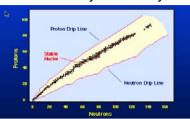
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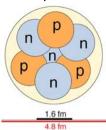
## The valley of stability



The stable nuclides are found in the indicated region. It is important to notice that the N/Z ration increases weakly towards higher mass numbers

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## How is the nucleus composed?



Size of a nucleon: approx. 1.6 fm Size of the nucleus  $\approx r_0 A^{1/3}$ 

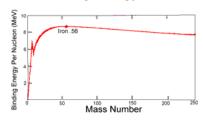
1 fm (femtometer, fermi) = 10<sup>-15</sup> m

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## Binding energy



The binding energy pr. nucleon as a function of the mass number

Increases rapidly at low masses, with a maximum at 56Fe Sinks slowly towards high masses

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### From the chart of nuclides

- We have roughly
- Stable nuclides: approx.275 (black)
- Nuclides occuring in nature: approx. 300
  Total number of nuclides: approx. 2200
- The radioactive ones are:
- β neutronrich nuclei
- β<sup>+</sup> neutrondeficient nuclei
- ► EC -neutrondeficient nuclei near stability
- α heavy nuclei
- ► SF very heavy nuclides
- Internal γ everywhere
- Exotic forms:
- ▶ ¹⁴C-emission

► Proton-emission

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#### More observations from the chart of nuclides

#### Stable nuclides

- ► Even Z, even N 163 60,8 %
- ► Even Z, odd N 55 20,3 %
- ► Odd Z, even N 49 18,9 % 4\* 1,5 %
- Odd Z, odde N
  \*) All with A≤ 14

### Stable nuclides

- · Only one on isobar chains with odd A
- Up to three on isobar chains with even A
- Max two stable isotopes wiith odd Z
- ► No stable isotopes for Z=43 and Z=61
- May be many stable isotopes for even Z (maxfor Sn, 10 stable)

#### Other observations

- · Many isomers for cerntain elements (e.g.
- almost every In isotope)
- Magic numbers
- Strong occurrence of α-activity right above
- ► No stable nuclides with A=5 or 8

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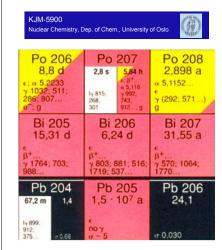
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### Part of the chart of nuclides



<sup>208</sup>Pb - double shell nuclide

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