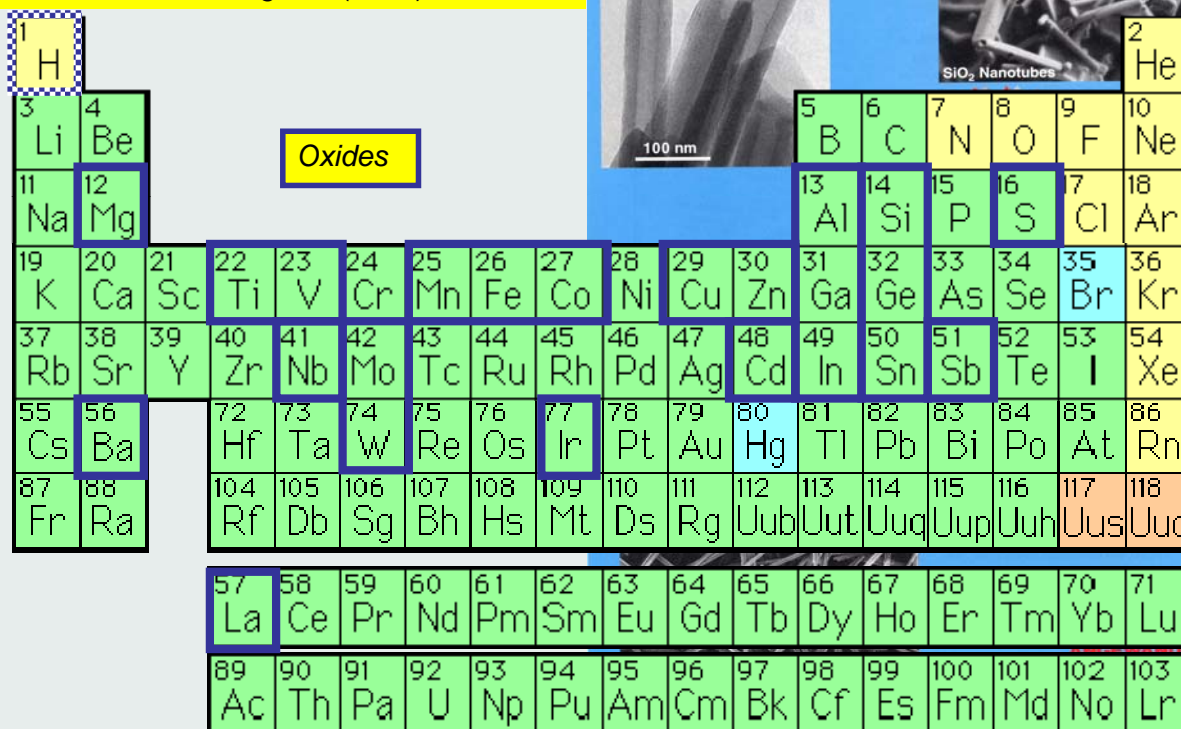


Oxidic Nanorods

G. R. Patzke, F. Krumeich, R. Nesper, *Angew. Chem. Int. Ed. Engl.* 14 (2002) 2447

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Rods, Fibers, Filaments

Table 1. Synthetic routes for oxidic nanorods (published since 1995).

Oxide	Synthetic route	Ref.	Oxide	Synthetic route	Ref.
BaCrO ₄	Fusion of reverse micelles and microemulsion droplets	[122]	MnO ₂	Template method with alumina membranes	[127]
BaSO ₄	Precipitation from aqueous solution in presence of polymers	[123]	MnO ₂	Hydrothermal synthesis	[142]
BaWO ₄	Reversed micelle templating method	[124]	MoO ₃	Template-directed reaction of molybdic acid and subsequent leaching process	[143]
CdO	Evaporation of metal oxide powders at high temperatures	[125]	MoO ₃	Templating against carbon nanotubes	[65]
CdWO ₄	Hydrothermal treatment of CdCl ₂ and Na ₂ WO ₄	[126]	MoO ₃	Templating against carbon nanotubes	[65]
Co ₃ O ₄	Template method with alumina membranes	[127]	PbTiO ₃	Sol-gel electrophoresis, deposition in polycarbonate membrane	[144]
CuO	Room temperature reaction of CuCl ₂ · 2H ₂ O and NaOH with PEG 400	[128]	RuO ₂	Templating against carbon nanotubes	[65]
Fe ₂ O ₃	Thin-film processing method	[129]	Sb ₂ O ₃	Microemulsion method for the system AOT-water-toluene ^[a]	[145]
Fe ₃ O ₄	Sonication of aqueous iron(II) acetate in the presence of β-cyclodextrin	[130]	Sb ₂ O ₃	Microemulsion method for the system AOT-water-toluene ^[a]	[145]
Ga ₂ O ₃	DC arc discharge of GaN powders in Ar/O ₂ mixture	[131]	Sb ₂ O ₃	Templating against carbon nanotubes	[65]
Ga ₂ O ₃	Gas reaction method starting from Ga and O ₂ at 780 °C	[132]	SnO ₂	Annealing of powders generated from inverse microemulsions	[146]
Ga ₂ O ₃	Physical evaporation at 300 °C from a bulk gallium target	[133]	SnO ₂	Evaporation of metal oxide powders at high temperatures	[125]
Ga ₂ O ₃	DC arc discharge (GaN, graphite, nickel powder)	[134]	SiO ₂	Helical mesostructured tubules from Vortex-Assisted Surfactant Templates	[147]
Ga ₂ O ₃	Electric arc discharge of GaN powders mixed with Ni and Co	[135]	TiO ₂	Sol-gel template method employing alumina membranes	[148]
Ga ₂ O ₃	Heating of Ga with SiO ₂ powder and a Fe ₂ O ₃ catalyst	[136]	V ₂ O ₅	Vanadium pentoxide gels	[149]
GeO ₂	Carbon-nanotube confined reaction of metallic Ge	[137]	V ₂ O ₅	Templating against carbon nanotubes	[65]
In ₂ O ₃	Evaporation of metal oxide powders at high temperatures	[125]	WO ₃	Templating against carbon nanotubes	[65]
In ₂ O ₃	Growth from Au droplets	[128]	YBCO	Laser ablation of a high T _c superconductor YBa ₂ Cu ₃ O ₇	[150]
IrO ₂	Templating against carbon nanotubes	[65]	ZnO	Gas reaction employing Zn and H ₂ O	[151]
K ₂ Ti ₆ O ₁₃	Calcination of KF and TiO ₂	[139]	ZnO	Evaporation of metal oxide powders at high temperatures	[125]
MgO	Vapor-solid growth process with in situ generated Mg vapor	[114]	ZnO	Catalyzed epitaxial growth	[152]
MgO	Heating of MgCl ₂ at 750 °C in mixture gas (Ar/H ₂) ■■■ not O ₂ ?	[140]	ZnO	Self-organization of nanoparticles	[153]
Mg(OH) ₂	Solvothermal treatment of Mg, H ₂ O, and ethylenediamine	[141]			

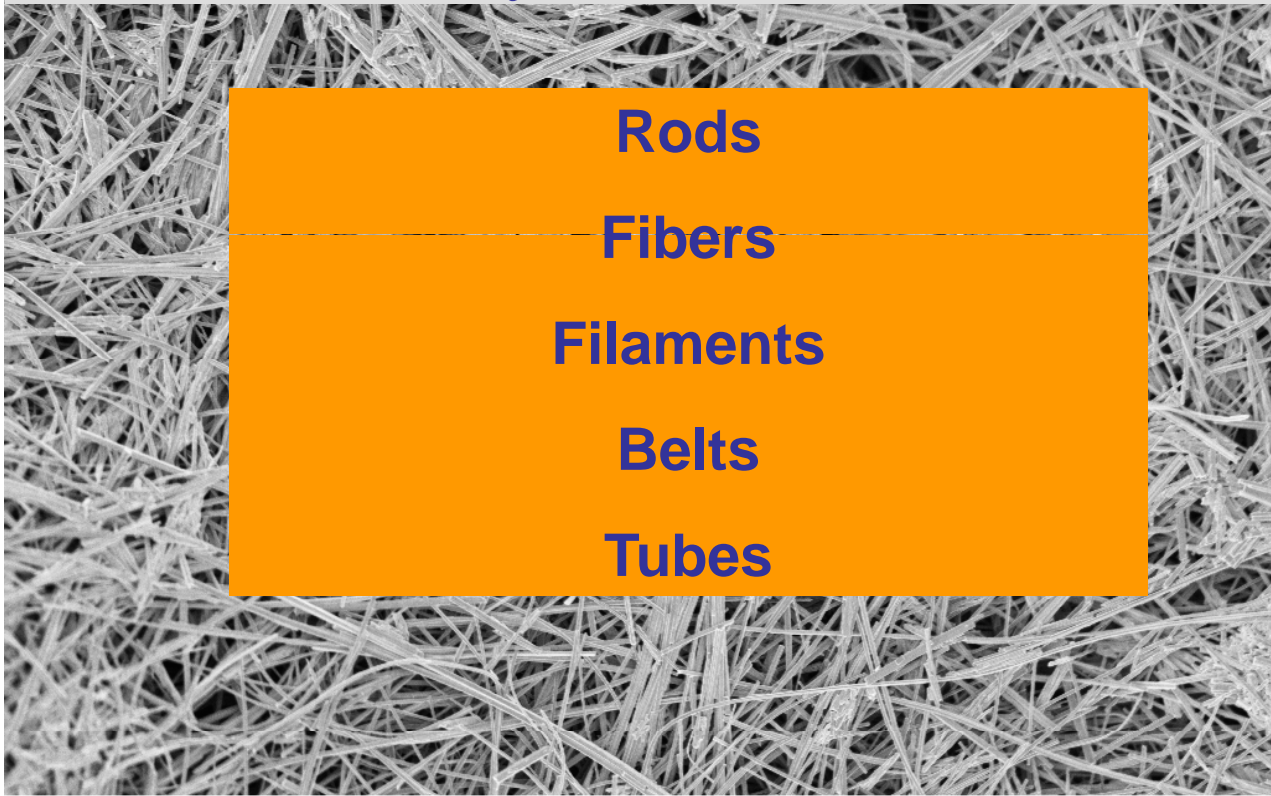
G. Patzke, F. Krumeich, R. Nesper, *Angew. Chem.* 2002, 114, 2554

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Anisotropic Particles



1µm

EHT = 1.00 kV
 WD = 5 mm

Signal A = InLens
 File Name = VFas-28_04.tif

Date :21 Nov 2002

Growth Conditions

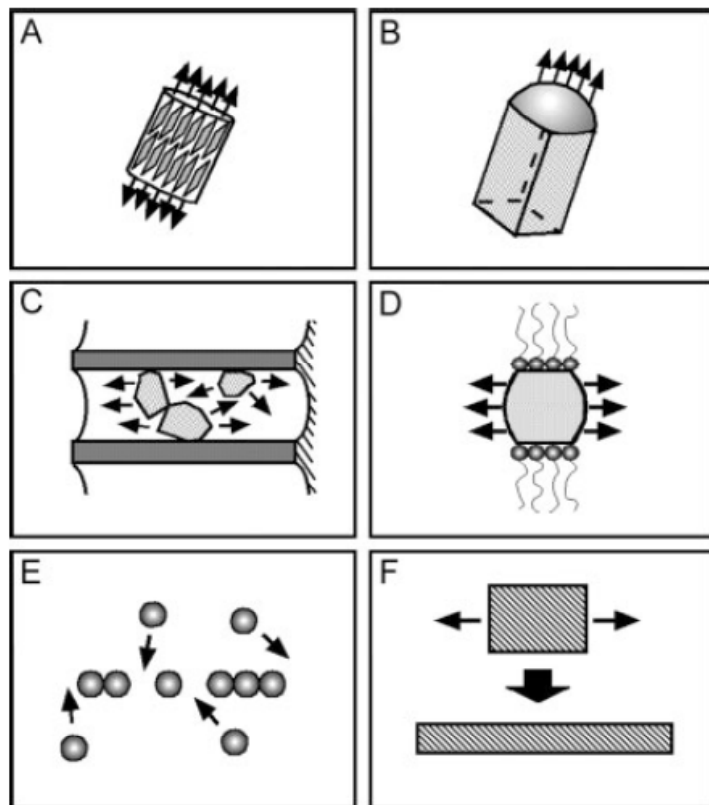
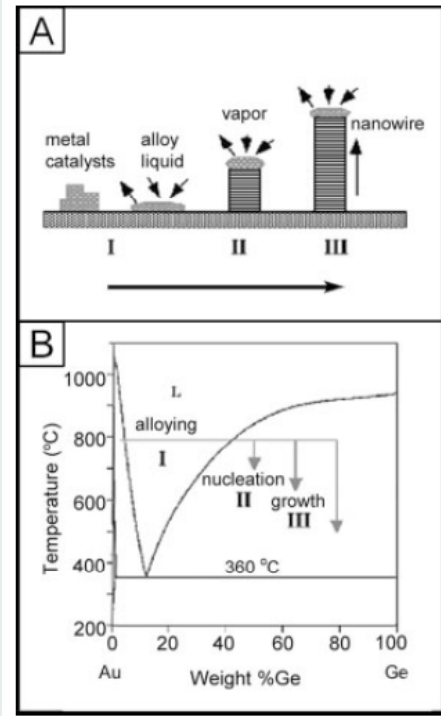


Fig. 1. Schematic illustrations of six different strategies that have been demonstrated for achieving 1D growth: a) dictation by the anisotropic crystallographic structure of a solid; B) confinement by a liquid droplet as in the vapor-liquid-solid process; C) direction through the use of a template; D) kinetic control provided by a capping reagent; E) self-assembly of 0D nanostructures; and F) size reduction of a 1D microstructure.

Phase Segregation



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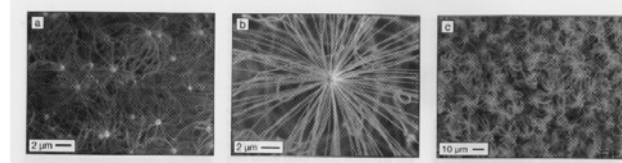
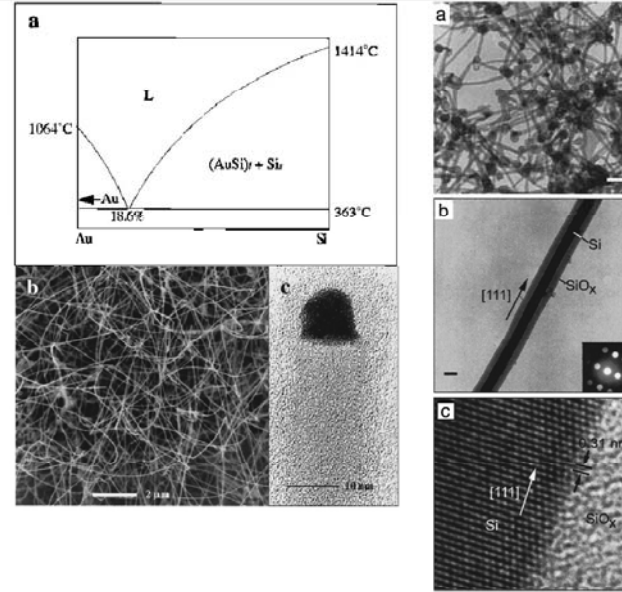


Figure 11. Scanning-electron-microscopy images of (a) a surface covered by flower-like silica nanostructures generated by heating SiC + Co at 1500°C for 30 min under an Ar-CO₂ atmosphere, (b) a typical three-dimensional feature radiating from a central spherical core-shell particle, and (c) a uniform 'nanoflower' film obtained from heating SiC + Co at 1500°C in a pure CO atmosphere.



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Surface Structures as Templates

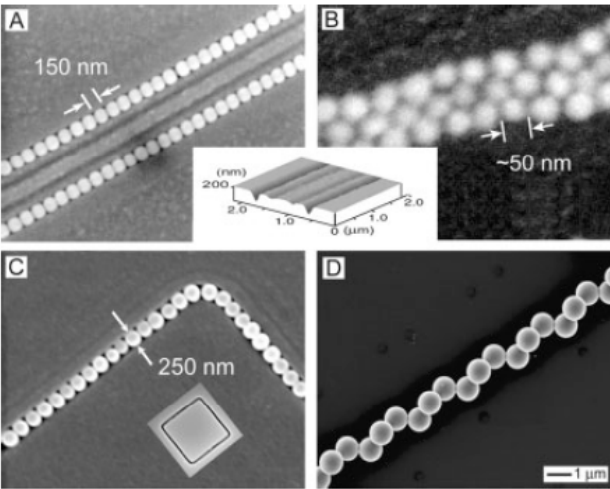


Fig. 21. A,B) Structures that were assembled from 150 nm polystyrene beads (A), and 50 nm Au colloids (B), by templating against 120 nm-wide channels patterned in a thin photoresist film (see the inset) [161a]. C) An L-shaped chain of Au@SiO₂ spheres assembled against a template (see the inset) patterned in a thin photoresist film [161c]. D) A spiral chain of polystyrene beads that were assembled by templating against a V-groove etched in the surface of a Si(100) wafer [161d].

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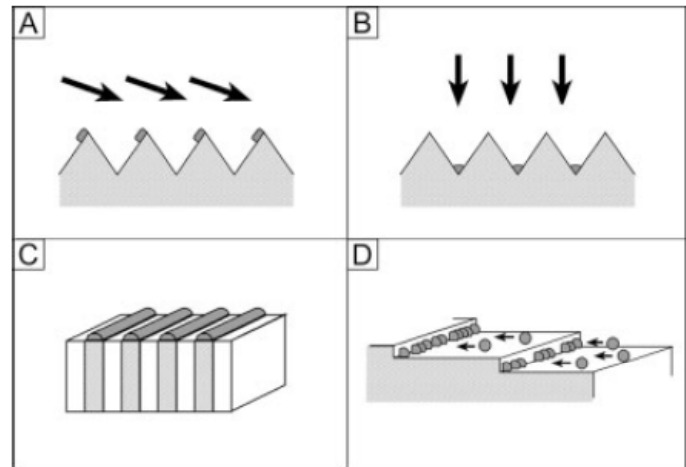
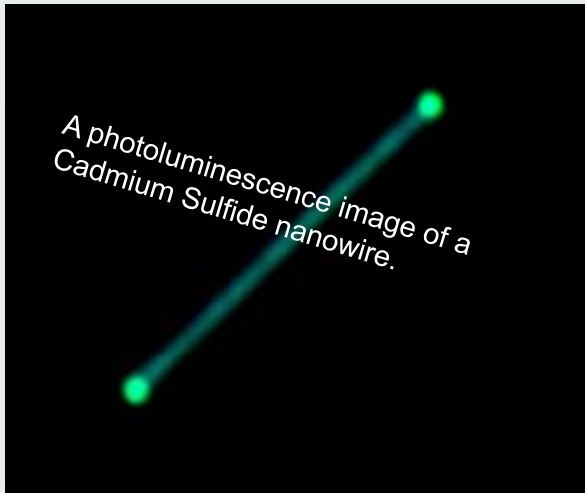


Fig. 6. Schematic illustrations of procedures that generated 1D nanostructures by A) shadow evaporation [58]; B) reconstruction at the bottom of V-grooves [60]; C) cleaved-edge overgrowth on the cross-section of a multilayer film [64]; and D) templating against step edges on the surface of a solid substrate [68].

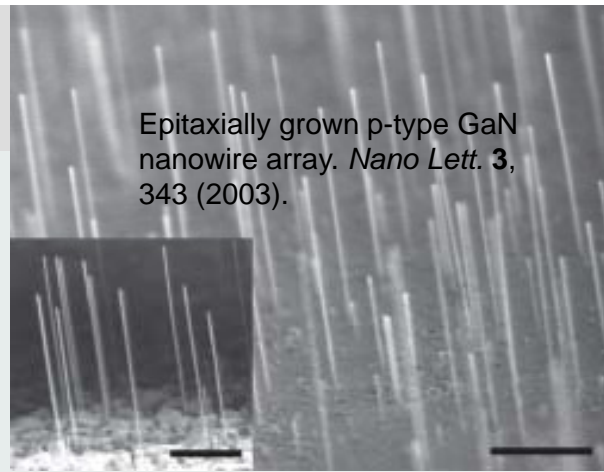
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Semiconductors



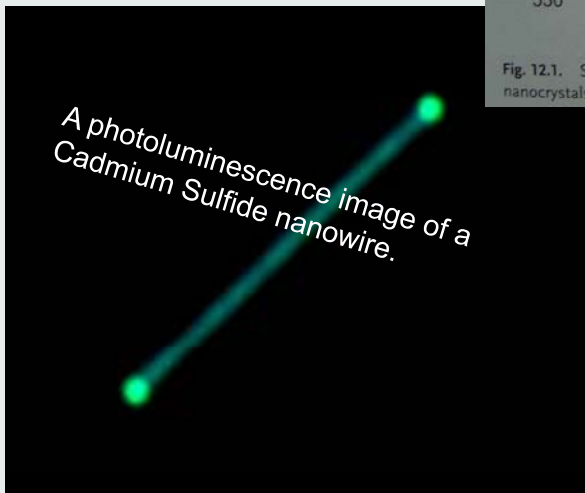
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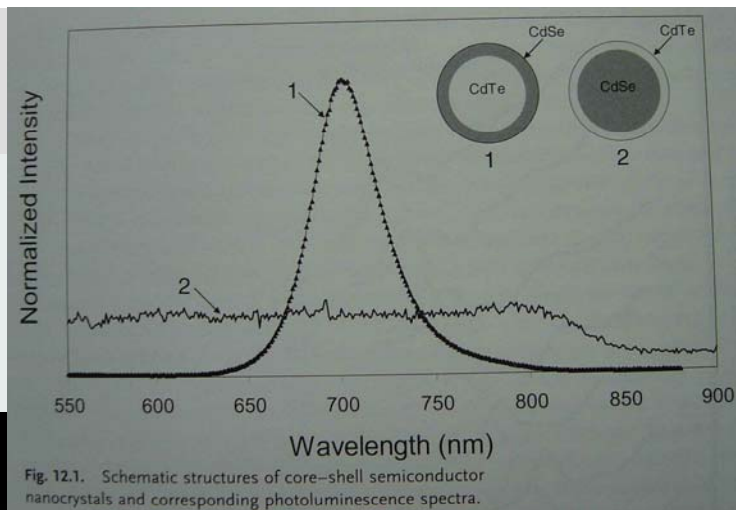
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Photoluminescence



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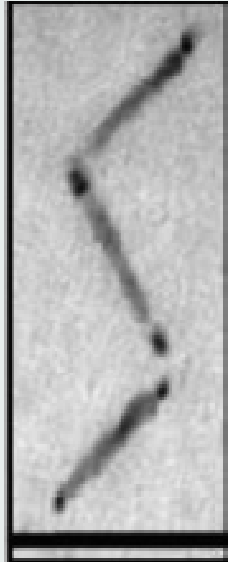
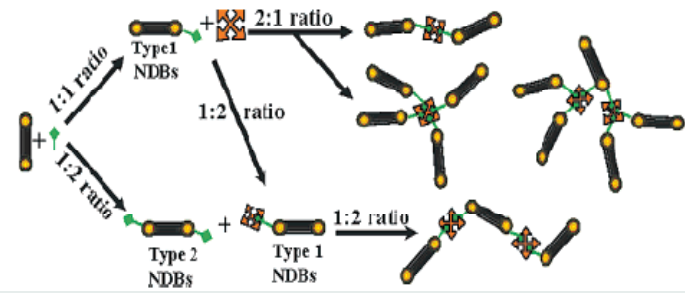
gold-tipped CdSe nanorods

Directed self-assembly of gold-tipped CdSe nanorods

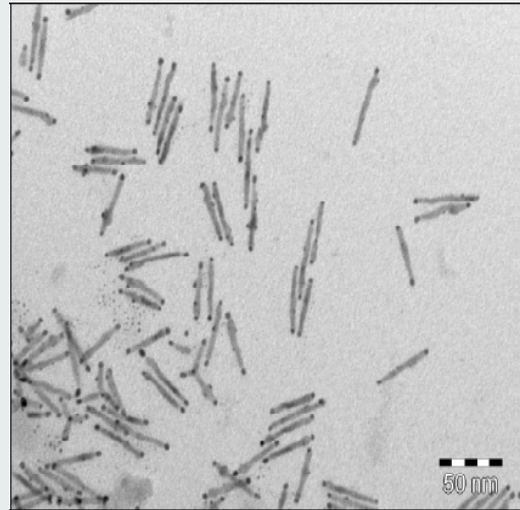
Asaf Salant, Ella Amitay-Sadovsky, Uri Banin

J. AM. CHEM. SOC. 2006, 128, 10006-10007

Scheme 1. Reacting Biotin Disulfide (Green) with NDBs and Avidin (Orange) Using Different Mole Ratios, Resulting in Dimers, Flowers, and Trimers



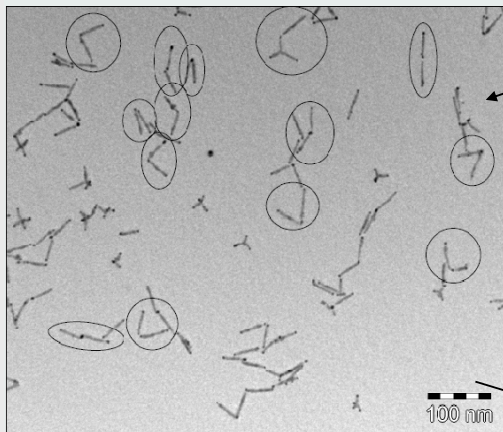
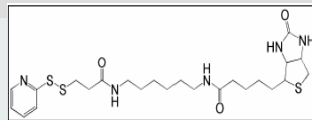
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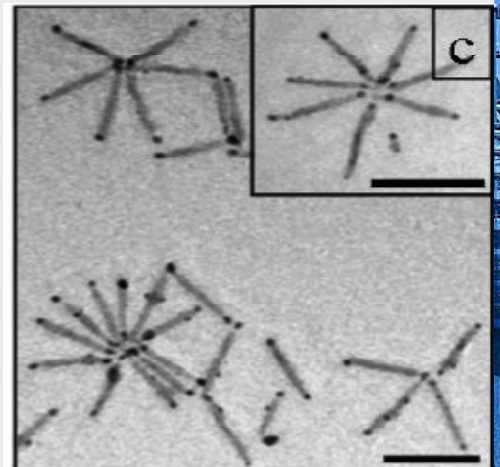
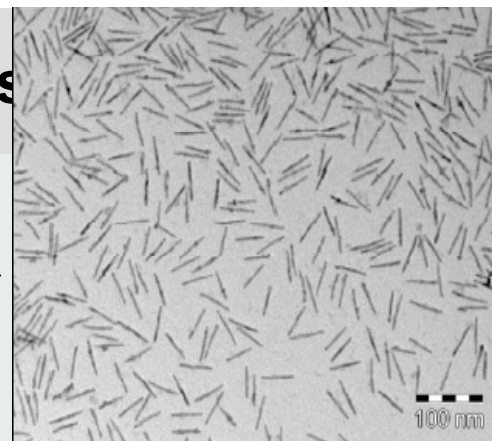
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Gold-tipped CdSe nanorods



CdSe nanorods were reacted as the dimer scheme, show mostly of side-to-side interactions.

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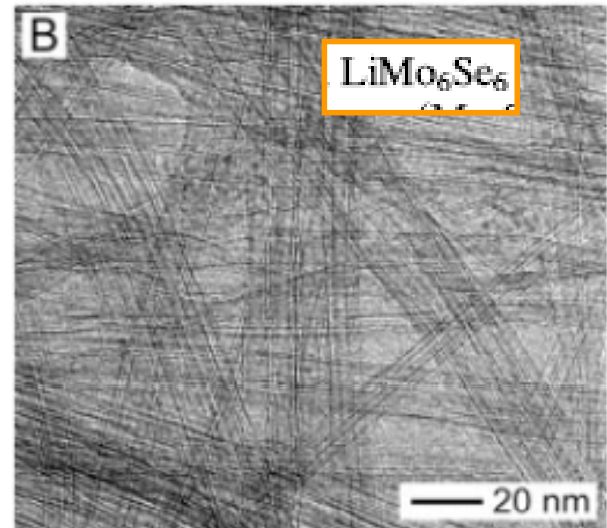
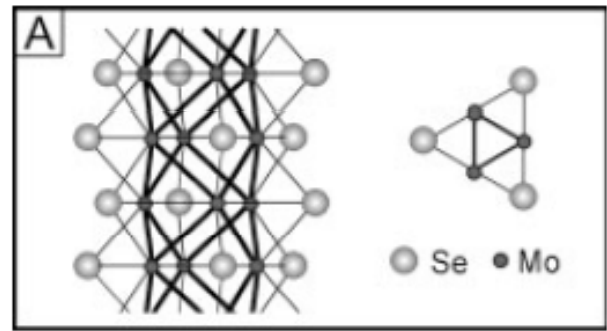
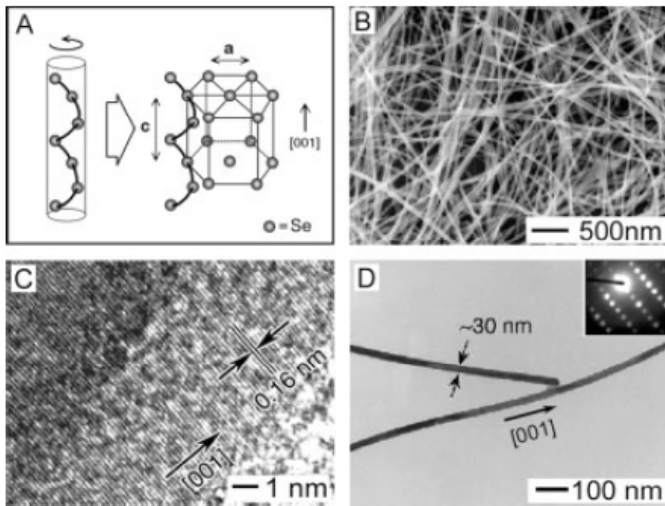
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Nanofibers by Crystal Structure

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Selenium



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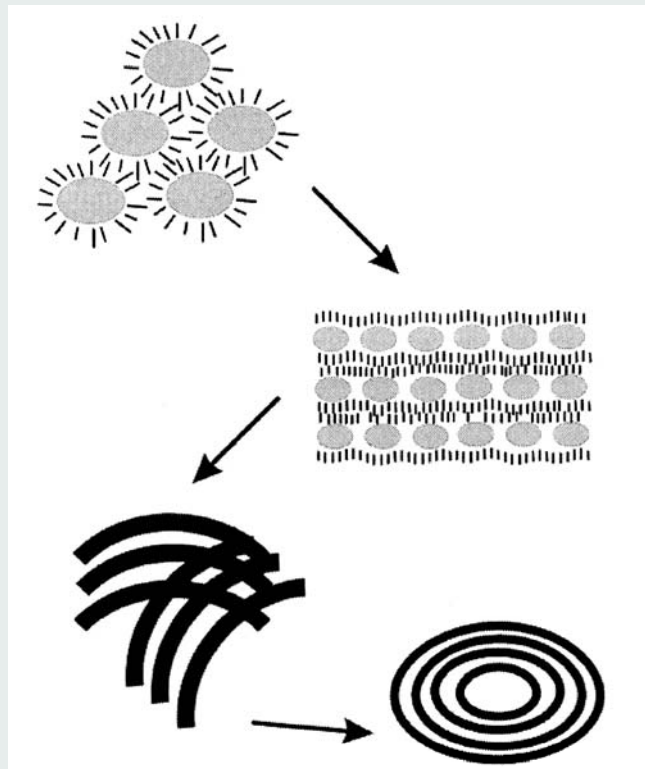
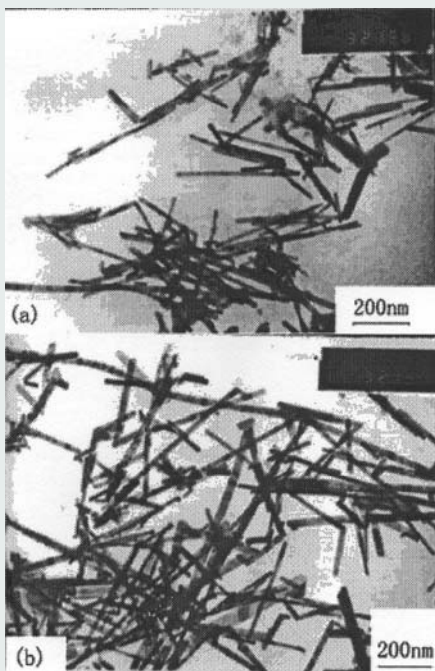
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SnO₂ NRs – Growth Mechanism

Synthesis and Characterization of Rutile SnO₂ Nanorods**

By Yingkai Liu,* Changlin Zheng, Wenzhong Wang, Chunrong Yin, and Guanghou Wang

Adv. Mater. **2001**, *13*, No. 24, December 17



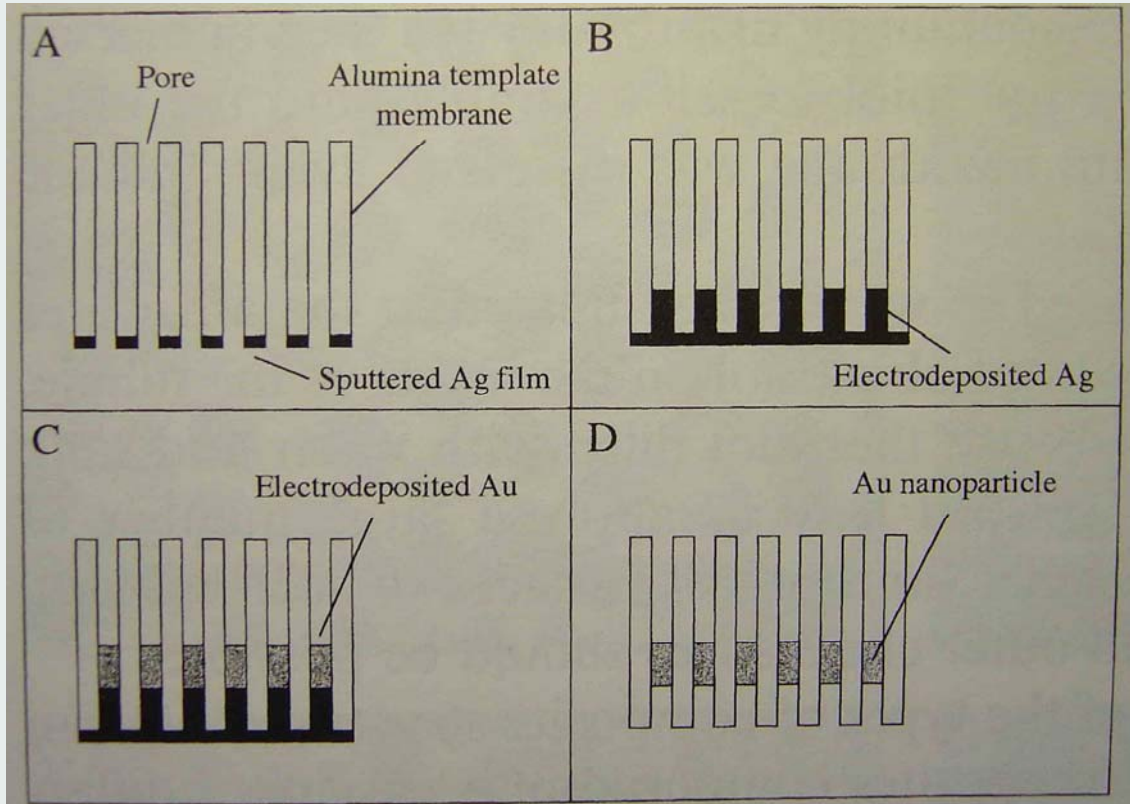
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ETH
Department of Chemistry, Institute of Physical Chemistry
St. Alban-Anlage 11, CH-8057 Zurich

Alumina Membranes as Outer Templates

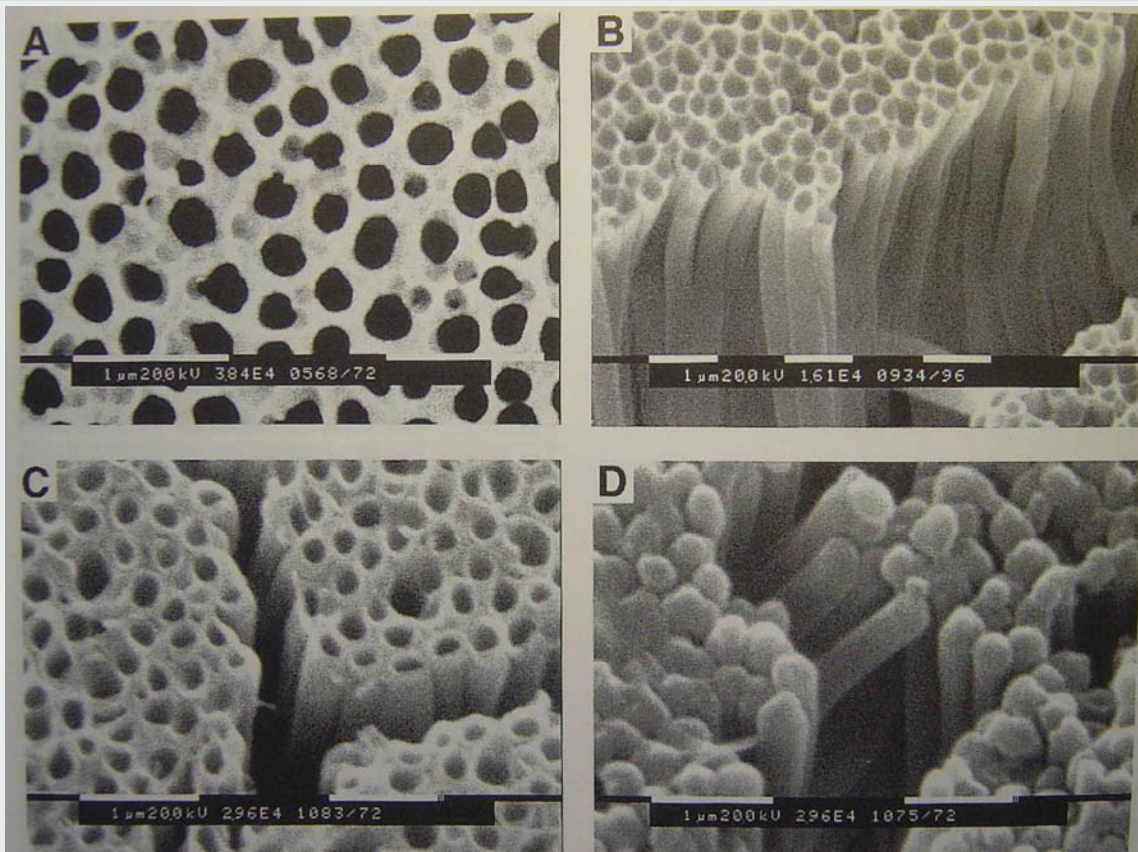


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Alumina Membranes as Outer Templates



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Hetero-Structured Nano Wires

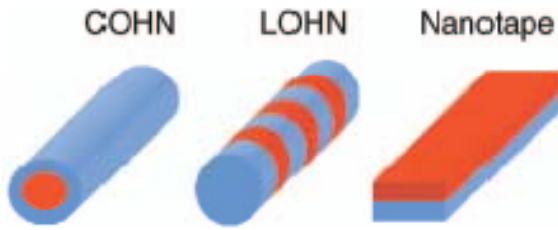


Figure 3. Three different types of heterostructured nanowires: a coaxial heterostructured nanowire (COHN), a longitudinal heterostructured nanowire (LOHN), and a nanotape.

Detailed models of phonon heat conduction in semiconducting nanowires that consider modified dispersion relations and all-important scattering processes predict a large decrease (90%) in the lattice thermal conductivity of wires tens of nanometers in diameter.

Peidong Yang
MRS BULLETIN
VOLUME 30, 2005

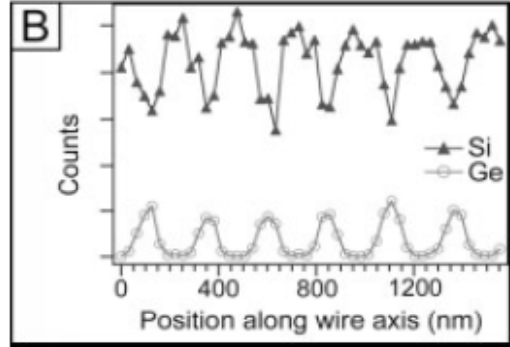
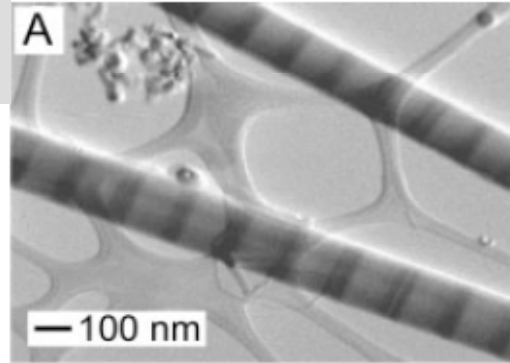


Fig. 23. A) A TEM image of two typical Si/SiGe superlattice nanowires. B) Compositional profiles showing the spatial modulation in Si and Ge contents along the longitudinal axis of a nanowire [186].