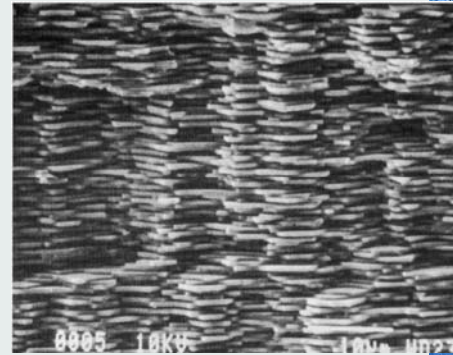
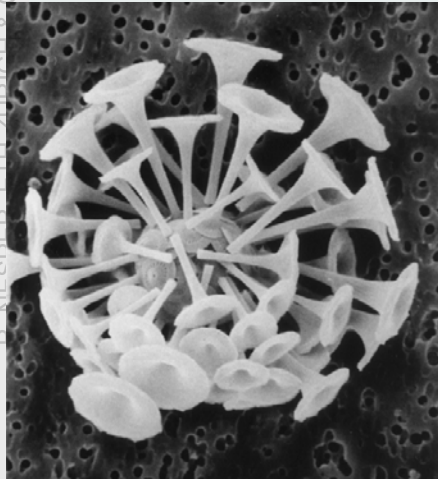
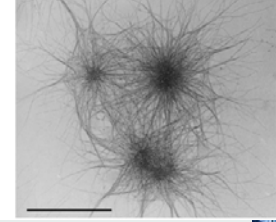
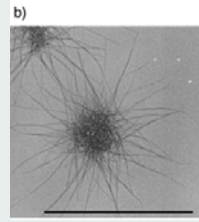
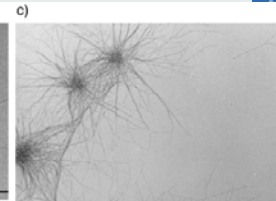
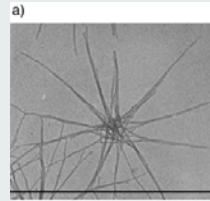
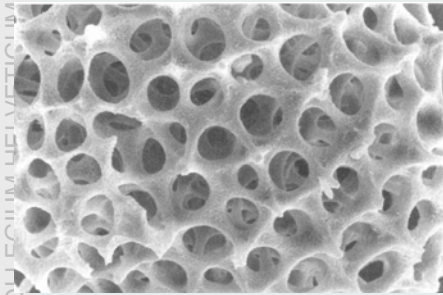
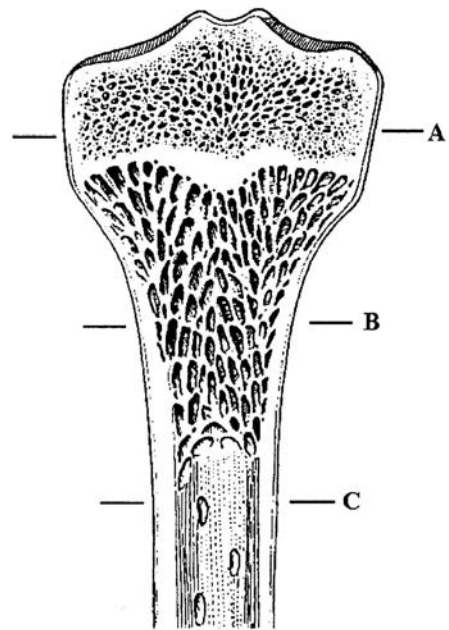


# Forms of Minerals in Biology



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- protection
- motion
- cutting and grinding
- buoyancy
- optical, magnetic and gravity sensing
- storage.



**Fig. 2.9** Internal structure of long bone with three different microstructures (A, B and C).

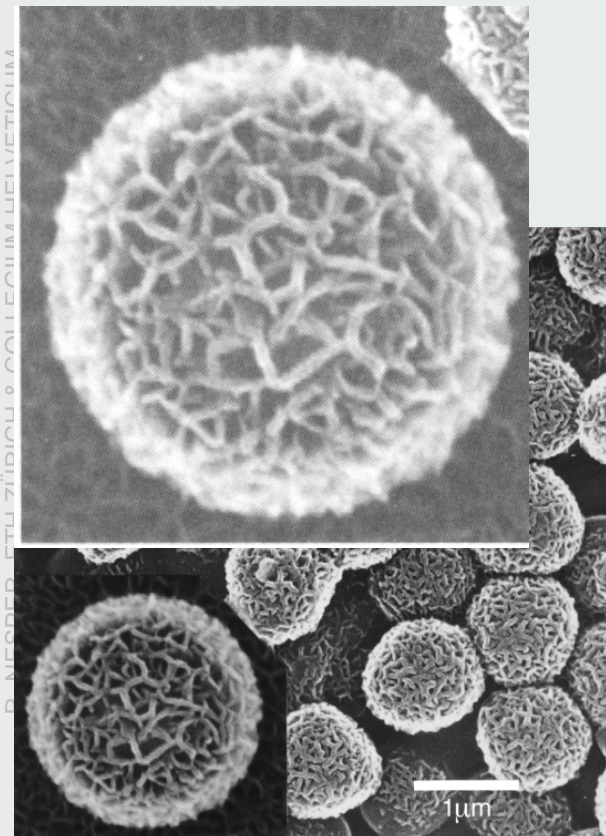
# Minerals in Biology

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**TABLE 11.8** Most Important Biominerals

Chemical Composition	Mineral Phase	Function and Examples
Calcium carbonate $\text{CaCO}_3$	Calcite Aragonite Vaterite Amorphous	Exoskeletons (e.g., egg shells, corals, mollusks, sponge spicules)
Calcium phosphates $\text{Ca}_{10}(\text{OH})_2(\text{PO}_4)_6$ $\text{Ca}_{10-x}(\text{HPO}_4)_x(\text{PO}_4)_{6-x}(\text{OH})_{2-x}$ $\text{Ca}_{10}\text{F}_2(\text{PO}_4)_6$ $\text{Ca}_2(\text{HPO}_4)_2 \cdot 2\text{H}_2\text{O}$  $\text{Ca}_2(\text{HPO}_4)_2$  $\text{Ca}_8(\text{HPO}_4)_2(\text{PO}_4)_4 \cdot \text{H}_2\text{O}$  $\text{Ca}_3(\text{PO}_4)_2$	Hydroxyapatite (HAP) Defect apatites Fluoroapatite Dicalcium phosphate dihydrate (DCPD)  Dicalcium phosphate (DCPA)  Octacalcium phosphate (OCP)  $\beta$ -Tricalcium phosphate (TCP)	Endoskeletons (bones and teeth)
Calcium oxalate $\text{Ca}_1\text{C}_2\text{O}_4 \cdot (1 \text{ or } 2)\text{H}_2\text{O}$	Whewellite Wheddelite	Calcium storage and passive deposits in plants calculi of excretory tracts
Metal sulfates $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ $\text{SrSO}_4$  $\text{BaSO}_4$	Gypsum Celestite  Baryte	Gravity sensors Exoskeletons (acantharia) Gravity sensors
Amorphous silica $\text{SiO}_n(\text{OH})_{4-2n}$	Amorphous (opal)	Defense in plants, diatom valves, sponge spicules, and radiolarian tests
Iron oxides $\text{Fe}_3\text{O}_4$  $\alpha, \gamma\text{-Fe}(\text{O})\text{OH}$ $5\text{Fe}_2\text{O}_3 \cdot 9\text{H}_2\text{O}$	Magnetite  Goethite, lepidocrocite Ferrihydrite	Chiton teeth, magnetic sensors Chiton teeth Chiton teeth, iron storage

## Forms of Minerals in Biology



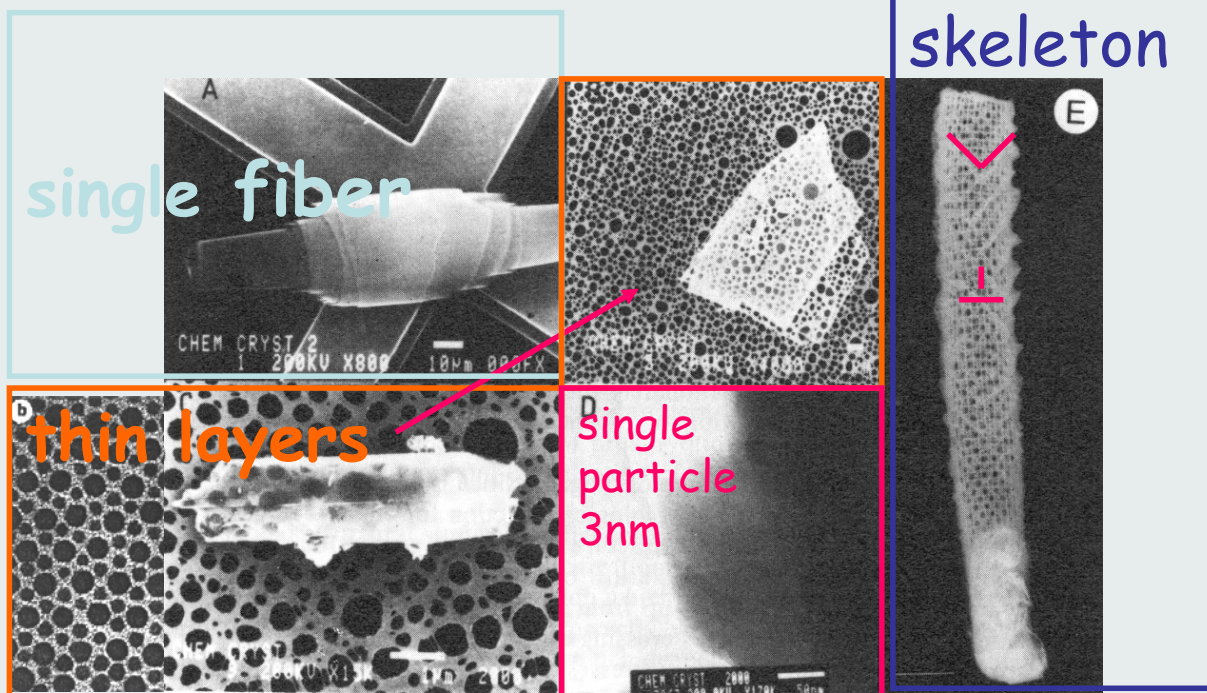
ETH  
Empire State University  
State University of New York  
State University of New York

# Reproducibility of Minerals in Biology

**Table 2.3** Chemical composition of calcium phosphate (hydroxyapatite) in human and shark enamel

Composition (wt%)	Human enamel	Shark enamel
Ca <sup>2+</sup>	37.55	37.26
Na <sup>+</sup>	0.75	0.76
Mg <sup>2+</sup>	0.27	0.32
PO <sub>4</sub> <sup>3-</sup>	17.68	17.91
CO <sub>3</sub> <sup>2-</sup>	3.6	1.1
F <sup>-</sup>	0.02	3.65

# Silica - deep sea sponge



# Selfassembly

## Spontaneous ordering of bimodal ensembles of nanoscopic gold clusters

C. J. Kiely\*, J. Fink†, M. Brust†, D. Bethell† & D. J. Schiffrin†

\* Materials Science and Engineering, Department of Engineering, The University of Liverpool, Liverpool L69 3BX, UK

† Department of Chemistry, The University of Liverpool, Liverpool L69 7ZD, UK

Nature (1998) 396, 444

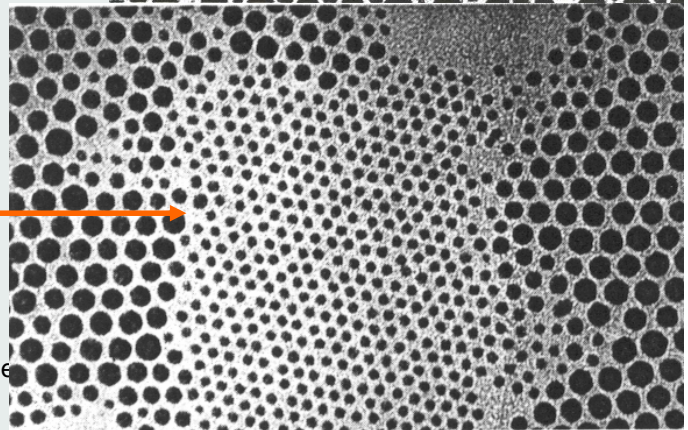
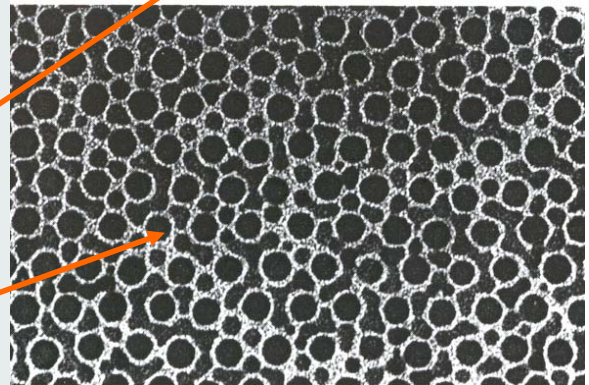
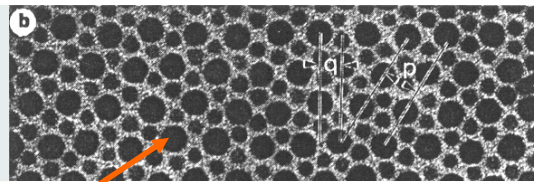
primary structure

after aging

phase separation

design - two cluster sizes

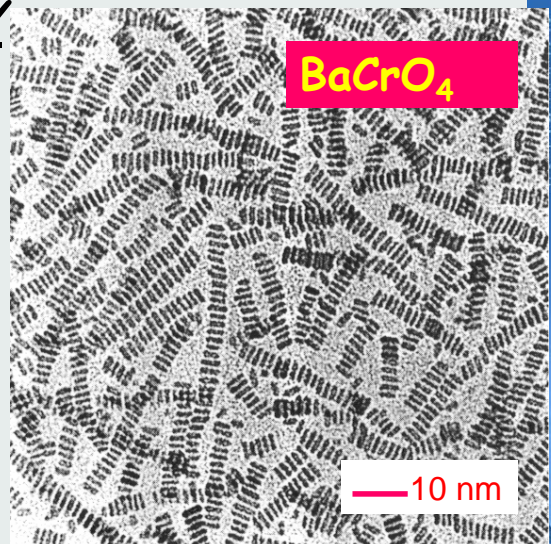
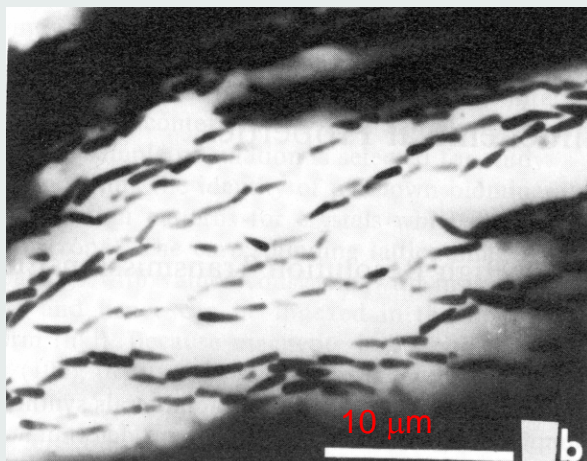
Nanochem



# Advanced Selfassembly

## Magnetosomes in algae

S. Mann, R.B. Frankel  
in S. Mann, J. Webb,  
R.J.P. Williams, Biomineralization,  
VCH 1989



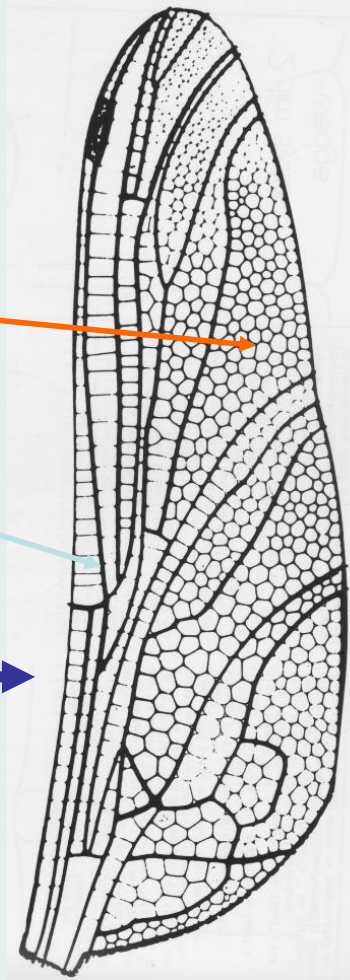
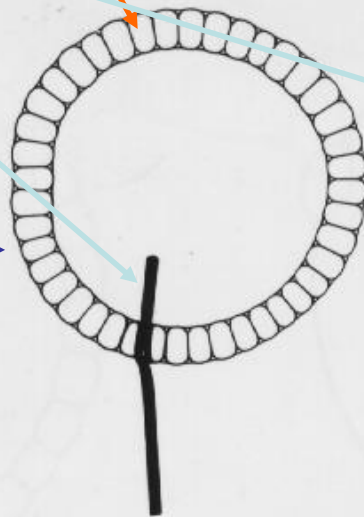
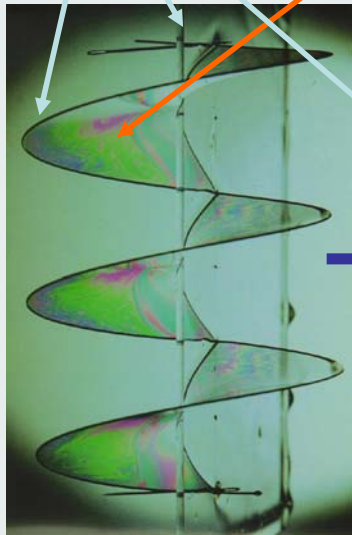
## Coupled synthesis and self-assembly of nanoparticles to give structures with controlled organization

Mei Li\*, Heimo Schnablegger† & Stephen Mann\*

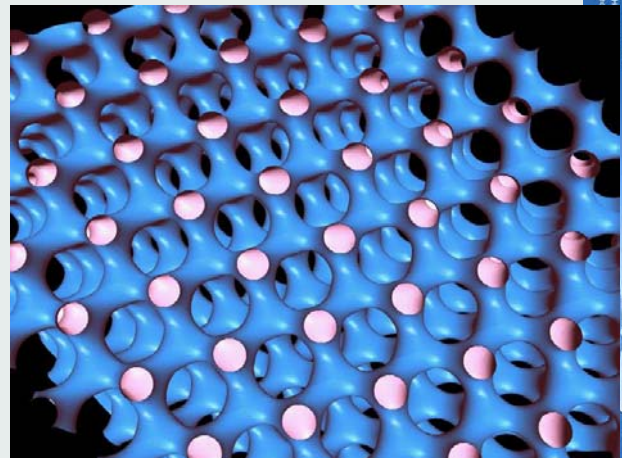
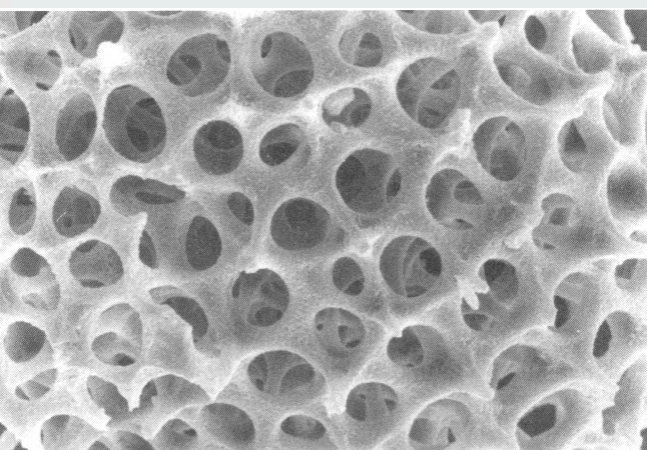
Nature (1999) 402, 393

# Bio"mineralization"

Design + Selforganisation

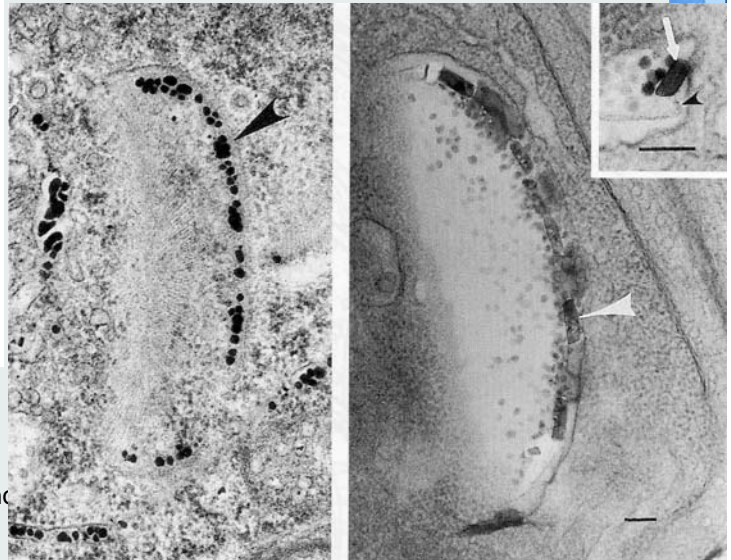
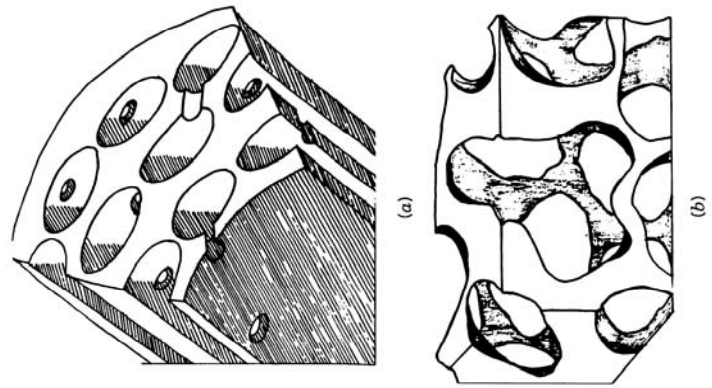
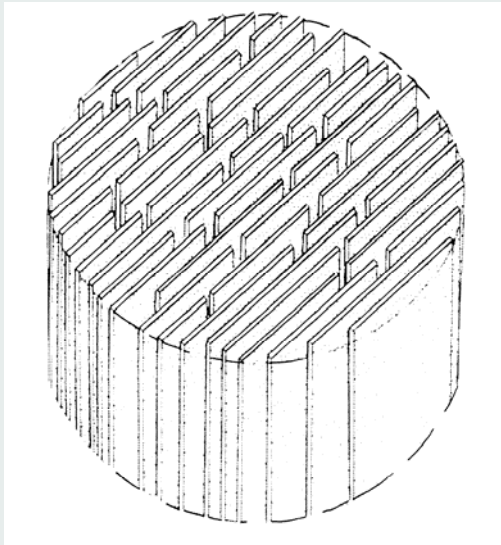


## More Complex Forms of Minerals in Biology



# Design of Complex Forms

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Nano

## How is it achieved ?

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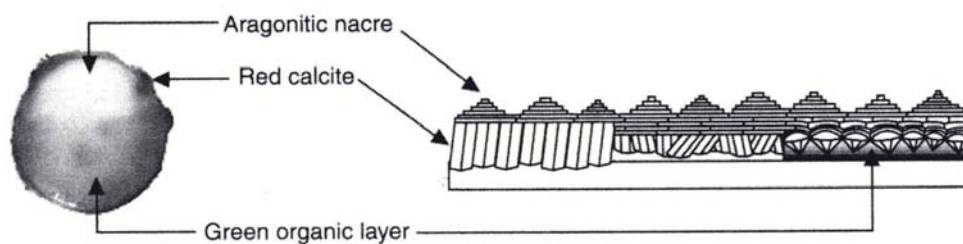
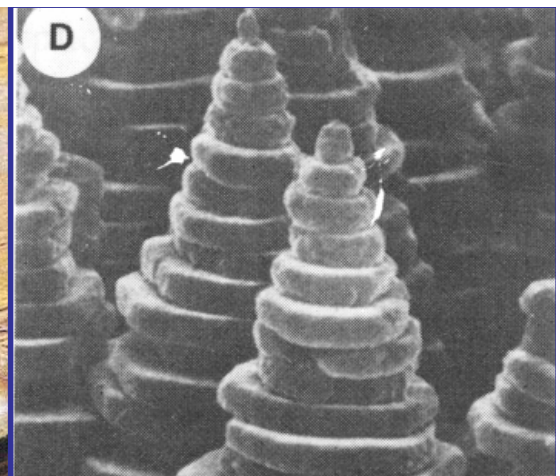
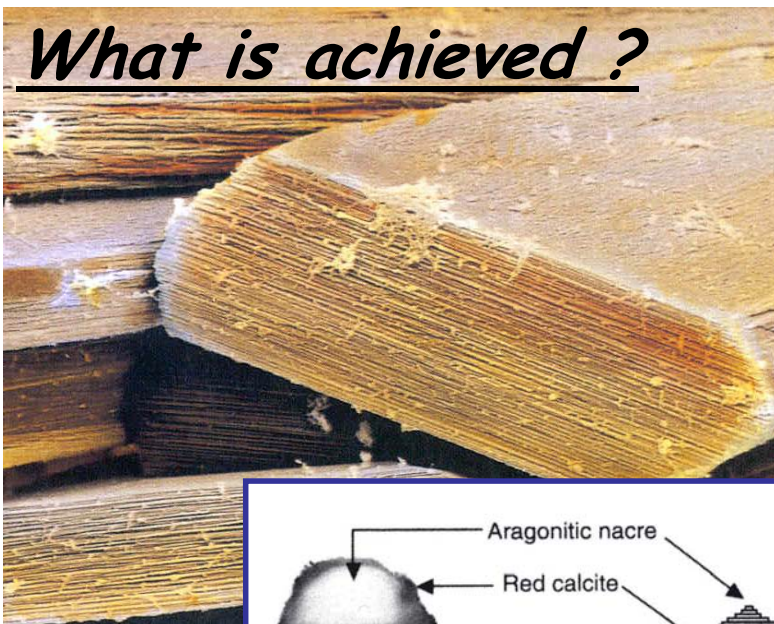
**Table 3.2** The main types of biomineralization processes

Process	Control mechanism	Concepts	Properties	Chapter
Precipitation (crystallization)	Chemical	Solubility Supersaturation Nucleation Growth	Solution composition Promotion Inhibition Phase transformation	4
Boundary-organized biomineralization	Spatial	Supramolecular preorganization	Physical boundary Diffusion-limited site Ion transport Size and shape Organization	5
Organic matrix-mediated biomineralization	Structural	Interfacial molecular recognition	Site-directed nucleation Oriented nucleation Supporting framework Mechanical design	6
Morphogenesis	Morphological	Vectorial regulation	Complex form Time-dependent form Patterning	7
Biomineral tectonics	Constructional	Multilevel processing	Higher-order assembly Hierarchical structures Integrative building modules Adaptive structures and functions	8

# What is achieved ?

- uniform particle sizes
- well-defined structures and compositions
- high levels of spatial organization
- complex morphologies
- controlled aggregation and texture
- preferential crystallographic orientation
- higher-order assembly into hierarchical structures.

# What is achieved ?



**Figure 15.3** A mature flat pearl. The *c*-axis of the nacre is perpendicular to the paper. The three different regions of the flat pearl are labeled red calcite, aragonitic nacre and green organic layer. The schematic diagram on the right shows the spatial organization of the regions.

# Toughness

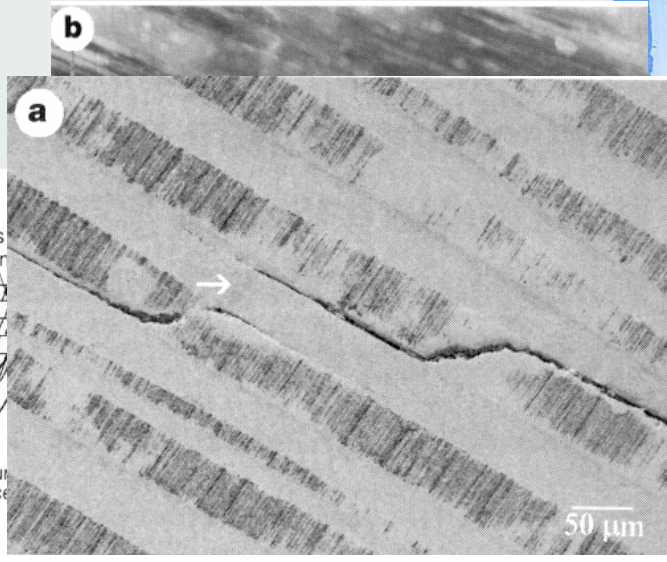
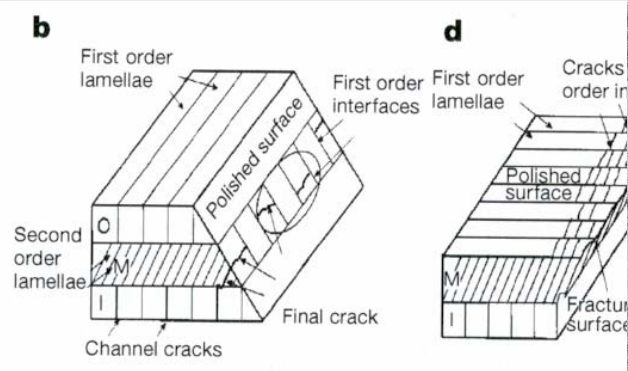
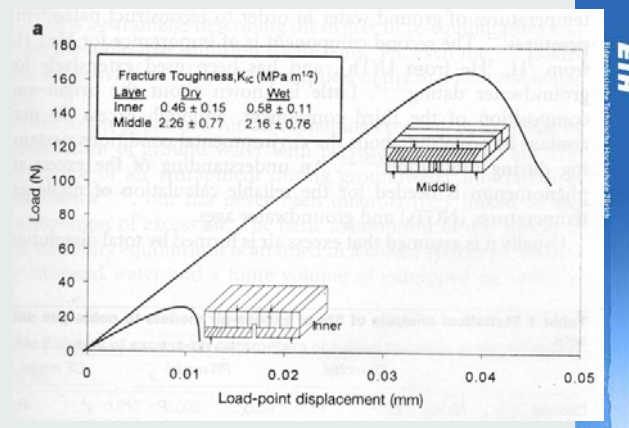
## Structural basis for the fracture toughness of the shell of the conch *Strombus gigas*

S. Kamat<sup>1</sup>, X. Su<sup>1</sup>, R. Ballarini<sup>1</sup> & A. H. Heuer<sup>1\*</sup>

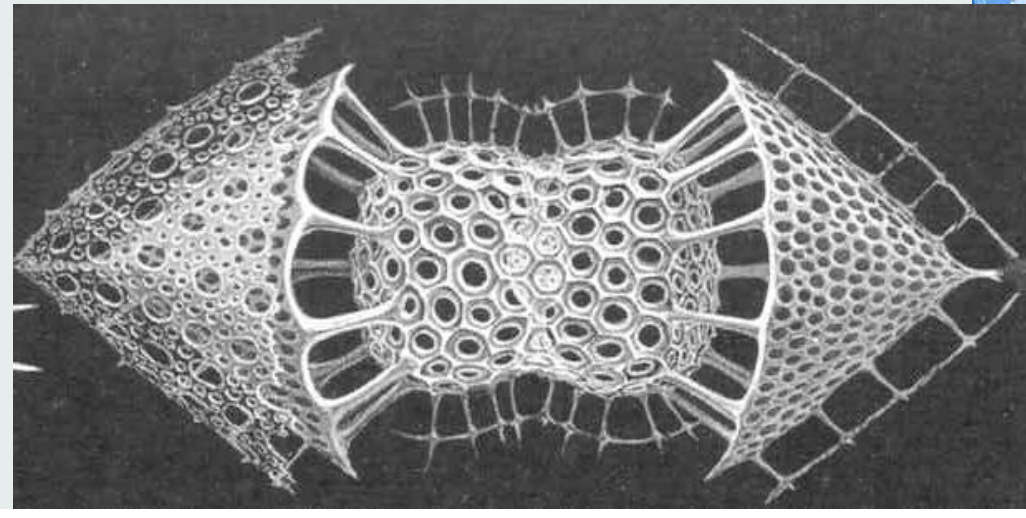
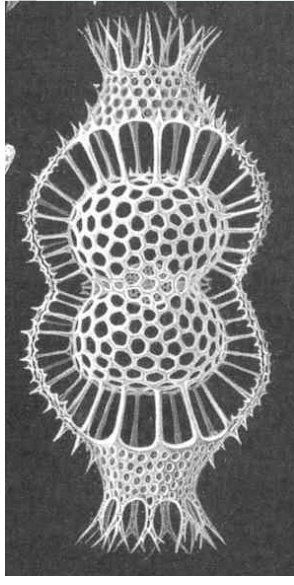
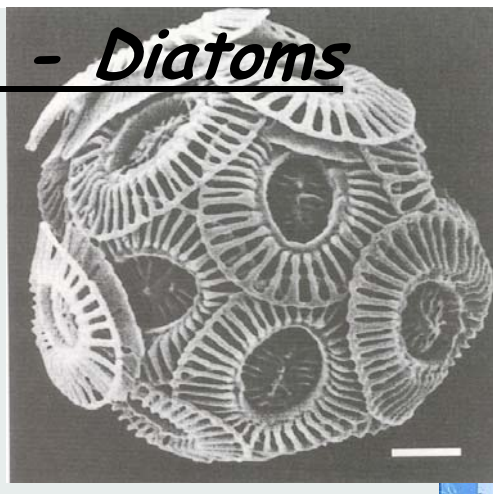
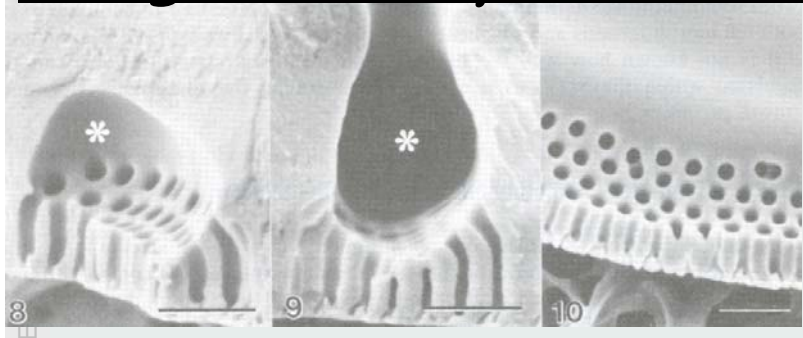
Nature 405 (2000) 1038

ZÜRICH & COLLEGIUM

- Layers : 0.5-2mm
1. Order lamella : 5-60mm
  2. Order lamella : 5-30mm
  3. Order lamella : 60-130nm

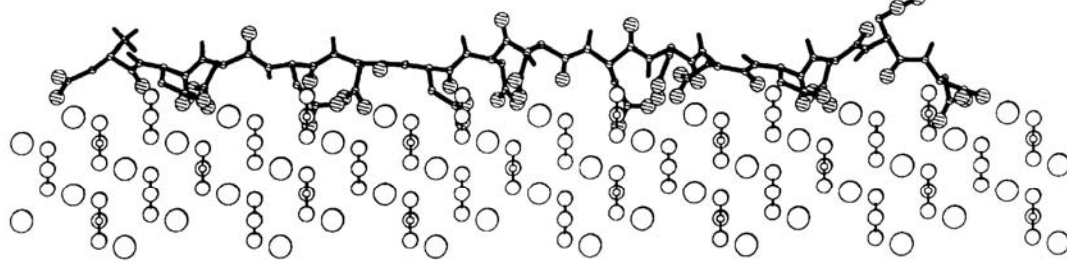
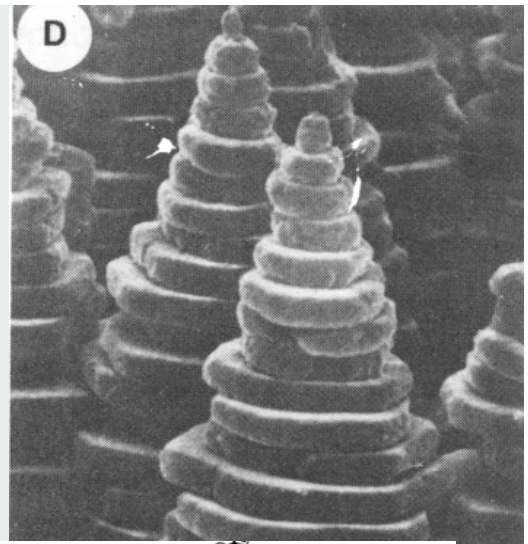
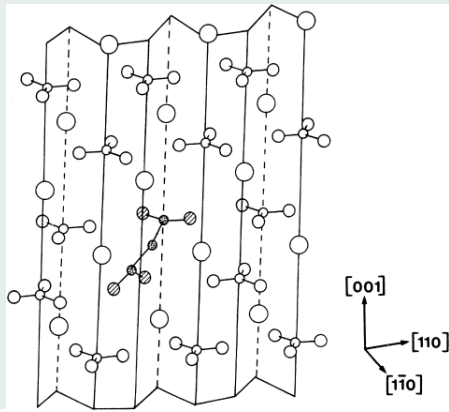


## Design of Complex Forms - Diatoms





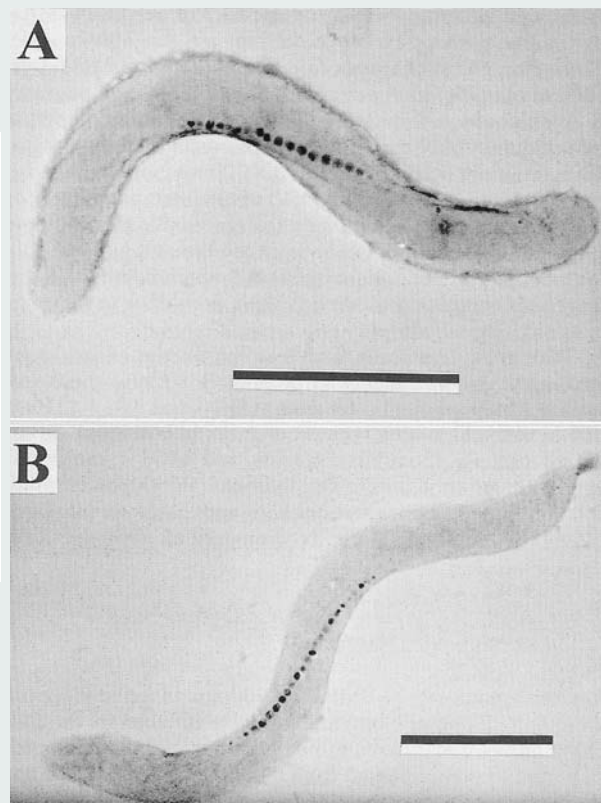
# How is it achieved ?



**Fig. 4.23** Computer model showing side view of the calcite  $\{110\}$  face with surface-bound polyaspartate ( $[\text{Asp}]_{11}$ ).

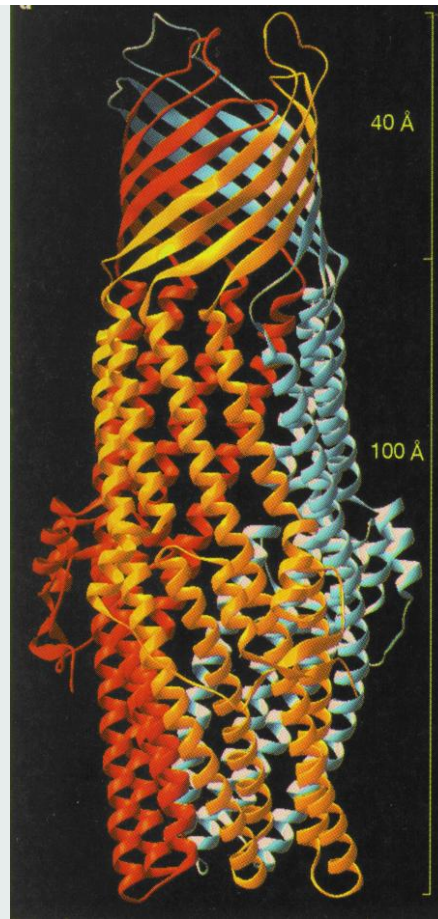
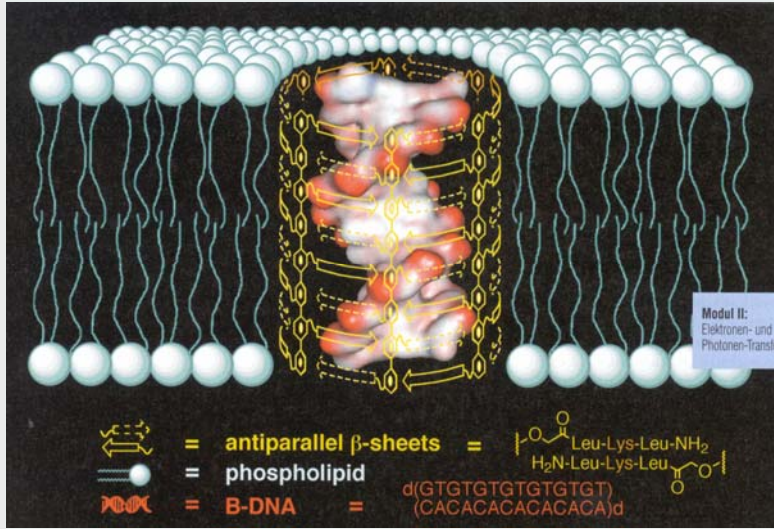
# Control and Function

- chemical
- spatial
- structural
- morphological
- constructional.



# Control and Function

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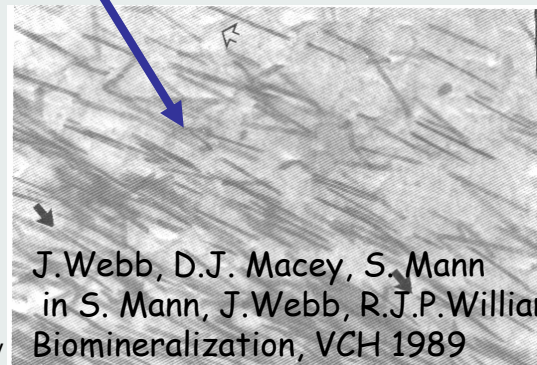
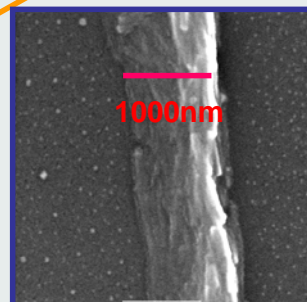
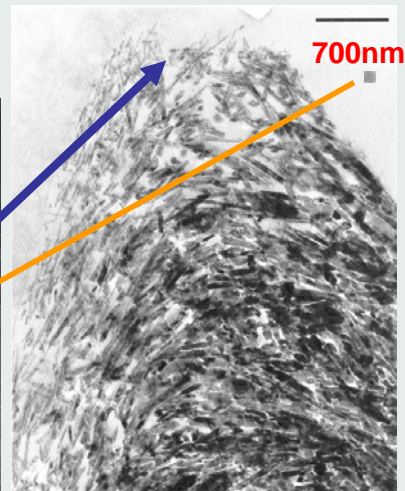
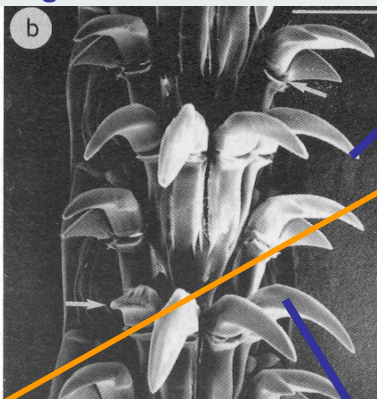
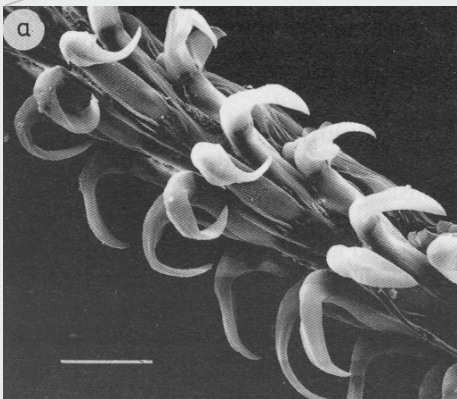


ETH  
 Eidgenössische Technische Hochschule Zürich  
 Swiss Federal Institute of Technology Zurich

# Structural Alignment

organic matrix / mould

Composite  
organic +  $\alpha$ -FeOOH



J. Webb, D.J. Macey, S. Mann  
 in S. Mann, J. Webb, R.J.P. Williams,  
 Biomineralization, VCH 1989

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 Eidgenössische Technische Hochschule Zürich  
 Swiss Federal Institute of Technology Zurich

# Biodynamic Restoration

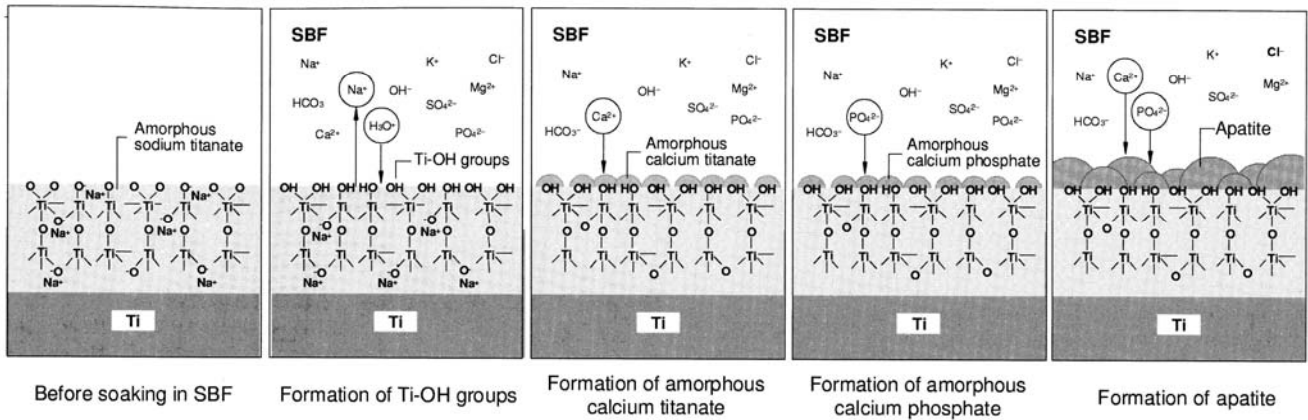
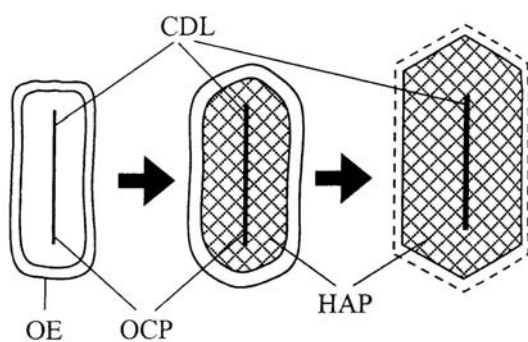


Figure 2. Schematic representation of the mechanism of apatite formation on NaOH-treated Ti metal in simulated body fluid (SBF) (from Reference 10).

# Enamel Formation

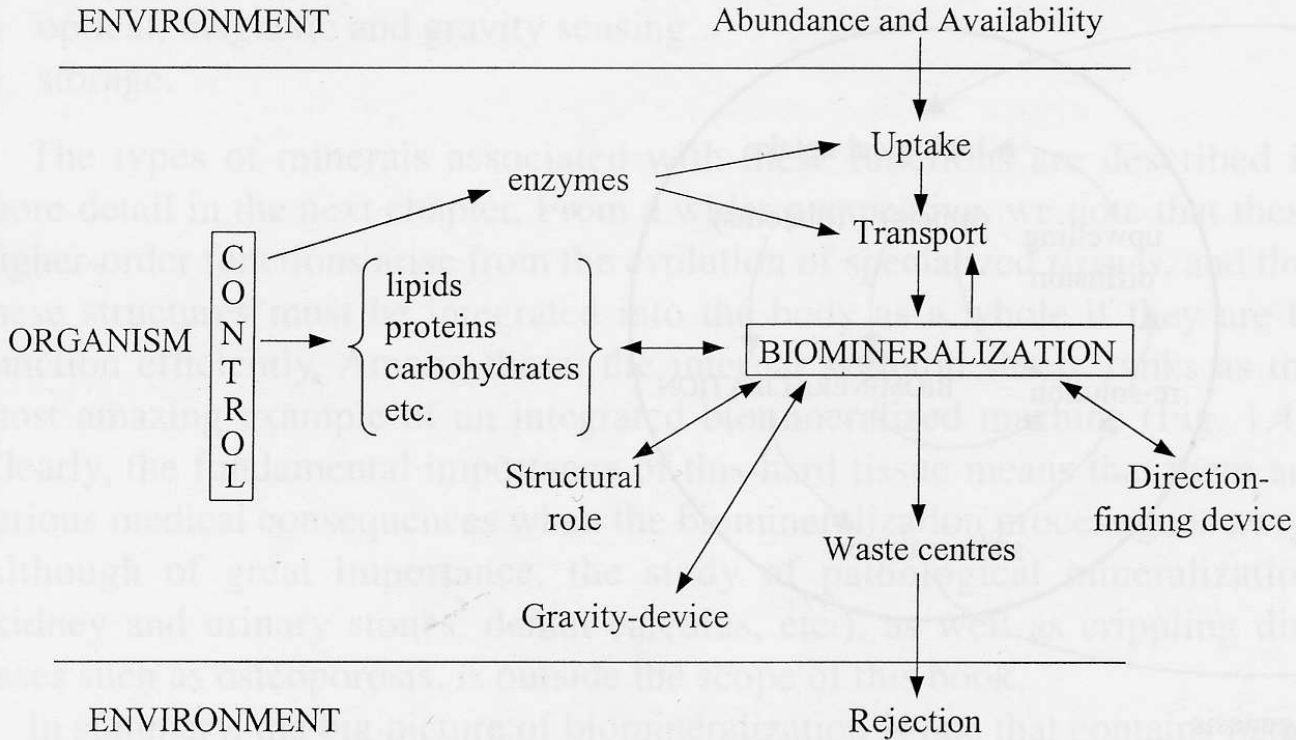


**Fig. 4.30** Formation of enamel crystals. An octacalcium phosphate (OCP) precursor phase is formed within an organic envelope (OE) and then overgrown with a single crystal of hydroxyapatite (HAP). Traces of the OCP phase are left as a central dark line (CDL) inside the mature crystal.

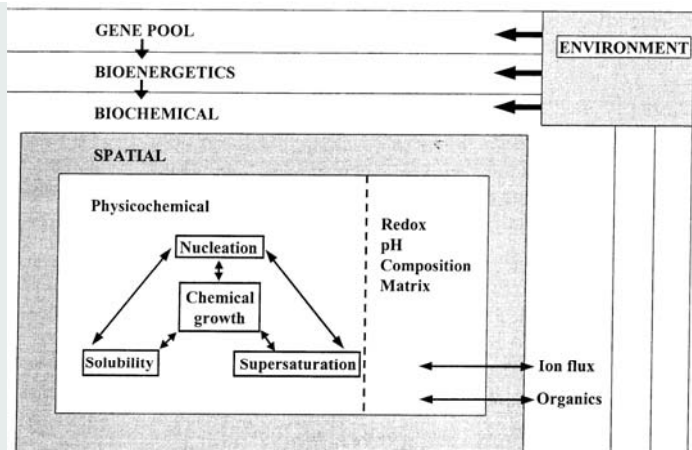
**Table 4.2** Epitaxial deposition of inorganic crystals on inorganic substrates. A positive percentage misfit value indicates that the overgrowth lattice is larger than the substrate lattice

Substrate	Overgrowth	Lattice misfit %
PbS	NaI	8
	KCl	5
	NaBr	-1
	NaCl	-6
	AgBr	-4
CaCO <sub>3</sub>	AgCl	-7
	RbBr	7
	RbCl	3
	KBr	3
	NaI	1
CaF <sub>2</sub>	KCl	-2
	NaBr	-7
	NaBr	8
	NaCl	3
	LiBr	0
NaCl	LiCl	-6
	NaBr	6
	NaCN	6
	AgBr	3
	AgCN	3
	AgCl	1
	KF	-5

# Living System



# Living System



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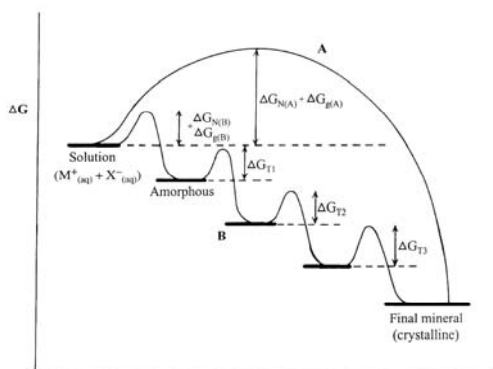


Fig. 4.25 Pathways to crystallization and polymorph selectivity: (A) direct, (B) sequential. See text for details.

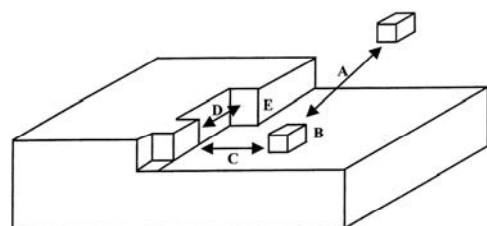


Fig. 4.6 Layer-by-layer mechanism of crystal growth. See text for details.

# Biosensors

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