

FYS3520 - Problem set 8

Spring term 2017

Corrections: Problem 2: $E_n = 170$ corrected to 1700 keV

Problem 1 – in class

- Figure 1 shows the electron spectra measured on β -particles from decay of ^{210}Bi . What does this spectra tell about the nature of β -decay?
- Comment on the three decays in Figure 2. What can you say about the half-life of the three different decays?
- Show in a (semi-)classical calculation that the angular momentum transfer l (in units of \hbar) to a electron in β -decay is much smaller than 1. Assume a typical Q -value of 3 MeV and $R=6$ fm for the nuclear radius. What do we learn about possible and forbidden β -decays?
- Frank presents some cool things about his research

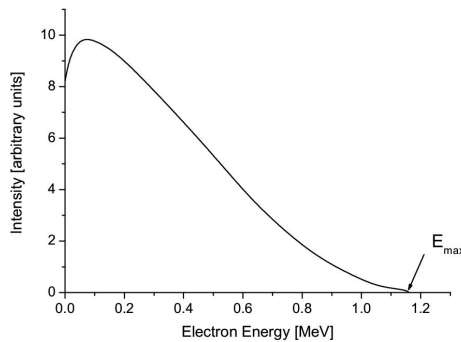


Figure 1: Electron spectra from β -decay of ^{210}Bi .

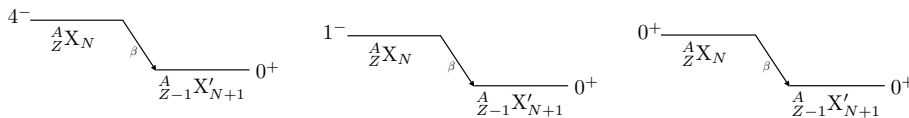


Figure 2: β -decays

Problem 2 β -delayed neutron emission

After β -decay of a excited state it is not only possible to emit γ -rays. Depending on the excitation energy of the daughter nucleus it can also emit particles. The β -delayed neutron emission is essential for the control of nuclear reactors. However, in this example we analyze the β -delayed neutron emission of ^{17}O , which leads to the gs. of ^{16}O . Here ^{17}O has before been populated by β^- -decay of ^{17}N . The decay scheme is given in Figure 3

As given in the figure three neutrons are emitted with energies 383, 1171 and 1700 keV, which populate the ground state of ^{16}O .

- a) Calculate the Q -value for the β -decay of ^{17}N .
- b) Derive a formula that allows you to calculate the excitation energy of ^{17}O as a function of the kinetic energy of the neutron and excitation energy of ^{16}O . Calculate the excitation energy of ^{17}O .
- Hint: You get the correct formula if you take into account energy *and* momentum conservation. You may regard only the ^{17}O - ^{16}O system and neglect ^{17}N .
- c) Why can the β -delayed neutron emission not populate the first excited state of ^{16}O at 6.05 MeV?

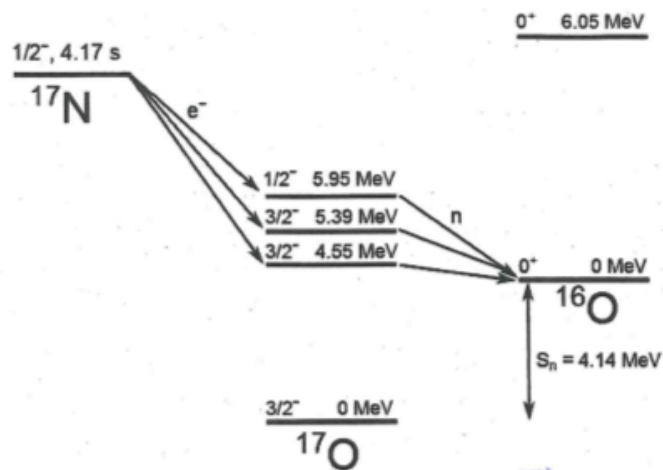


Figure 3: β -delayed neutron emission from ^{17}O