Exercises KJ 5220

November 23, 2009

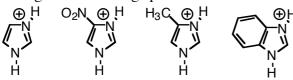
Left overs from last week

1

Benzimidazole has 2 ionization constants (pKa 12.9 and 5.3). To which acid-base equilibrium do these correspond.

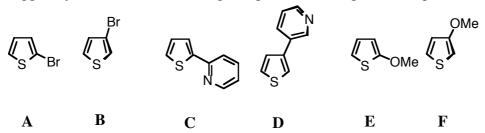
2.

Arrange the following species in order of increasing acidity, explain



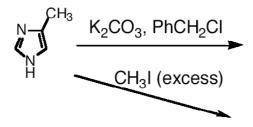
3.

Suggest synthesis of the following compounds, starting from thiophene



4

Which products will you expect from the following reactions



5.

Design syntheses of all isomers of bromothiazole from commercially available starting materials

6.

Suggest at least 3 different ways for this transformation

7

Explain / complete the following reactions

$$\begin{array}{c}
1) \text{ n-BuLi} \\
2) \text{ Bu}_3 \text{SnCl} \\
 & A
\end{array}$$

Br
$$\frac{1) \text{ n-BuLi}}{-70 \text{ °C - r.t.}}$$
 $\frac{2) \text{ H}^+ / \text{H}_2\text{O}}{\text{OH}}$

8.

Suggest syntheses of all isomers of trimethylstannylthiazole. Available heterocyclic starting materials are thiazole and 2,4-dibromothiazole.