



More observations from the chart of nuclides

● Stable nuclides

- ▶ Even Z, even N 163 60,8 %
- ▶ Even Z, odd N 55 20,3 %
- ▶ Odd Z, even N 49 18,9 %
- ▶ Odd Z, odd N 4* 1,5 %

*) All with $A \leq 14$

● Stable nuclides

- ▶ Only one on isobar chains with odd A
- ▶ Up to three on isobar chains with even A
- ▶ Max two stable isotopes with odd Z
- ▶ No stable isotopes for Z=43 and Z=61
- ▶ May be many stable isotopes for even Z (max for Sn, 10 stable)

● Other observations

- ▶ Many isomers for certain elements (e.g. almost every In isotope)
- ▶ Magic numbers
- ▶ Strong occurrence of α -activity right above ^{208}Pb
- ▶ No stable nuclides with A=5 or 8



Part of the chart of nuclides

<p>Bi 208 $3,68 \cdot 10^5$ a € γ 2615</p>	<p>Bi 209 100 σ 0,011 + 0,023</p>	<p>Bi 210 $3,0 \cdot 10^6$ a $5,013$ d α 4,946; 4,908... β^- 1,2 γ 266; 305... α 4,649; 4,686 σ 0,054 γ (305; 266)</p>	
<p>Pb 207 22,1 σ 0,70</p>	<p>Pb 208 52,4 σ 0,00049</p>	<p>Pb 209 3,253 h β^- 0,6 no γ</p>	
<p>Tl 206 3,7 m 4,20 m γ 686; 453; 216; 266; 1021... β^- 1,5... γ (803...)</p>		<p>Tl 207 1,33 s 4,77 m γ 1000; 351 β^- 1,4... γ (898...)</p>	
<p>Tl 208 3,053 m β^- 1,8; 2,4... γ 2615; 583; 511; 860; 277...</p>			

^{208}Pb - double shell nuclide



<p>Po 206 8,8 d</p> <p>ε; α 5,2233 γ 1032; 511; 286; 807... e⁻; g</p>	<p>Po 207</p> <table border="1"> <tr> <td>2,8 s</td> <td>5,84 h</td> </tr> </table> <p>ε; β⁺... α 5,116 γ 992; 743; 912...; g</p> <p>ly 815; 268; 301</p>	2,8 s	5,84 h	<p>Po 208 2,898 a</p> <p>α 5,1152... ε γ (292; 571...) g</p>
2,8 s	5,84 h			
<p>Bi 205 15,31 d</p> <p>ε β⁺... γ 1764; 703; 988...</p>	<p>Bi 206 6,24 d</p> <p>ε β⁺... γ 803; 881; 516; 1719; 537...</p>	<p>Bi 207 31,55 a</p> <p>ε β⁺... γ 570; 1064; 1770...</p>		
<p>Pb 204</p> <table border="1"> <tr> <td>67,2 m</td> <td>1,4</td> </tr> </table> <p>ly 899; 912; 375... σ 0,68</p>	67,2 m	1,4	<p>Pb 205 1,5 · 10⁷ a</p> <p>ε no γ σ ~ 5</p>	<p>Pb 206 24,1</p> <p>σ 0,030</p>
67,2 m	1,4			