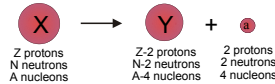




α -disintegration (rep)

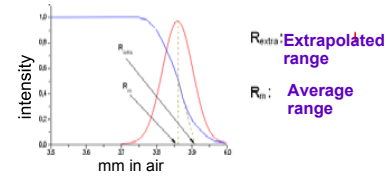


- An α -particle is a ^4He -ion, consisting of 2 protons and 2 neutrons.
- Two-particle process
- The decaying nucleus gets 4 atomic mass units lighter.
 - In other words: $A_{\text{stab}} = A_{\text{parent}} - 4$.
- The decaying atom is transformed into a new element containing Z-2 protons.
- The α -particles are monoenergetic and have typical energy between 2 and 7 MeV.
- An α -particle is easily slowed down and eventually becomes a neutral He-atom (after picking up two electrons).



α -disintegration

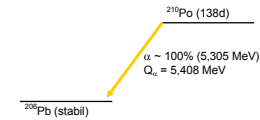
- Range in air: 3.5 cm
- Range in water, 50-80 μm
- LET: approx. 100 keV/ μm
- Sharp range



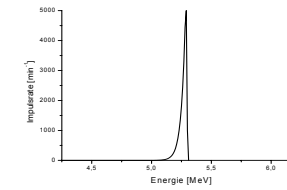
Lage variation in half-life from some nano-seconds up to 10^{15} years
 Little variation in energy (2,5 - 9 MeV)



α -disintegration



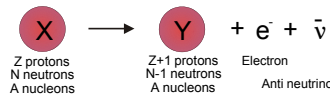
α -disintegration gives line spectrum



$$\ln \lambda = -a \cdot E_{\alpha}^{-1/2} + b$$



β^- -disintegration (rep.)

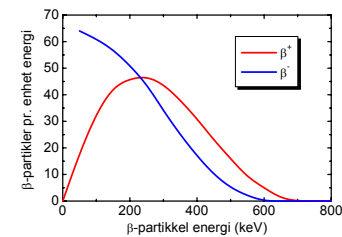
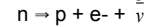


- β^- -particles are highly energetic electrons with negativ charge
- β^- particles are not mono-energetic, like the α -particles, but may have energies between 0 and $E_{\beta, \text{max}}$. Their average energy is 1/3 of the energy liberated in the disintegration.
- β^- -particle emission is always accompanied by the emission of an antineutrino.
 - The antineutrino is without mass and charge
 - Antineutrinos practically do not interact with matter and will "disappear" without leaving a trace (with very few exceptions).



Conservation laws.

Correct disintegration:



Explains continuous β -spektrum



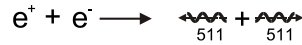
β^+ -disintegration



- β^+ -particles are highly energetic positrons (positiv charge)
- Like the β^- -particles, β^+ -particles are not monoenergetic.
- Also for β^+ -disintegration an additional particle is emitted, the neutrino.
- β^+ -particles may therefore also have energies between 0 og $E_{\beta, \text{max}}$. Their average energy is 1/3 of the energy liberated in the disintegration.
- The neutrino is the antiparticle of the anti-neutrino and behaves correspondingly.



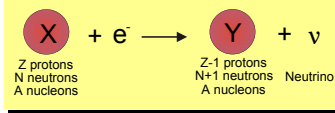
β^+ -disintegration



- A positron is an "anti-particle" and is not stable.
- After losing its kinetic energy it will react with an electron. The total restmass will be emitted as two electromagnetic quants.
- The quants have the energy 511 keV, one electron mass.
- The quants are emitted in opposite direction due to the conservation of momentum.



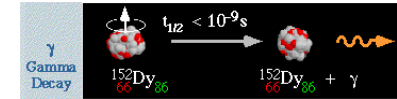
Electron capture



- Electron capture is considered to be a type of β^- -disintegration (weak process)
- Instead of sending out a positron, an electron is captured by the nucleus.
- One gets the same product as for β^- -disintegration, but there is no emission of electron/positron, just a (mono-energetic) neutrino.
- Characteristic X-rays are emitted when the electron structure is rearranged.

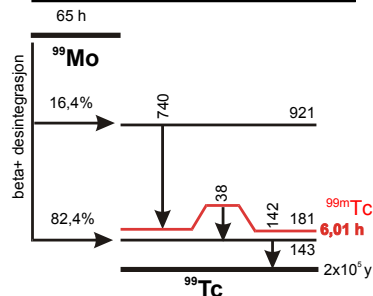


- γ -disintegration, transition between two levels in the same nucleus, usually rapid, $\sim 10^{-15}$ s.
- Transition between two levels without change in nucleon number or proton number
- Isomeric γ -disintegration, long half-life
 - White indications in the chart of nuclides
- Isomers arise due to changes of nuclear spin with many units (≥ 3)
- F.ex $^{99m}\text{Tc} \rightarrow ^{99}\text{Tc} + \gamma$ (142 keV) (6.0 h) (213 000 \AA)

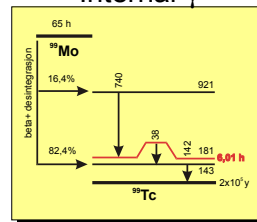


Nuclear Isomers

- Nuclear isomers are levels with extremely long half-life
 - Up to 10^{15} y (^{180m}Ta)
 - Normally 1s - some days
 - Lower limit, $\sim 1\mu\text{s}$ (individual)



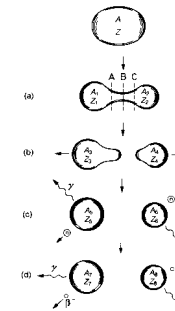
Internal γ



- Excited levels with long half-life are called *metastable*.
- Nuclei in these excited states are called metastable or isomers.
- This is marked by writing an "m" after the atomic number, in this case it is ^{99m}Tc .
- A metastable nucleus is exactly the same nucleus as in the ground state, but it has "extra" energy.
- Metastable nuclei need not emit γ , but may totally or partly disintegrate in other ways.



Spontaneous fission



Spontaneous fission is a mode of decay where the whole nucleus breaks up into two heavy fragments. (Flerov 1940)



Spontaneous fission -yields

